



• *Applications of Cinchona OrganoCatalysts in  
Asymmetric Catalysis*

Anthony J. Burke,  
Faculty of Pharmacy,  
University of Coimbra, PORTUGAL.  
(ajburke@ff.uc.pt)

6 June 2023





# Content

- Imine Hydrosilylation
- Cinchona-Picolinamides
- API Targets
- Mechanism
- Continuous Flow
- Cinchona-Squaramides (Michael addition)
- Catalytic Chiral Solvents (Deep Eutectic solvents)

# The Nobel Prize 2021



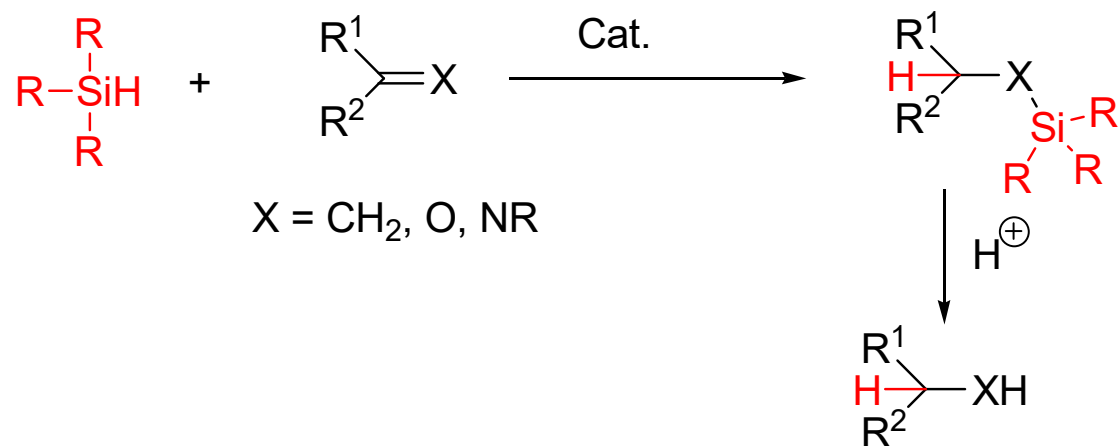
David MacMillan



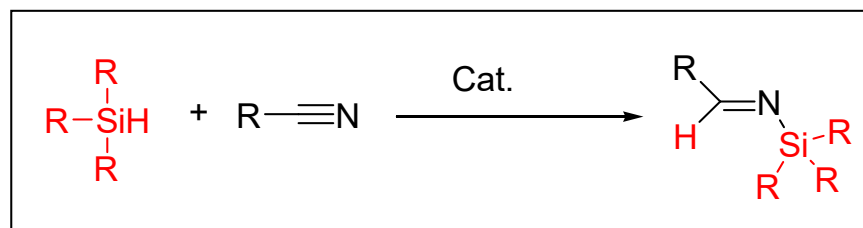
Ben List

“for the development of asymmetric organocatalysis.”

# Hydrosilylation of Double Bonds



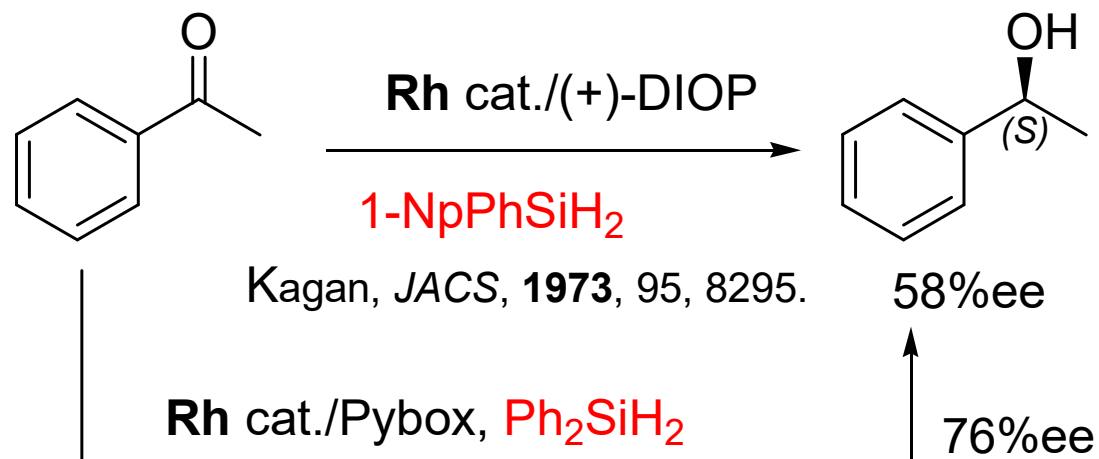
And Nitriles



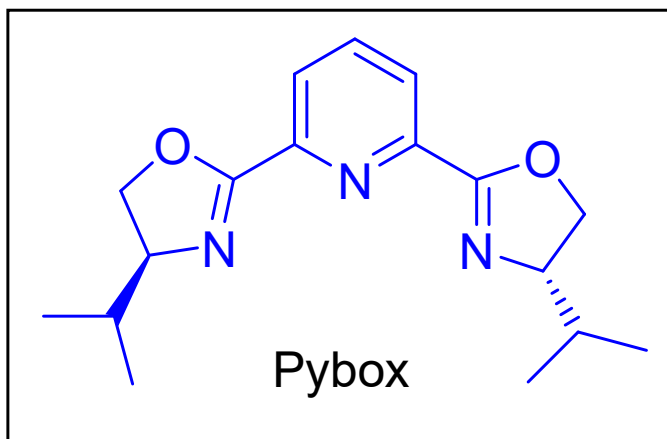
And Enamines!

Hashimoto *et al.* *JACS*, **2003**, *22*, 2199.

# Pioneering work with Ketones

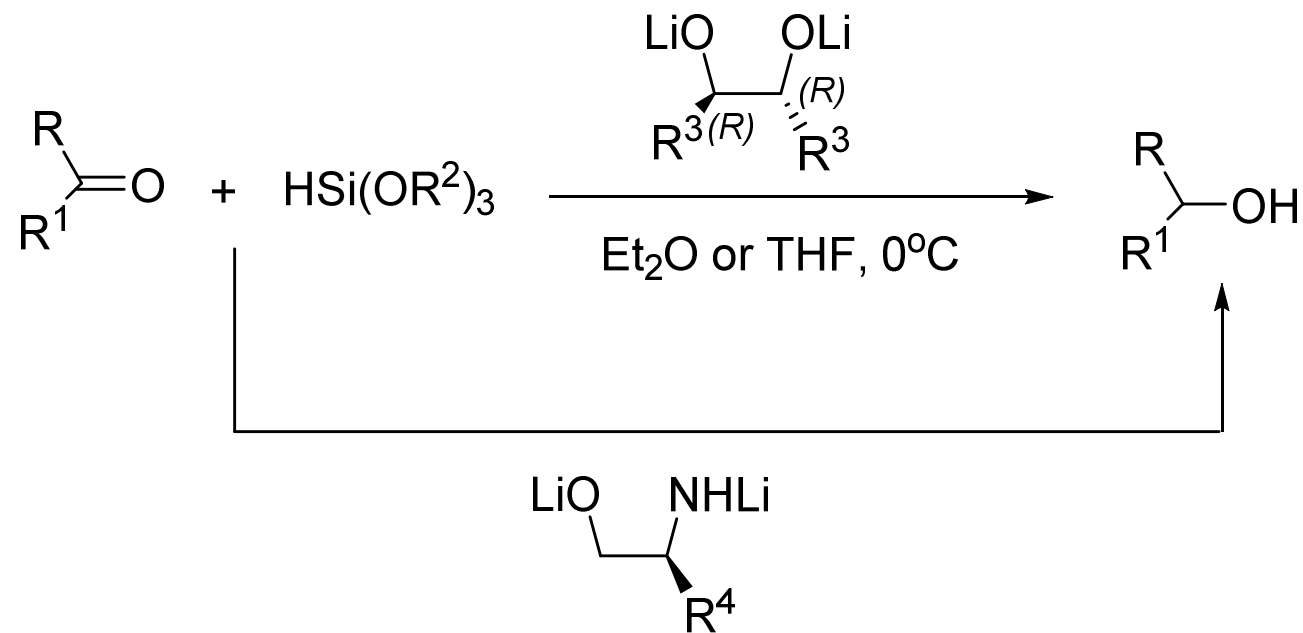


Nishiyama, *Organometallics*, **1989**, 8, 846.



# Using organocatalysts

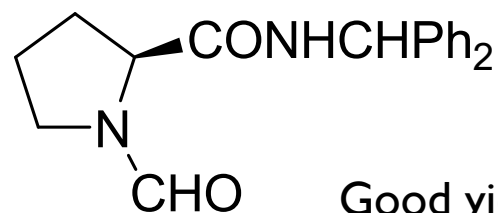
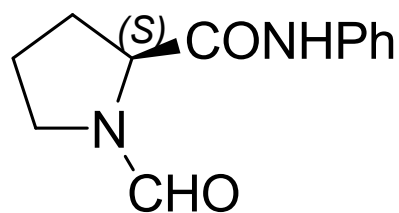
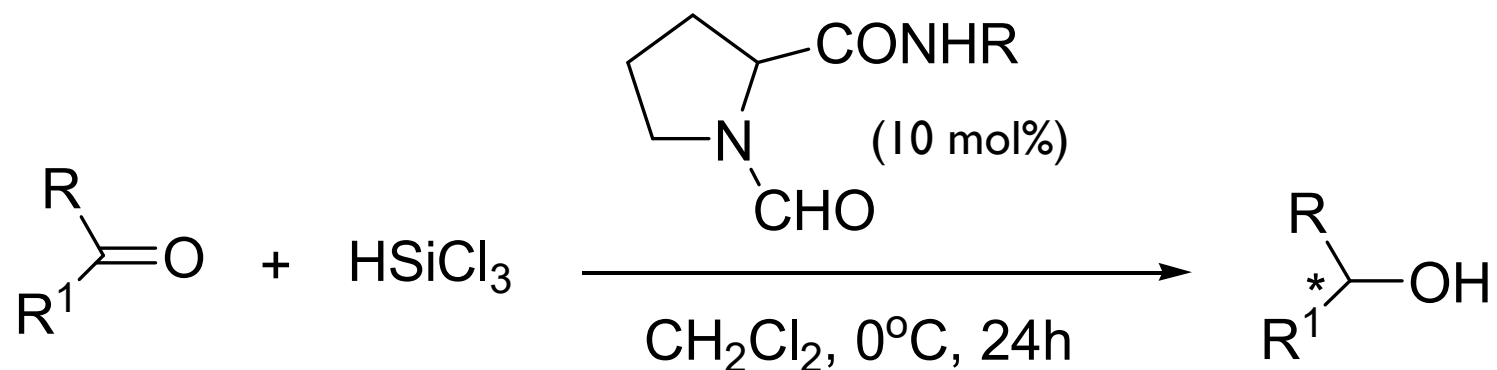
Lithium Diolates and aminoalcoholates



Hosomi, *Tet. Lett.* 1988, 29, 89.

# Milestone: *N*-Formylpyrrolidine activation

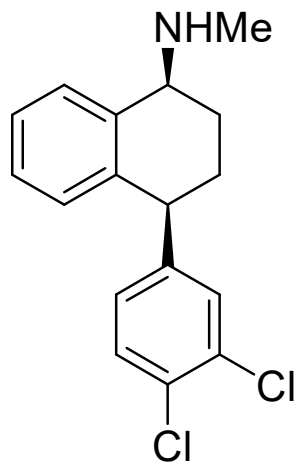
(DMF-HSiCl<sub>3</sub>), Kobayashi, *Chem. Lett.* 1996, 25, 407.



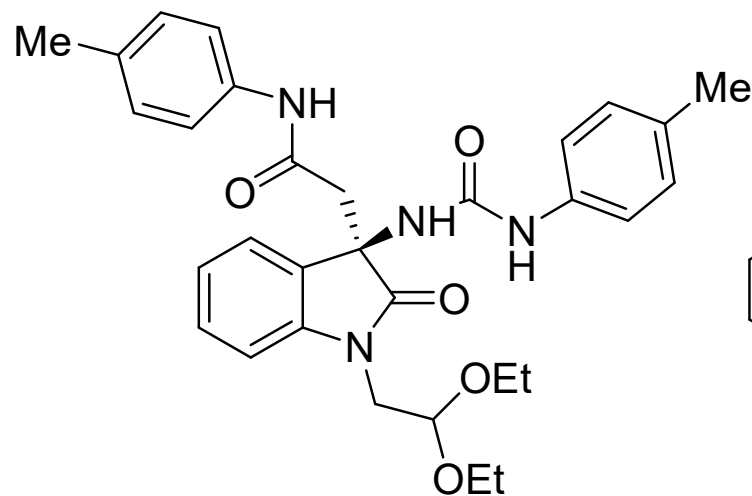
Good yields but low  
ees!

Matsumura, *Tet. Lett.* 1999, 40, 7507.

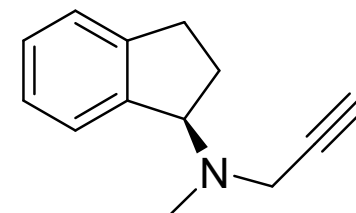
# Bioactive Chiral Amines



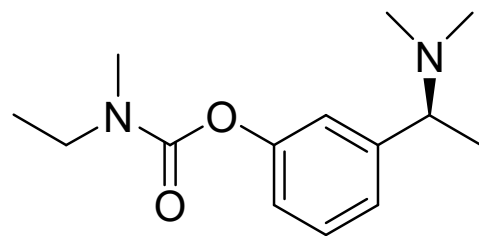
**Sertraline**  
(anti-depressant)



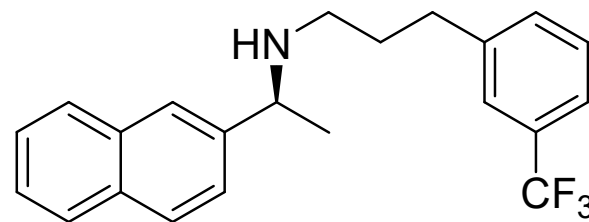
**AG-041R**  
(Gastrin/CCK-B  
receptor agonist)



**Rasagline**



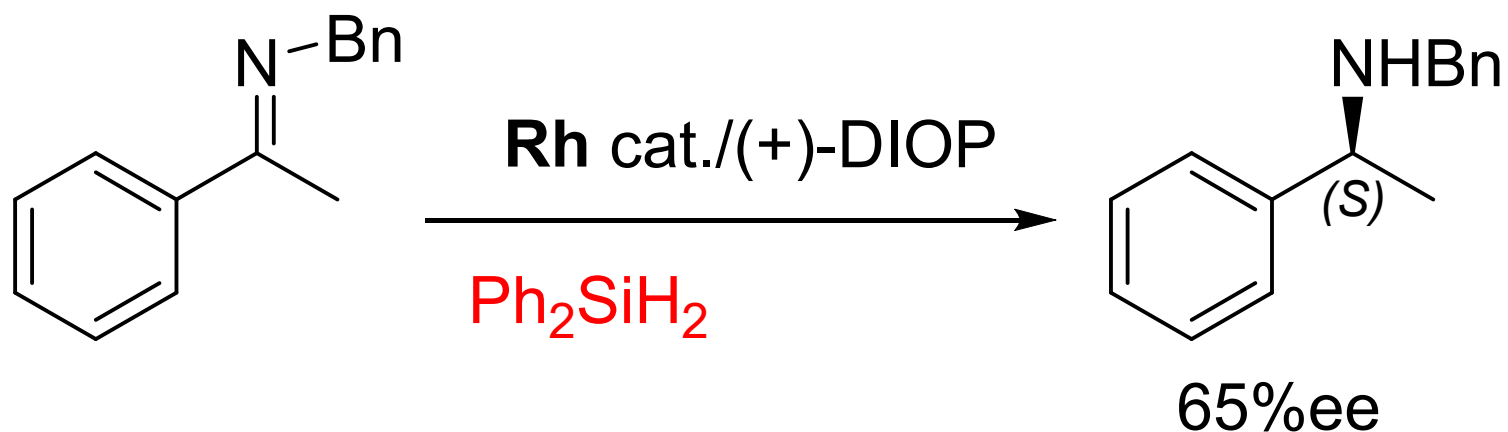
**Rivastigmine**



**Cinacalcet**

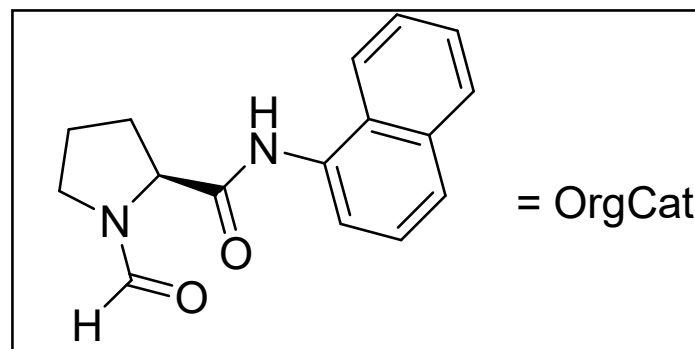
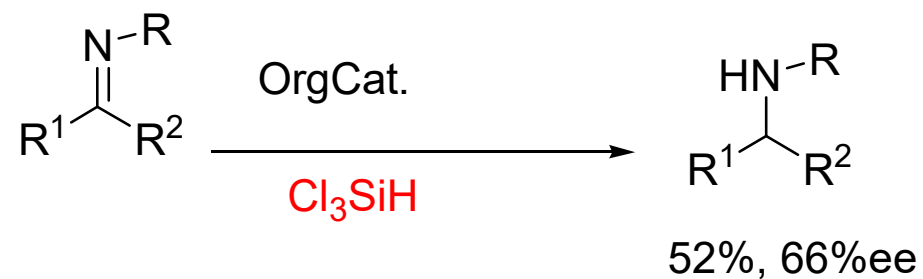


## Early developments with Ketimines



Kagan, *J. Organomet. Chem.* **1975**, 90, 353.

# With Organocatalysts

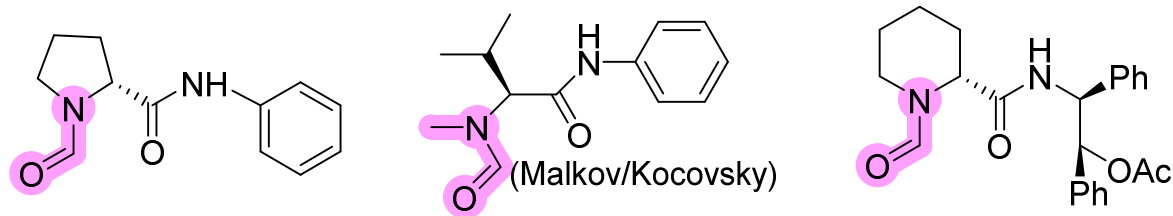


Ketones and Enamines

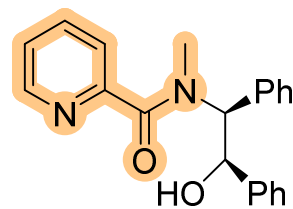
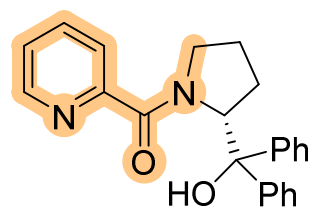
Matsumura, *Tet. Lett.* 2001, 42, 2525.

# Organocatalytic Imine Hydrosilylation: Types

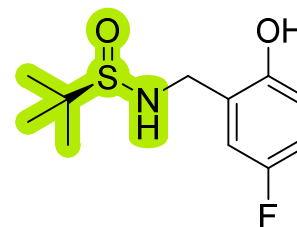
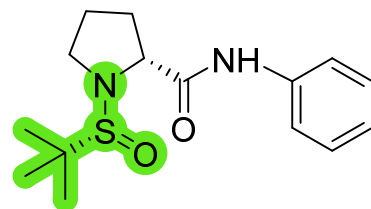
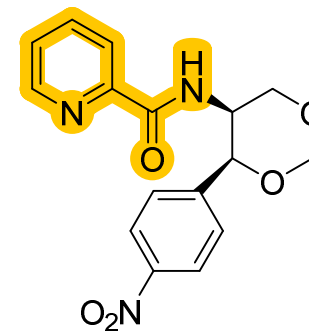
## Chiral Lewis Bases (with $\text{Cl}_3\text{SiH}$ )



Formamides



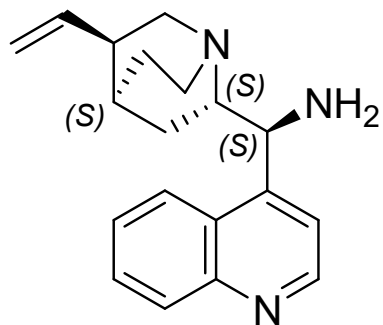
Picolinamides



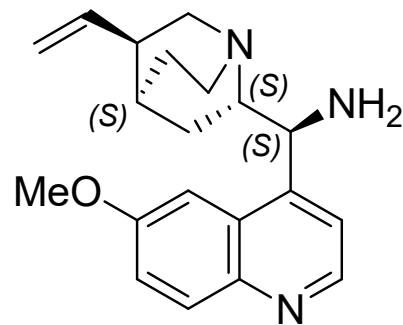
Sulfinilamides

# Cinchona Organocatalysts: Bifunctional catalysts

Bernardino Gomes (1811)

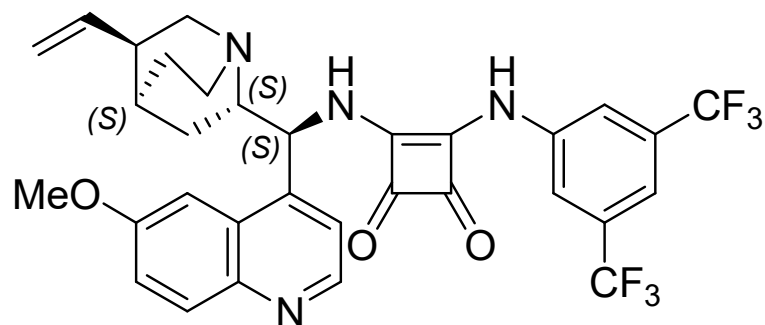


9-Amino-*epi*-Cinchonidine

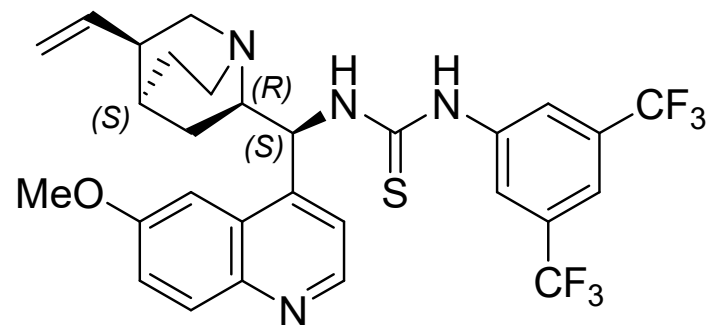


9-Amino-*epi*-Cinchonine (Brunner)

Enamine



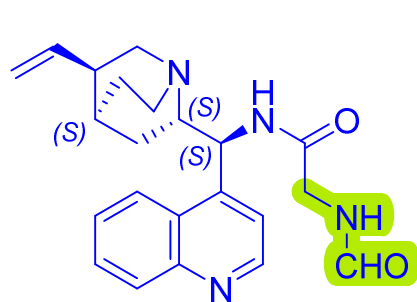
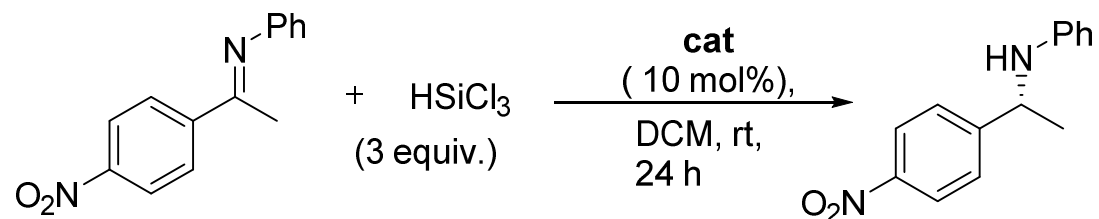
Cinchona Squaramide (Rawal, 2008)



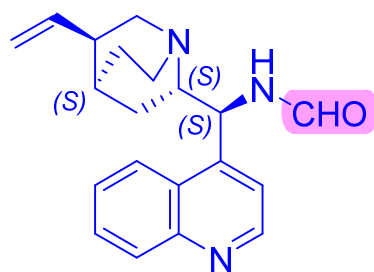
Cinchona thiourea (Dixon, 2005)

H-Bonding networks

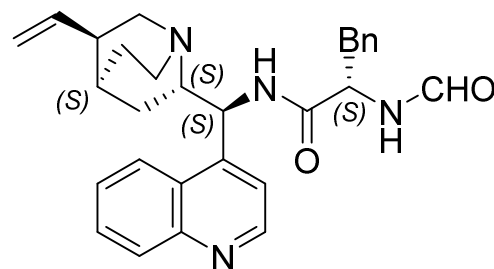
# Cinchona-Formamides



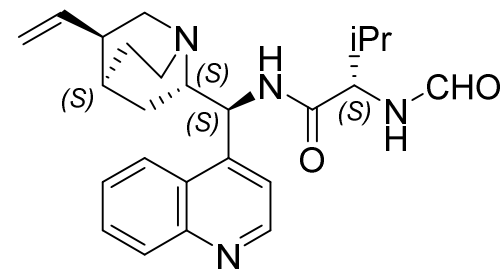
66% (13% ee)



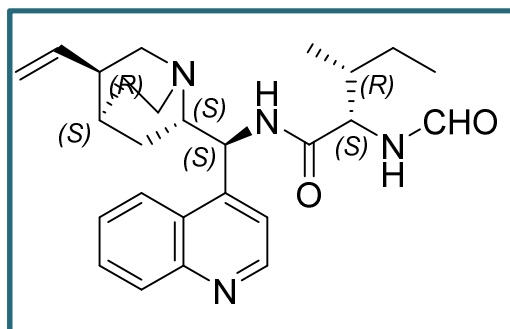
64% (7% ee)



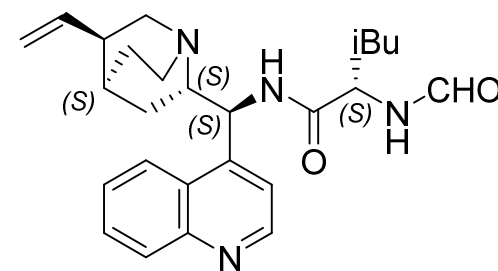
62% (33% ee)



56% (35% ee)

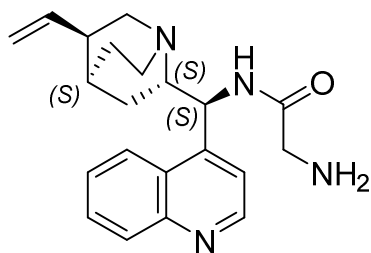
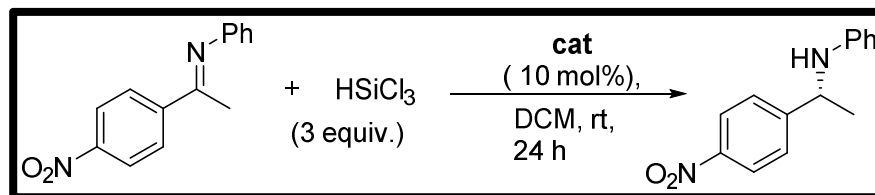


76% (27% ee)

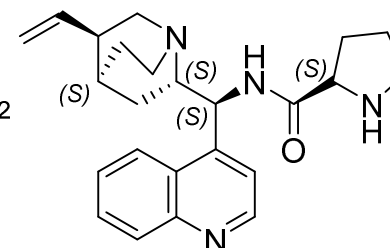
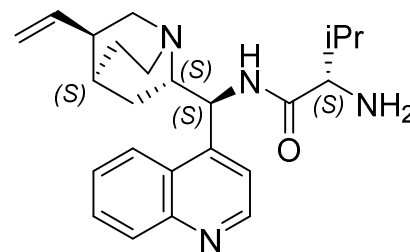
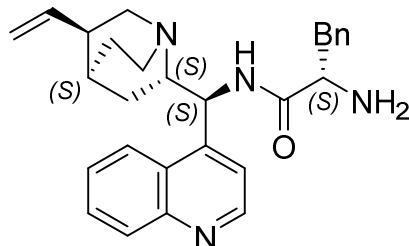


74% (29% ee)

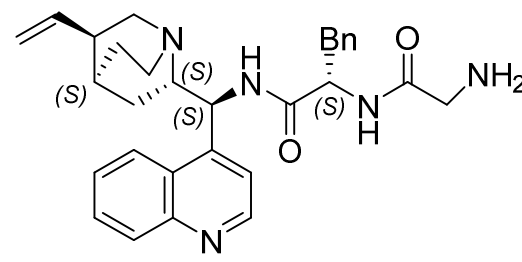
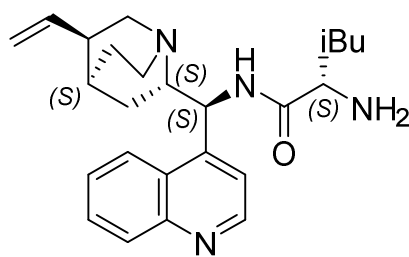
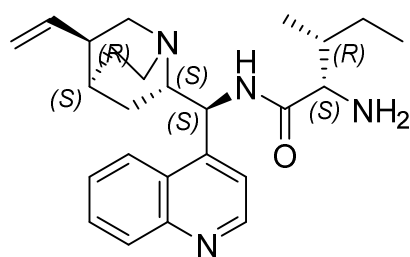
# Cinchona-amine-aminoacids



72%, 31% ee



74%, 37% ee



84%, 45% ee

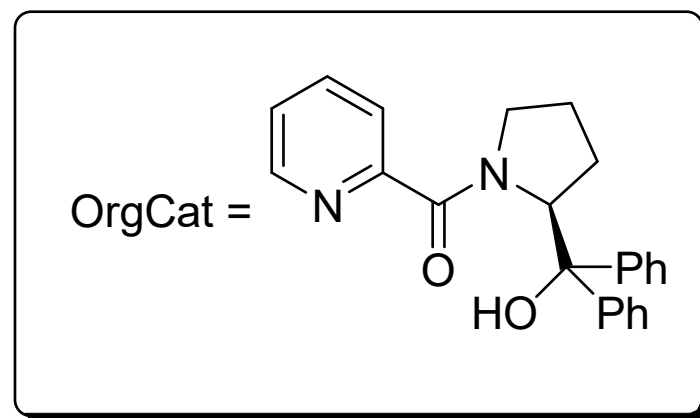
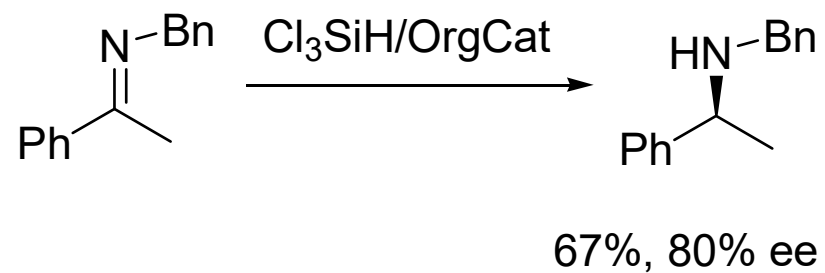
**Synthesis of novel cinchona-amino acid hybrid organocatalysts for asymmetric catalysis**



Pedro Barrulas<sup>a</sup>, Maurizio Benaglia<sup>b,†</sup>, Anthony J. Burke<sup>a,\*</sup>

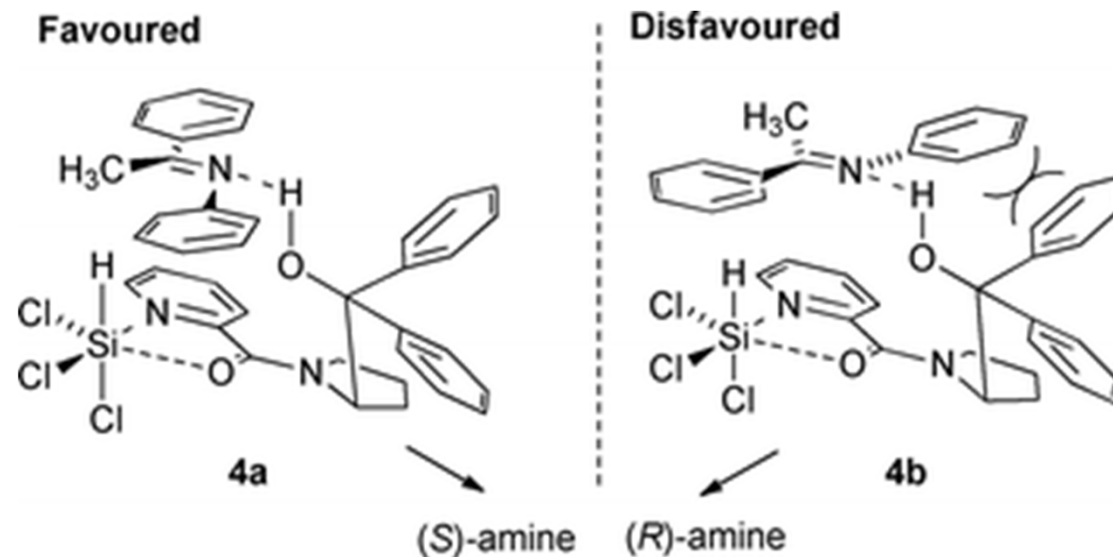
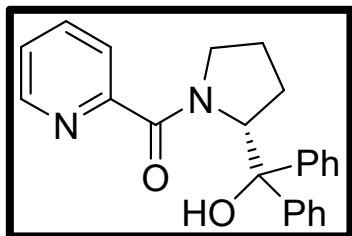
*Tetrahedron: Asymmetry* 25 (2014) 923–935

# Chiral Picolinamides



Matsumura, *JP 2005029503, Tet. Lett.* 2006, 47, 3751.

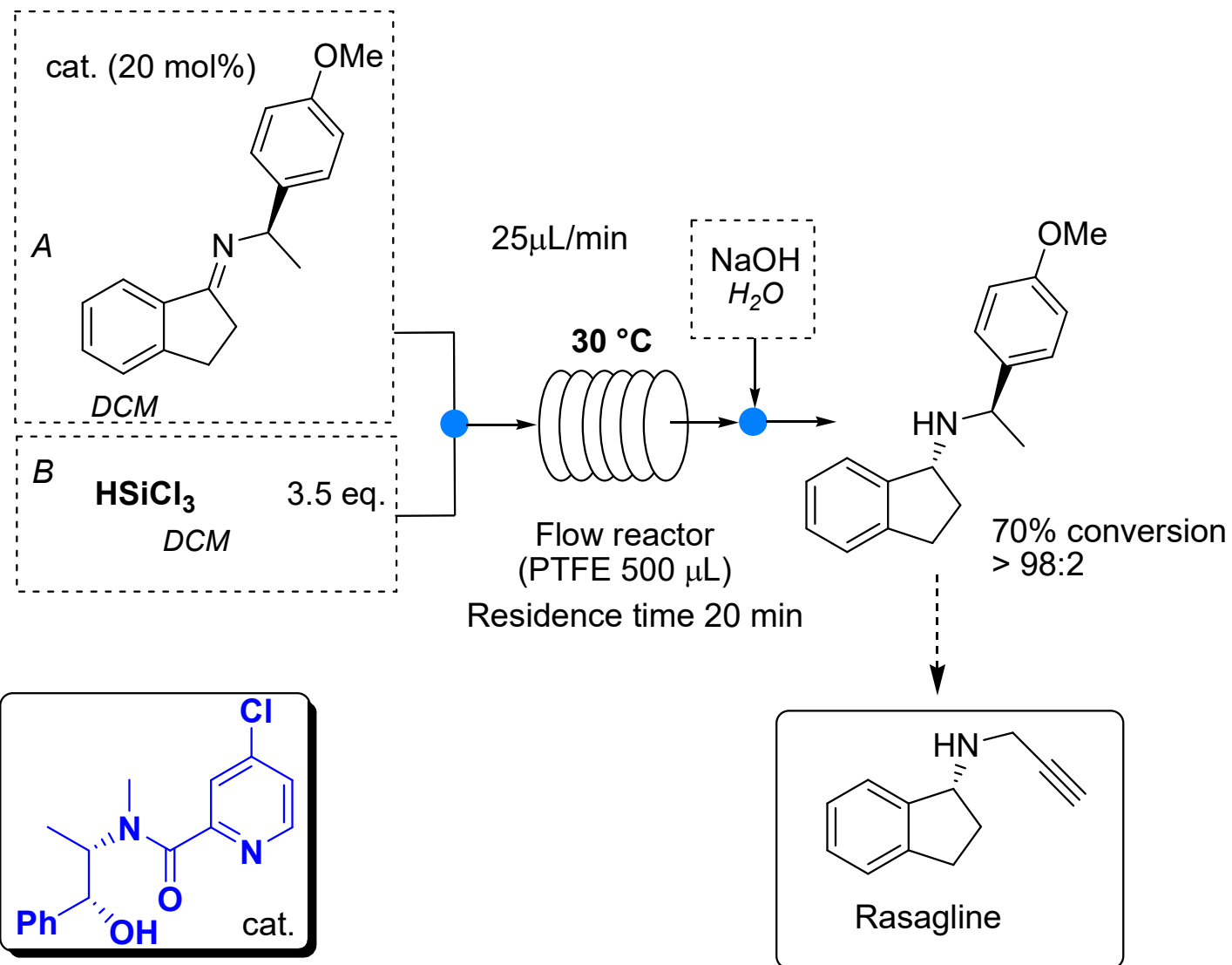
# Picolinamide Organocatalysts



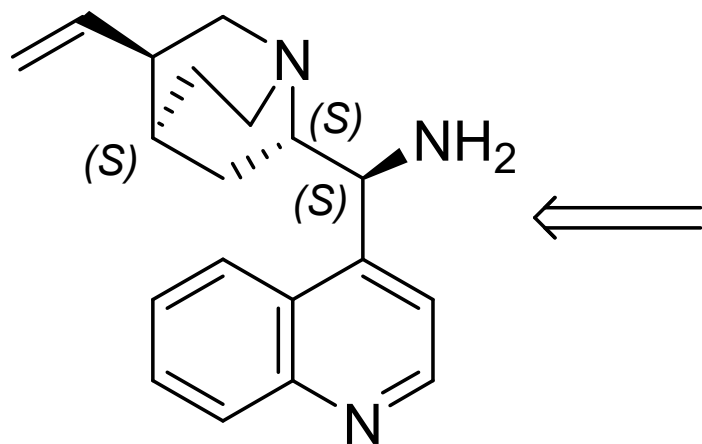
Matsumura's working model (Reproduced from Jones, *Org. Biomol. Chem.* **2012**, 10, 2189)



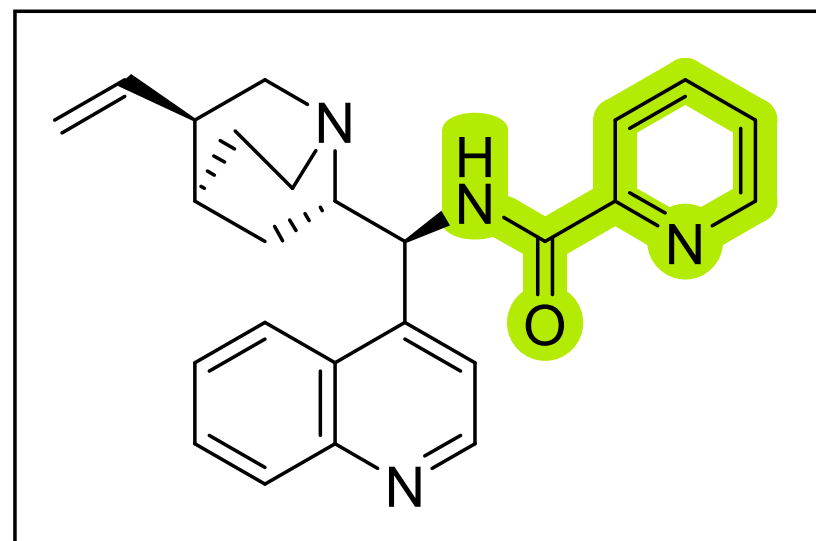
Brenna, *Eur. J. Org. Chem.* **2017**, 39-44



# Picolinamides

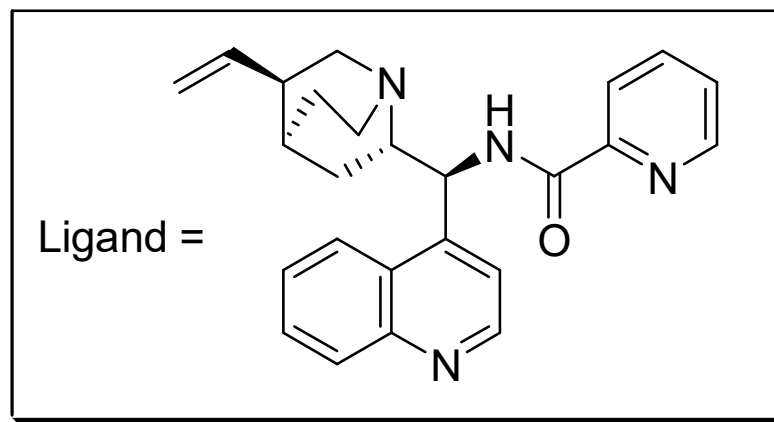
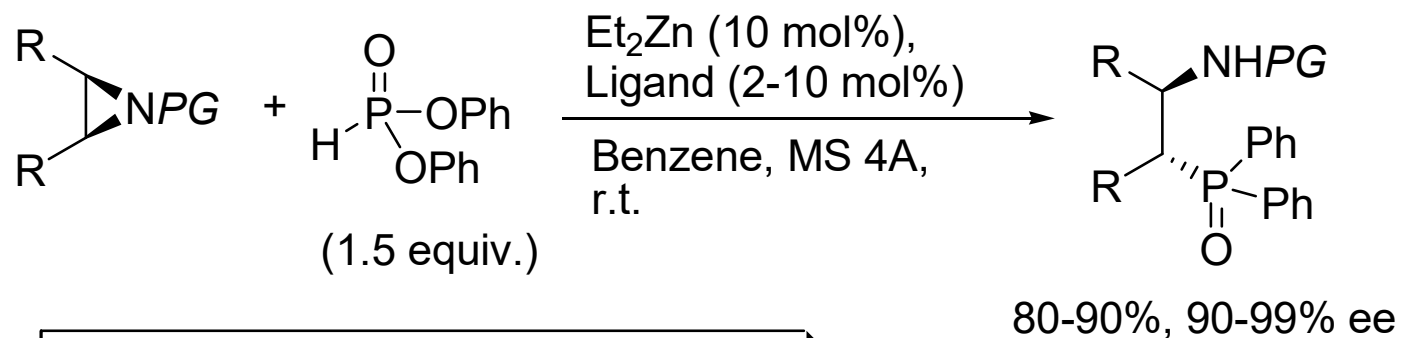


9-Amino-*epi*-Cinchonidine



Prototype catalyst

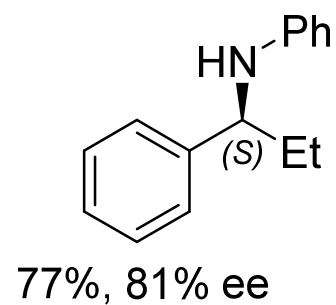
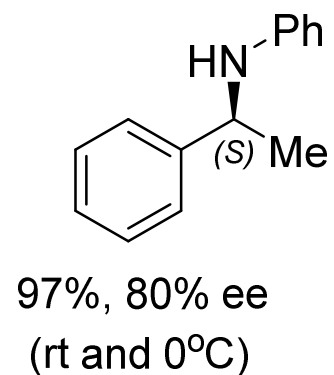
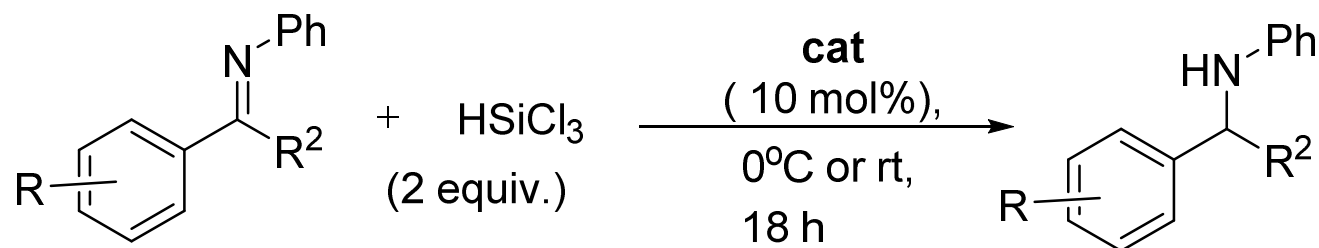
# Enantioselective Desymmetrization



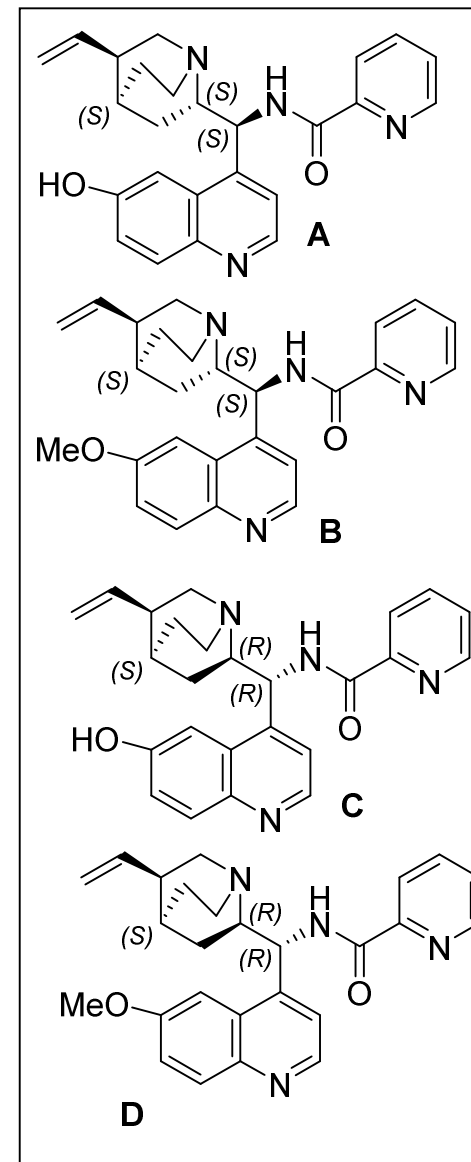
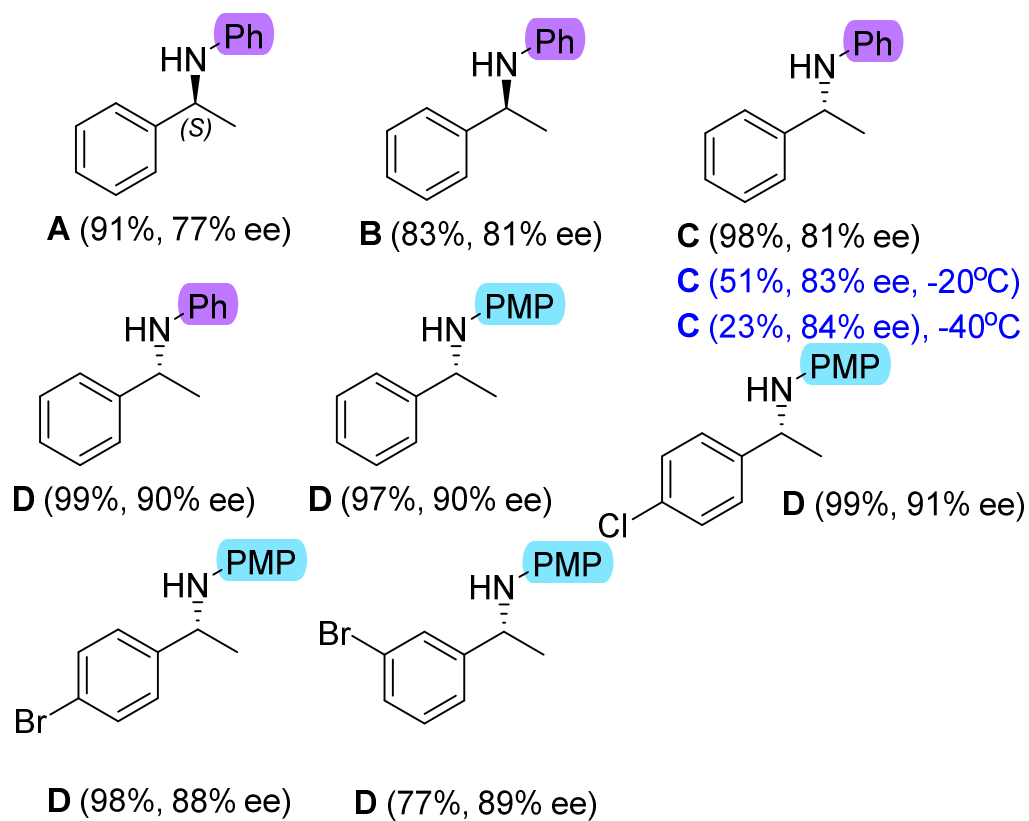
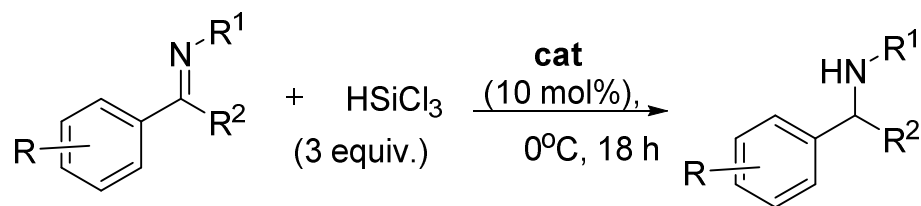
Nakamura, *JACS*. **2012**, 134, 19366.

# Initial Results

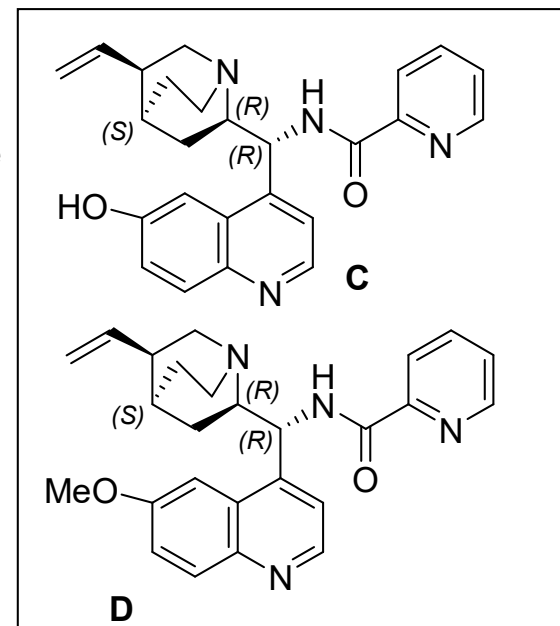
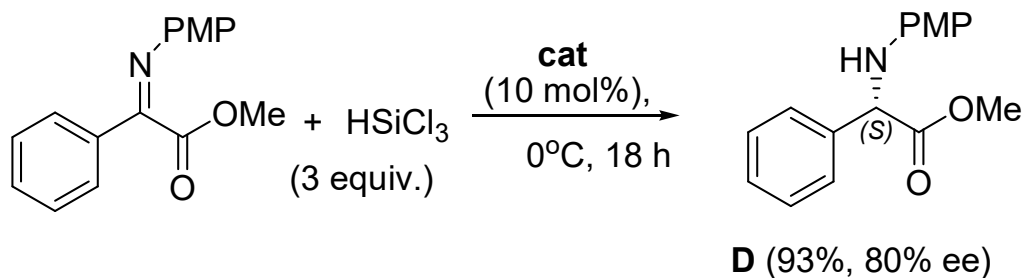
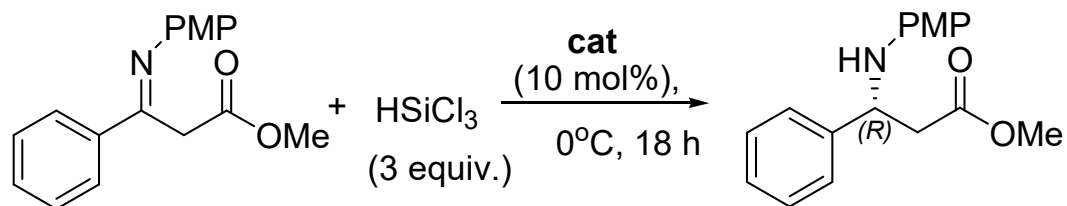
Prototype Catalyst



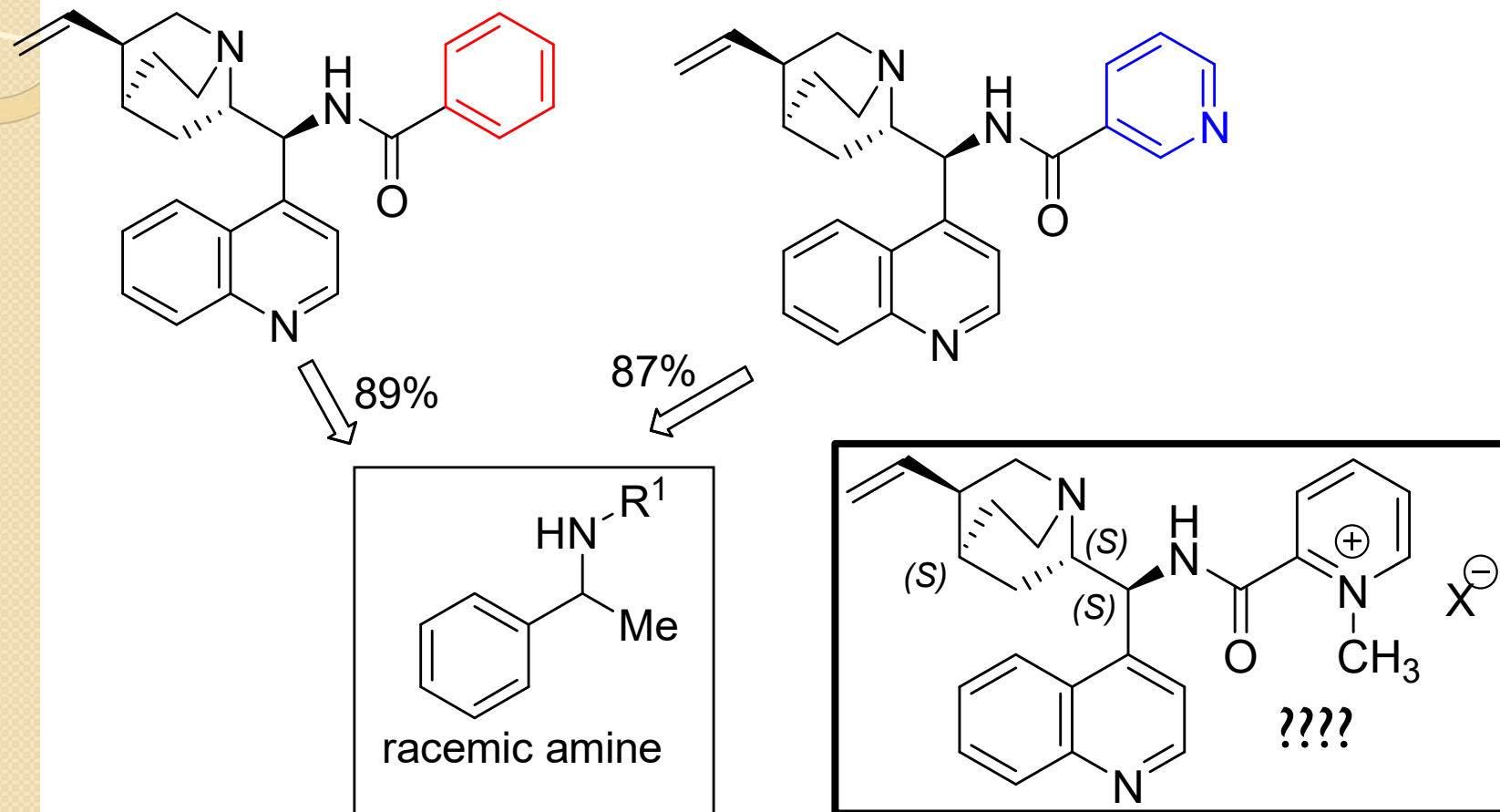
# Catalyst Screening



# Further Screening



## Probing the Mechanism



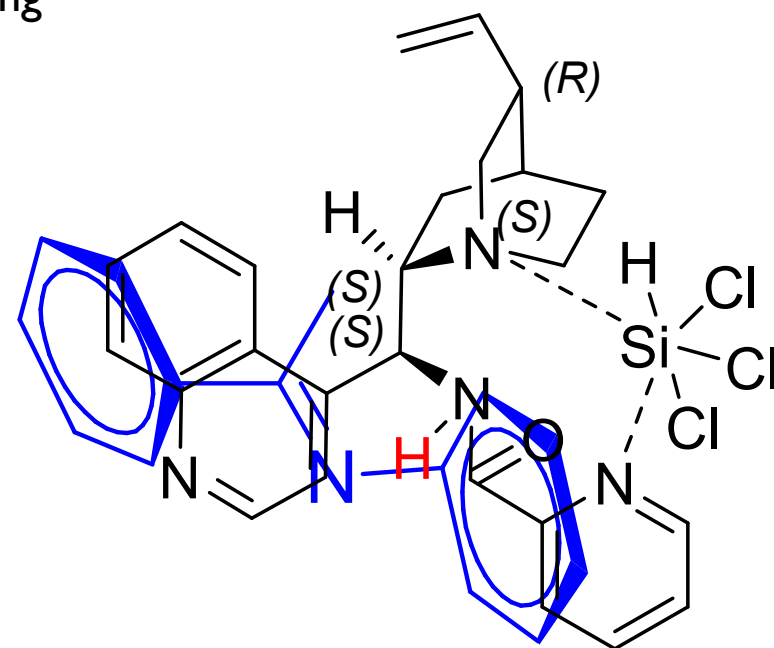
# Initial Working Model

$\pi$ - $\pi$  stacking

Transition state Stabilization

Catalyst preorganization

Tight/compact transition state



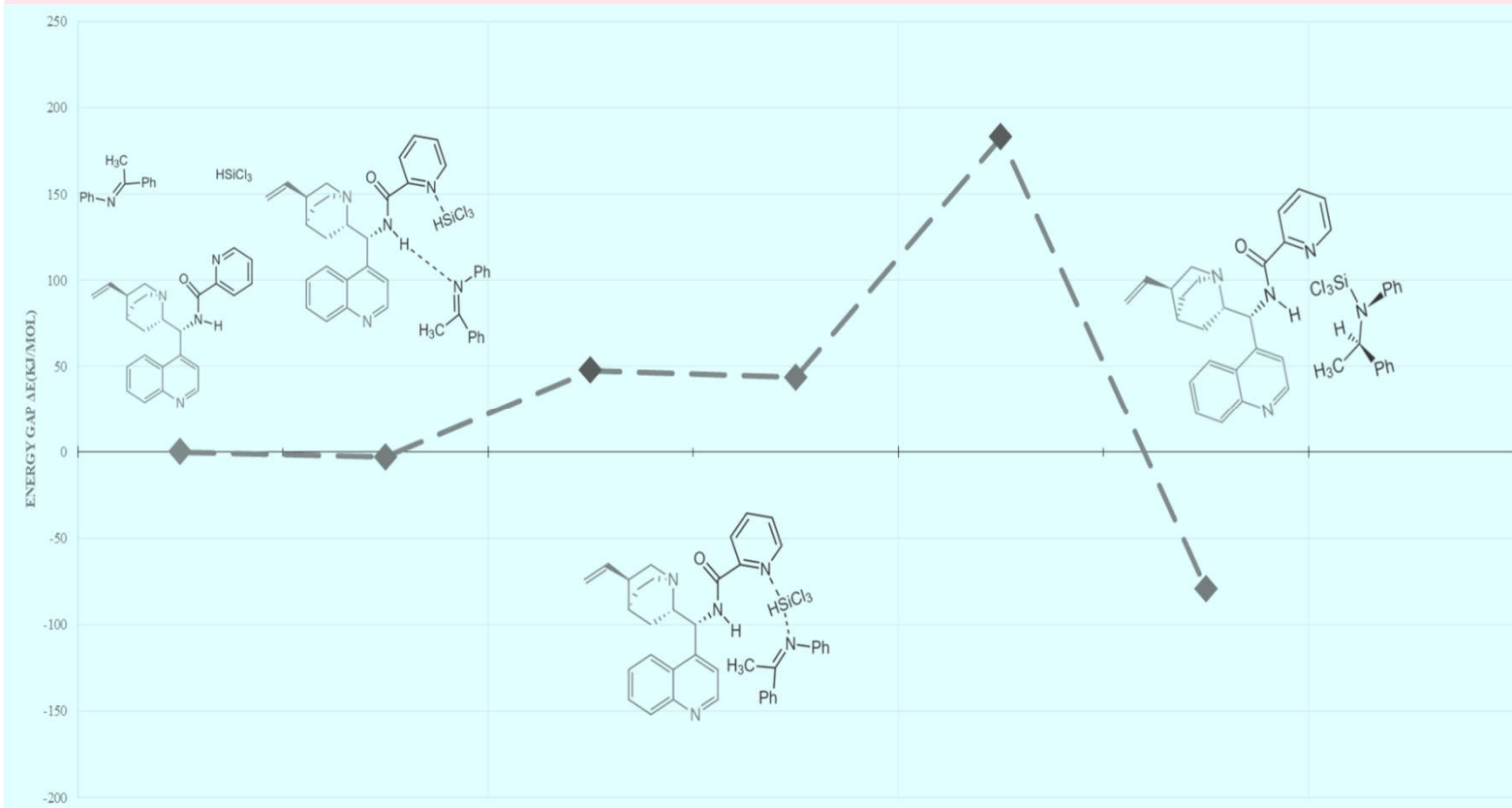
**Si-face attack of the hydride**



# DFT calculations

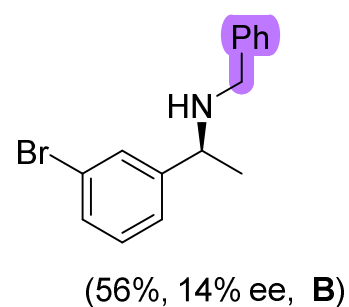
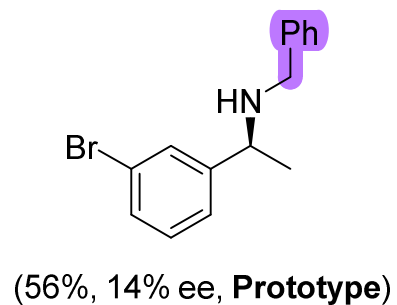
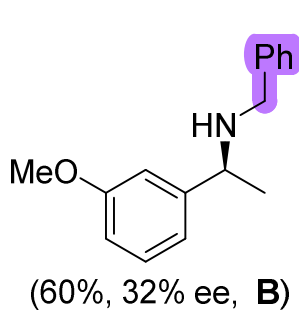
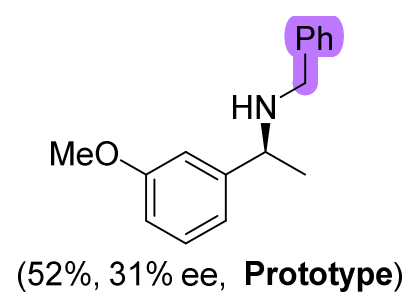
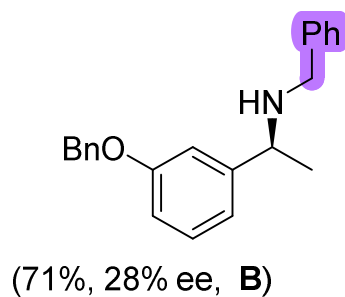
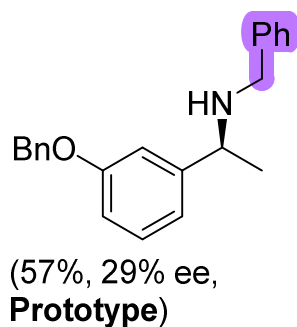
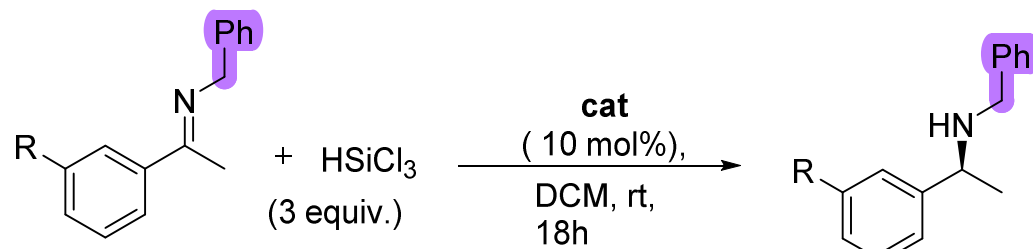
Tools = B97 functional with a D3 dispersion correction and triple-zeta basis set.

- **Transition state profile**

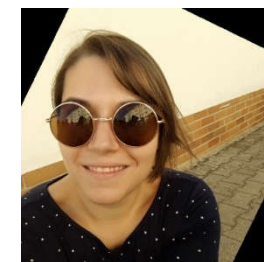
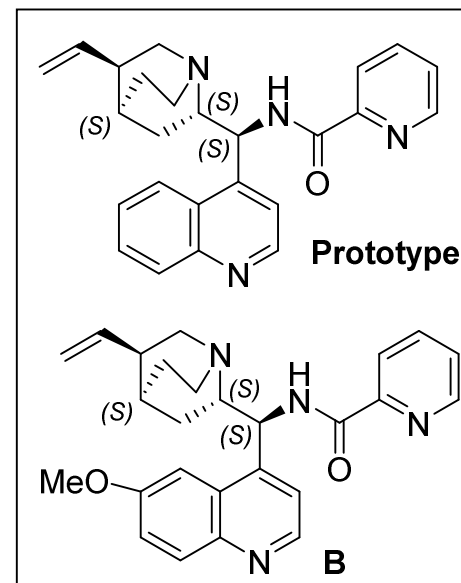


# Further studies

Supports our mechanistic model.



You need the Phenyl!

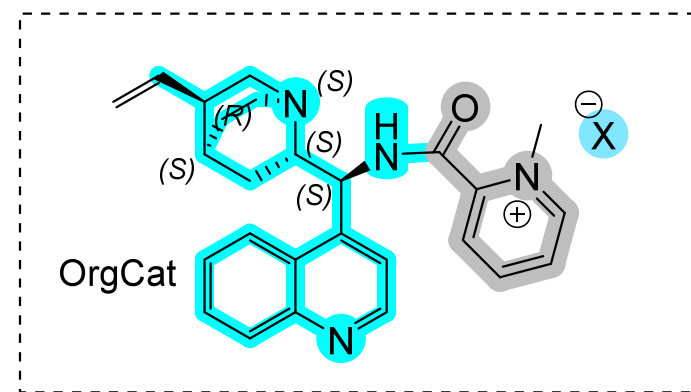
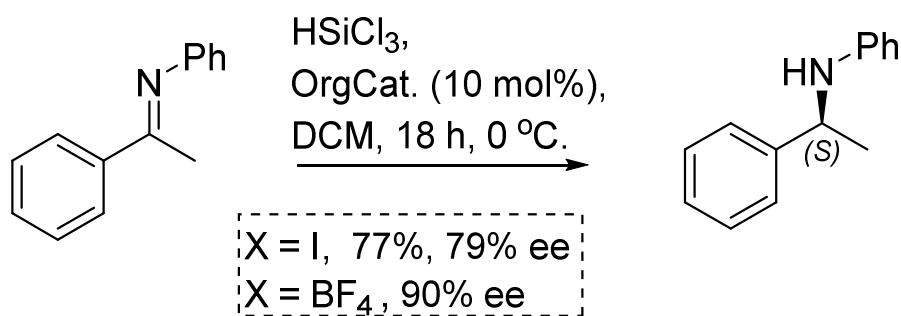
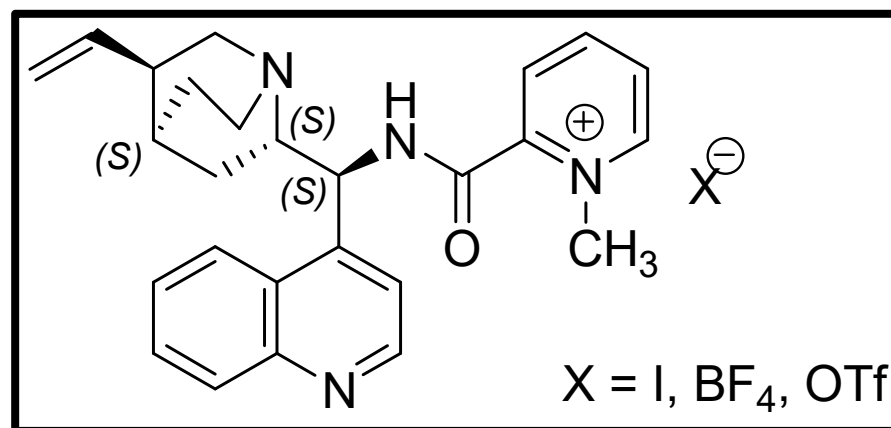


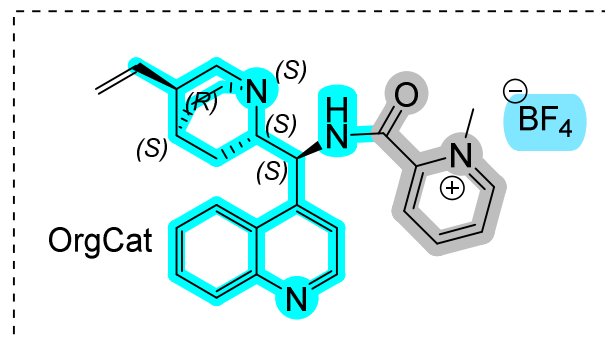
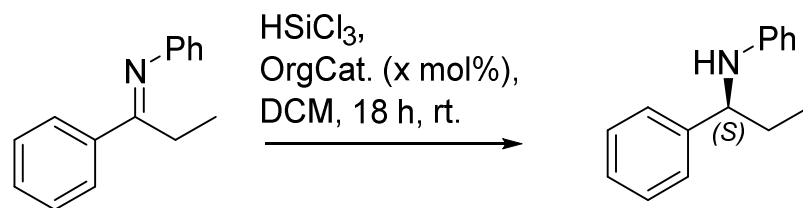
Sílvia Fernandes

# N-Methyl Cinchona-Picolinamide

Cation- $\pi$  interactions  
Counter ion effects

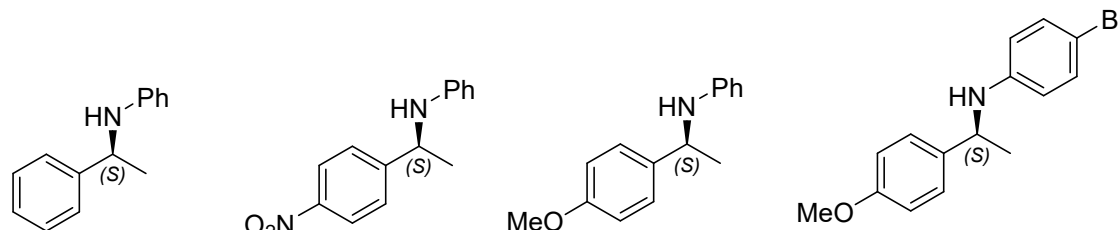
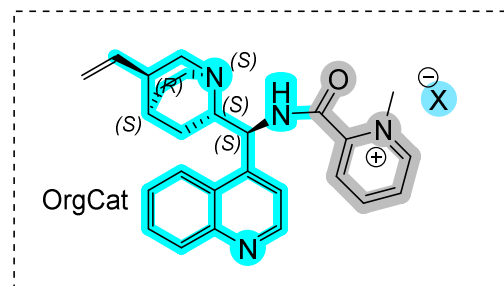
Jacobsen, *ACIE*, **2016**, 55, 12596.





Entry	x mol %	Yield %	e.e. %	TOF (mol.mol <sup>-1</sup> .h <sup>-1</sup> )
1	20	85	85	0.28
2	10	86	84	0.56
3	5	88	82	1.11
4	1	86	80	5.56
5	0.5	78	72	11.1
6	0.1	53	18	55.6

P. Barrulas, PhD Dissertation, University of Evora, 2014.

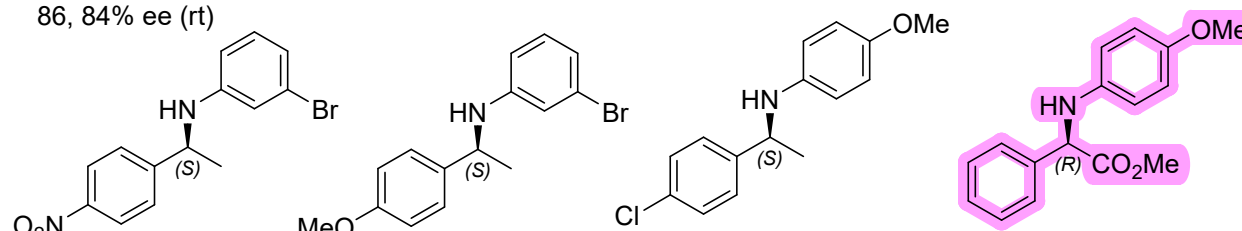


79%, 80% ee (-20 °C)  
 80%, 80% ee (0 °C)  
 86, 84% ee (rt)

68%, 83% ee (rt)

58%, 84% ee (rt)

10%, 69% ee (rt)

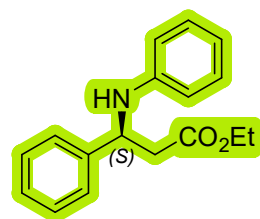


75%, 76% ee (rt)

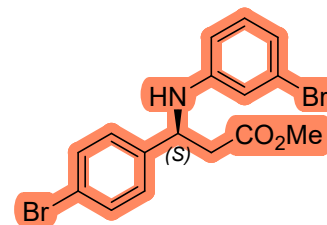
20%, 69% ee (rt)

85%, 83% ee (0 °C)

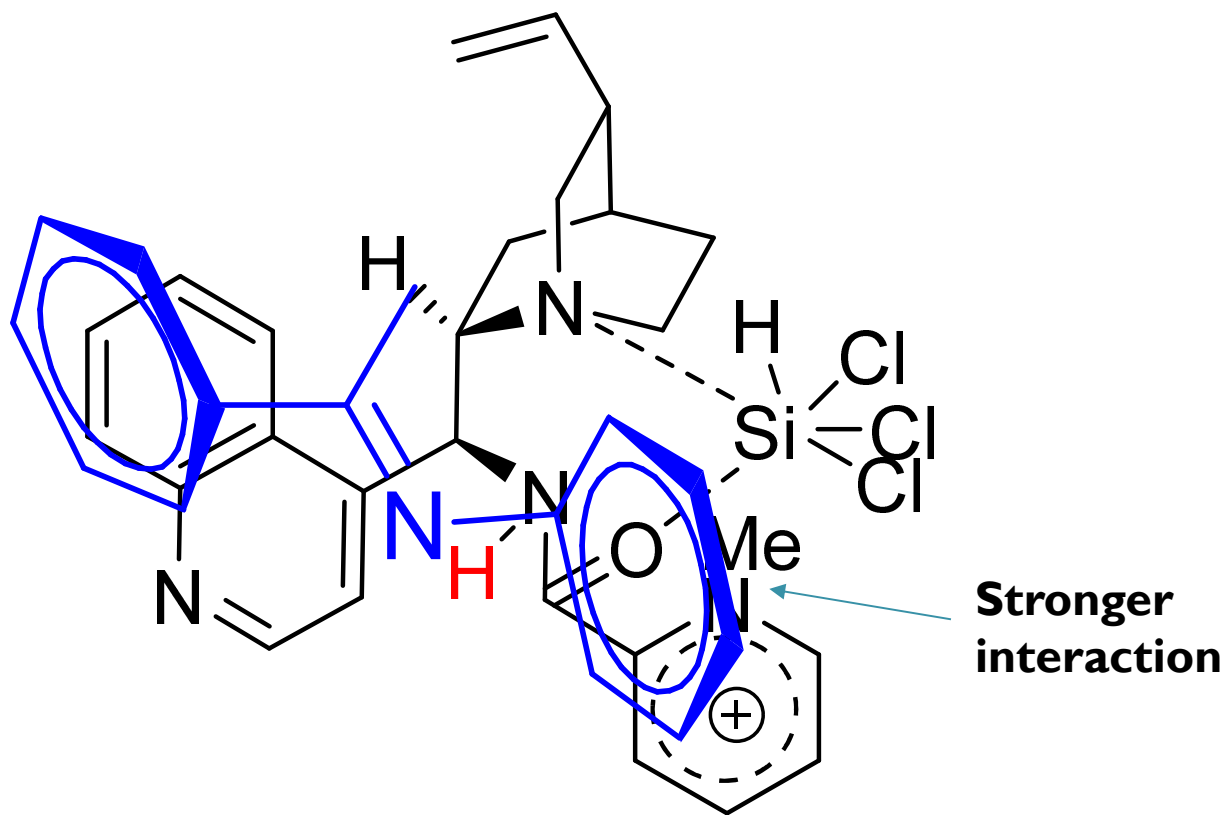
93%, 81% ee (0 °C)



72%, 70% ee (rt)



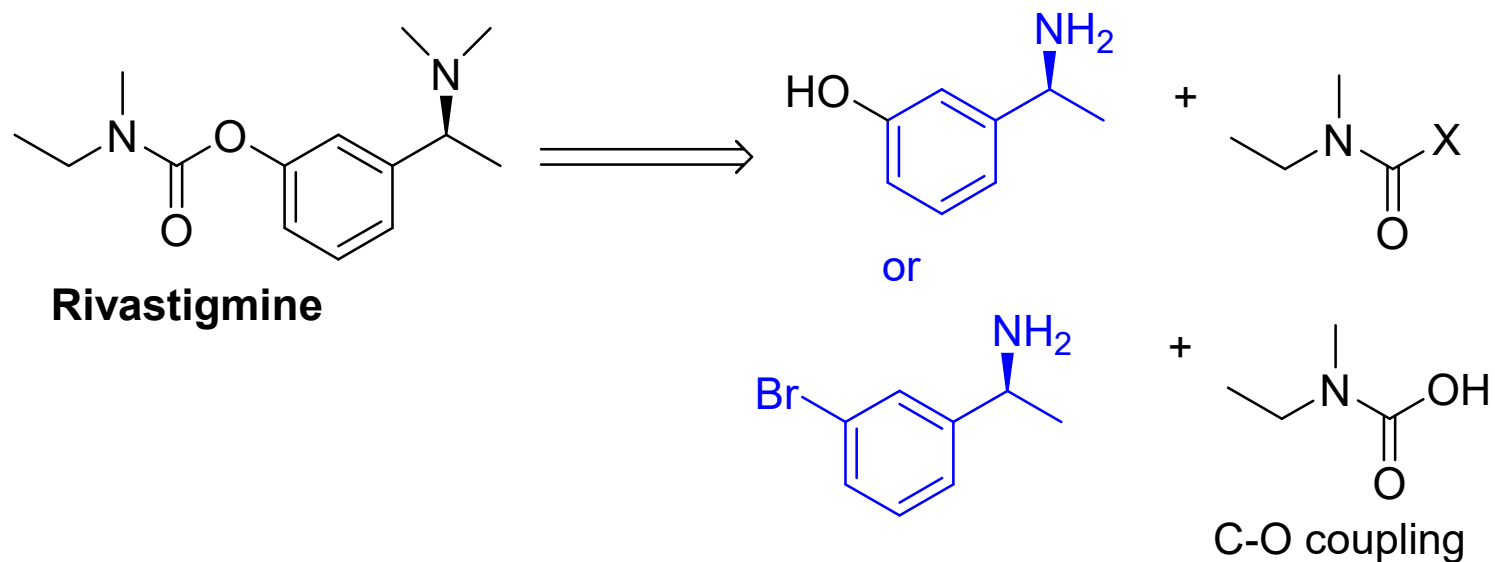
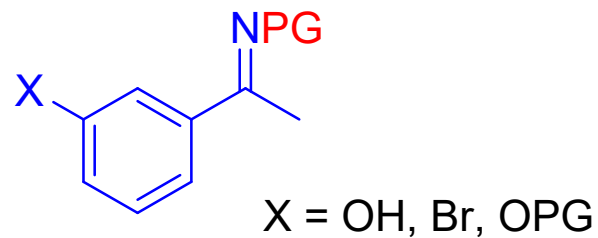
>99%, 85% ee (0 °C)



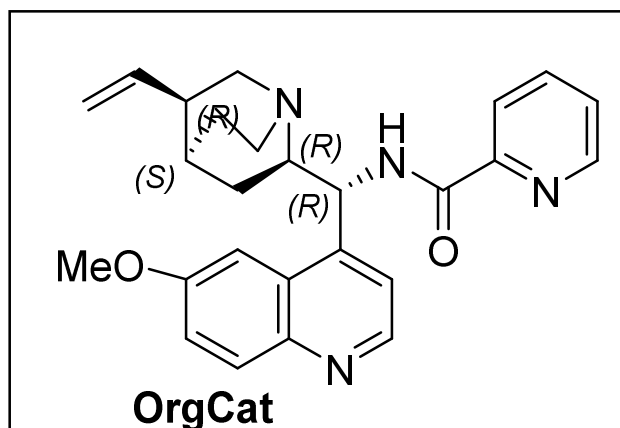
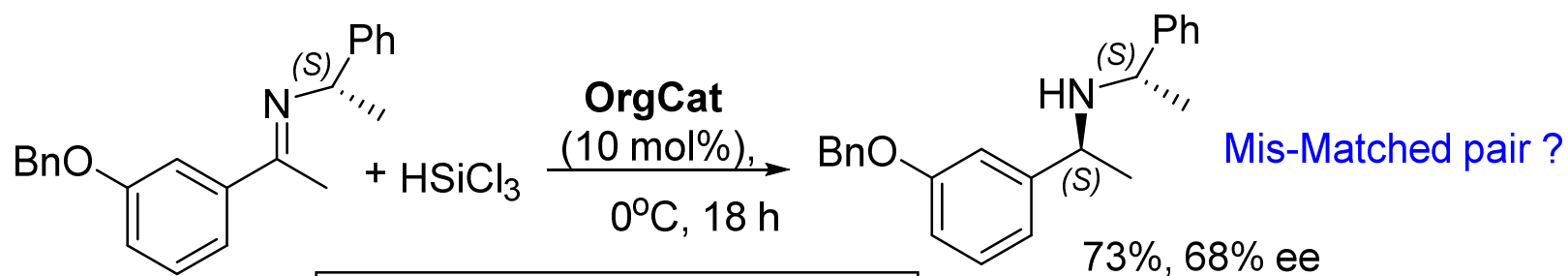
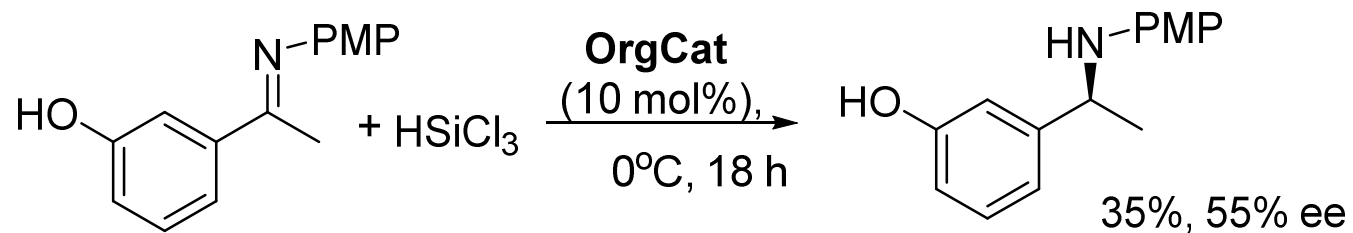
# Formal Route to Rivastigmine



Maurizio Benaglia

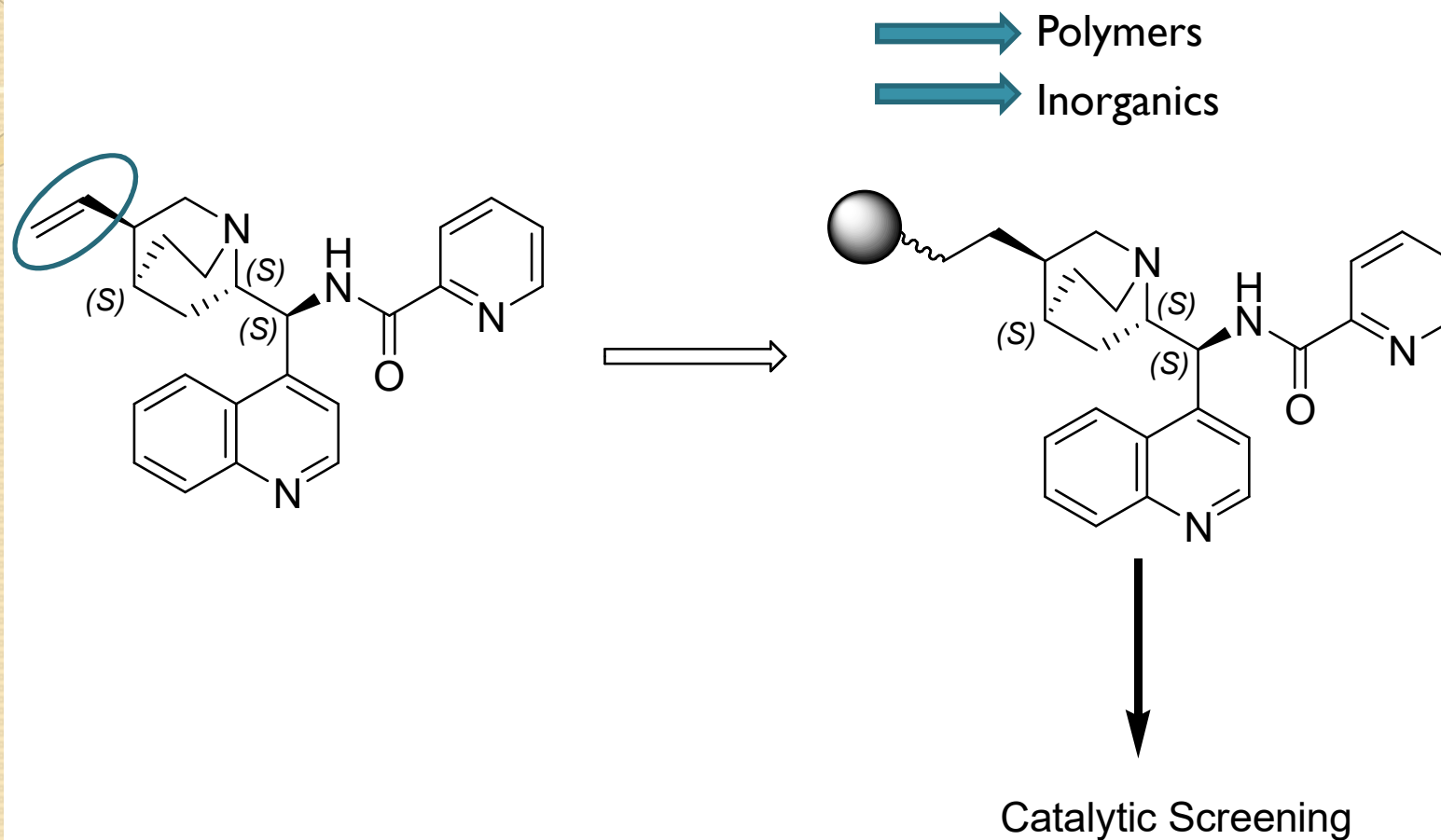


# Key reactions



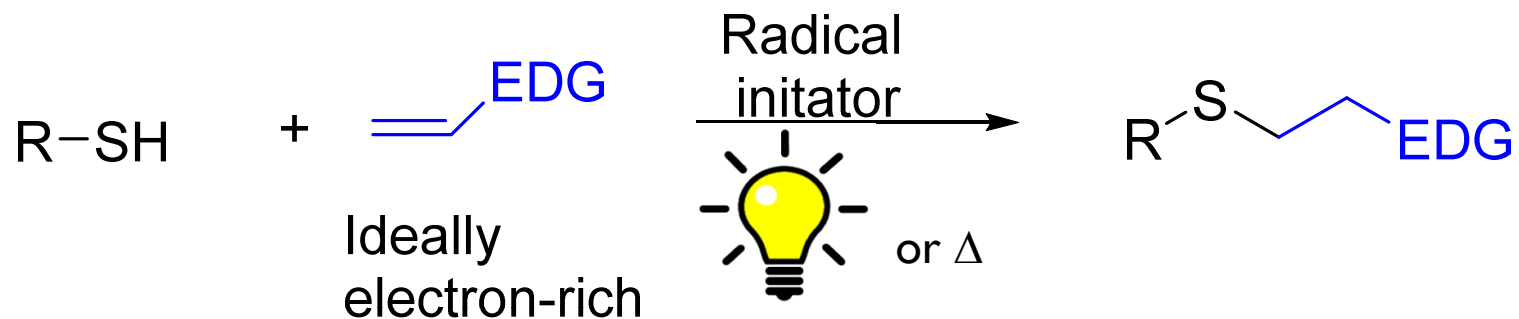


# Immobilized Picolinamide catalysts



# Thiol-Ene Click (TEC) Chemistry

(Morgan, Ketley, Grace, 70s)

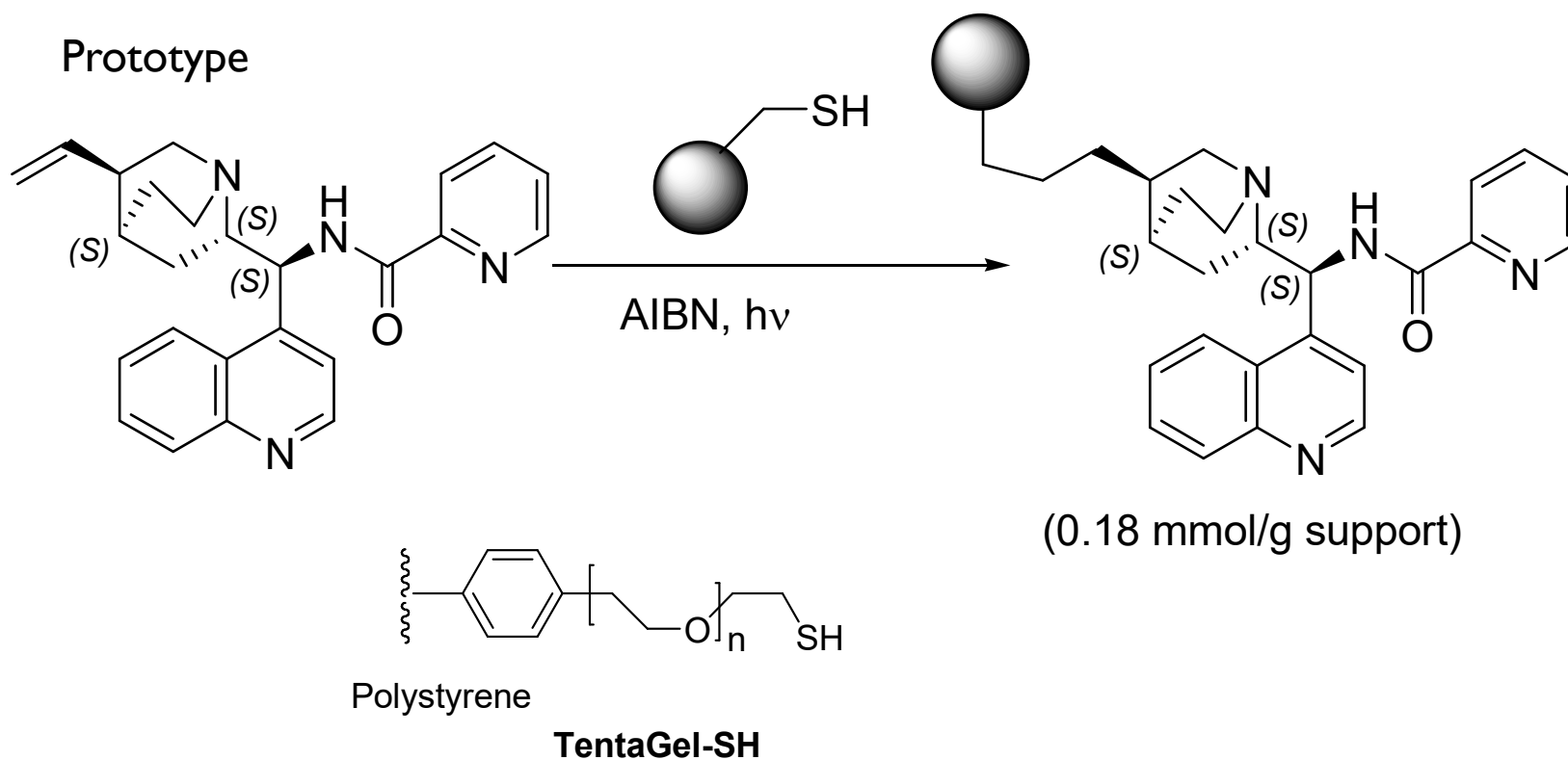


# Polymer immobilized Catalyst I

Thiol-ene chemistry (Garrell, *JACS*, **2011**, 133, 11026)



Marina Costa

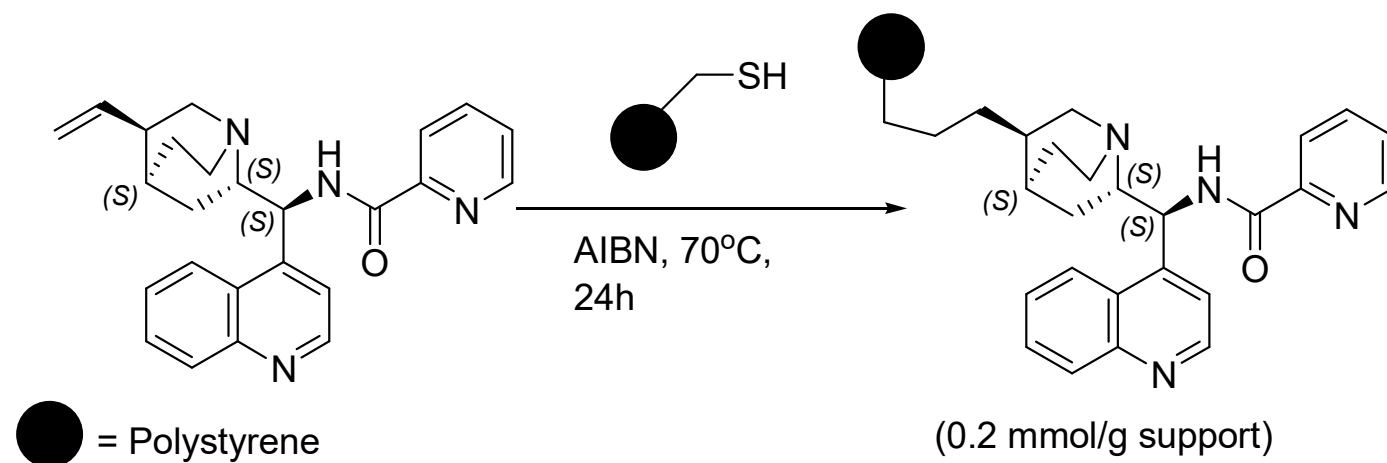


# Catalytic Cycles: Proof of Concept Studies

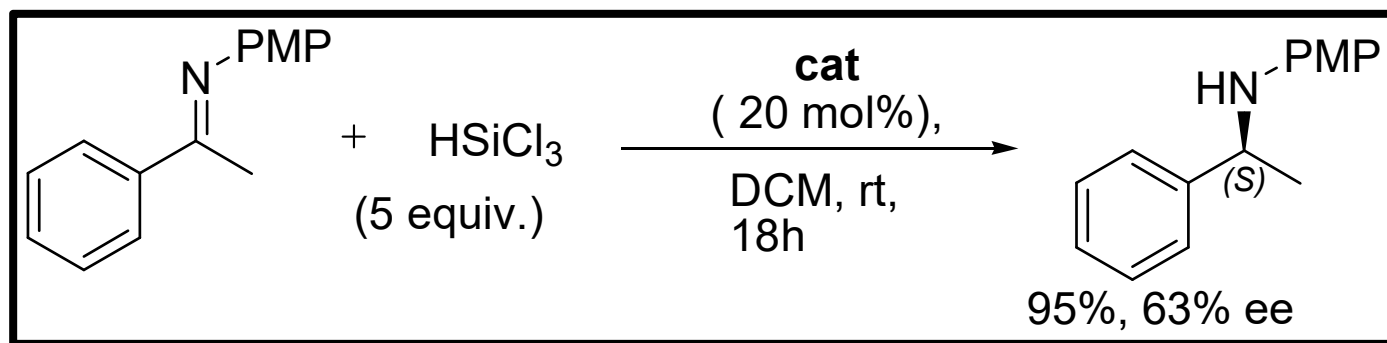


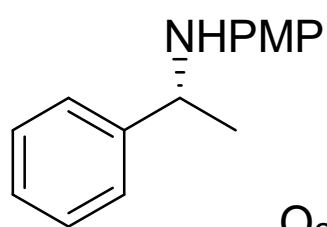
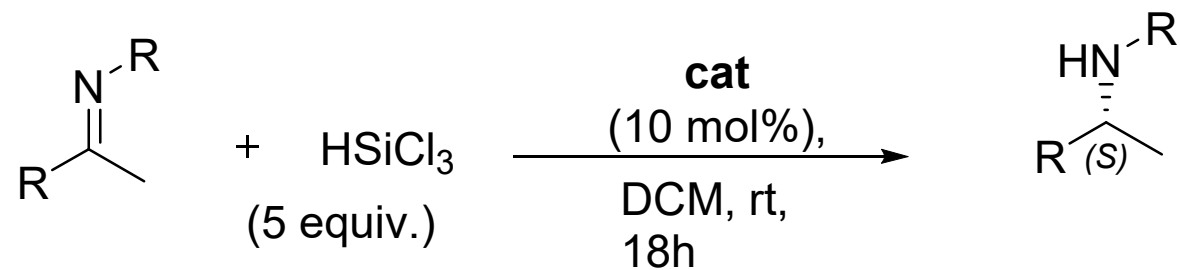
Cycle	Yield (%)	Ee (%)
1	80	69
2	21	12
3	22	19
4	46	1

# Polymer immobilized Catalyst II

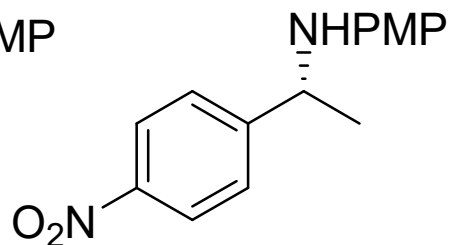


Braslau, *React. Funct. Polym.* **2013**, *73*, 624.

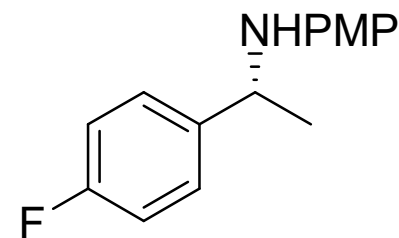




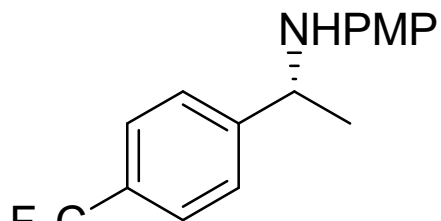
95%, 91%ee



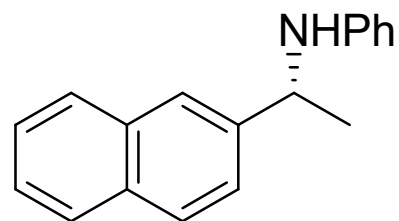
93%, 85%ee



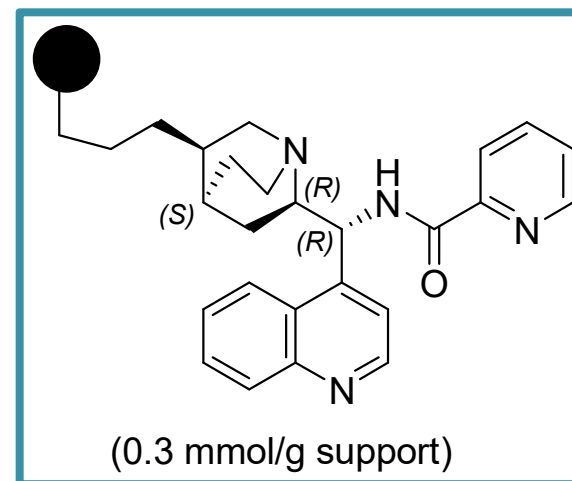
98%, 87%ee



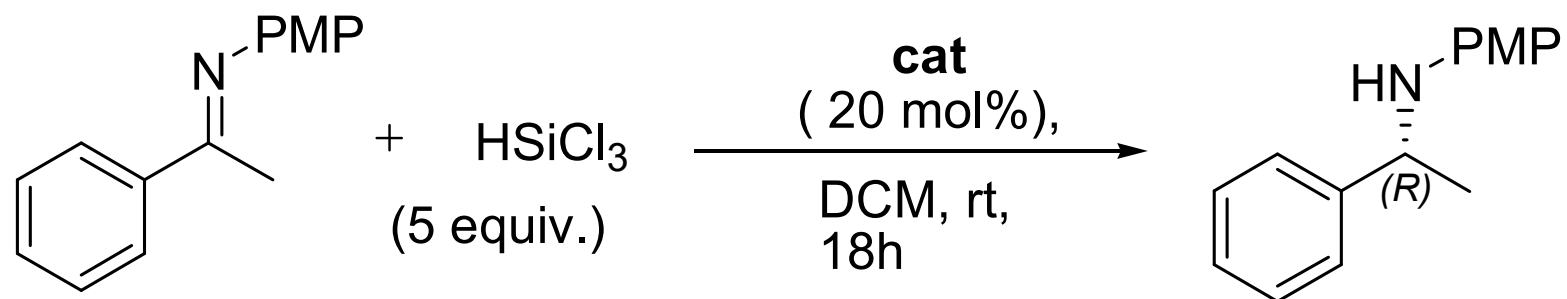
97%, 84%ee



96%, 85%ee

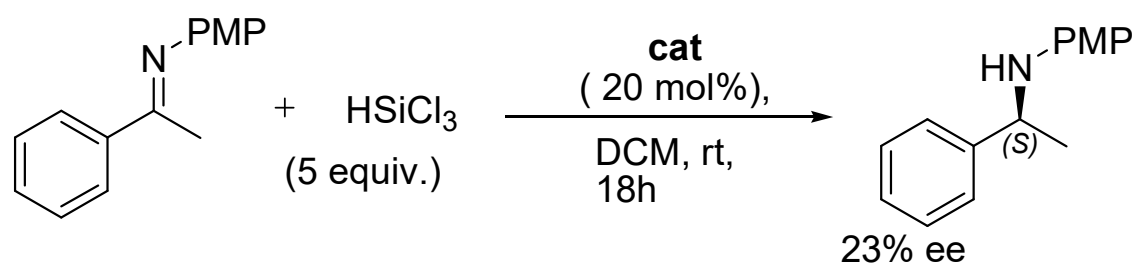
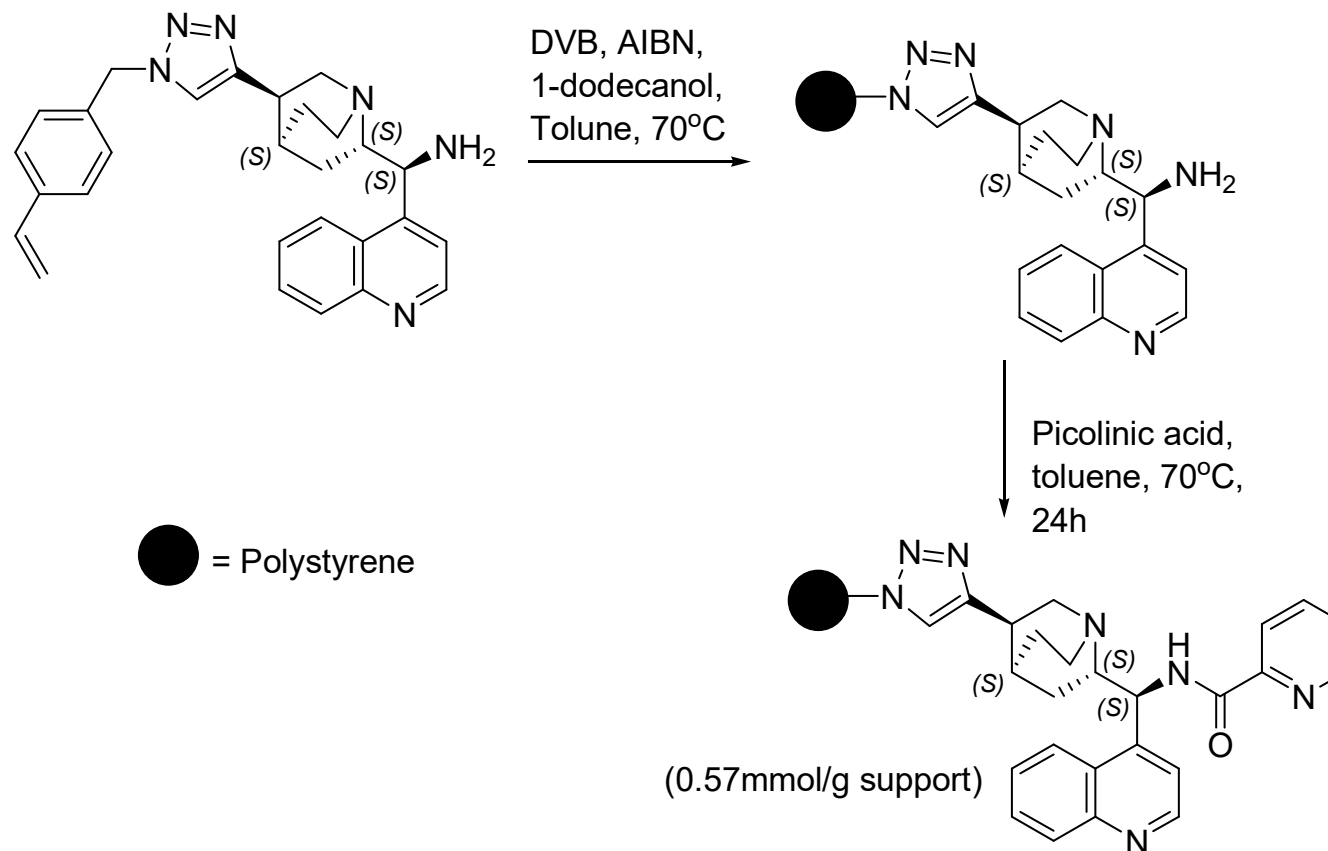


# Catalyst Recycling



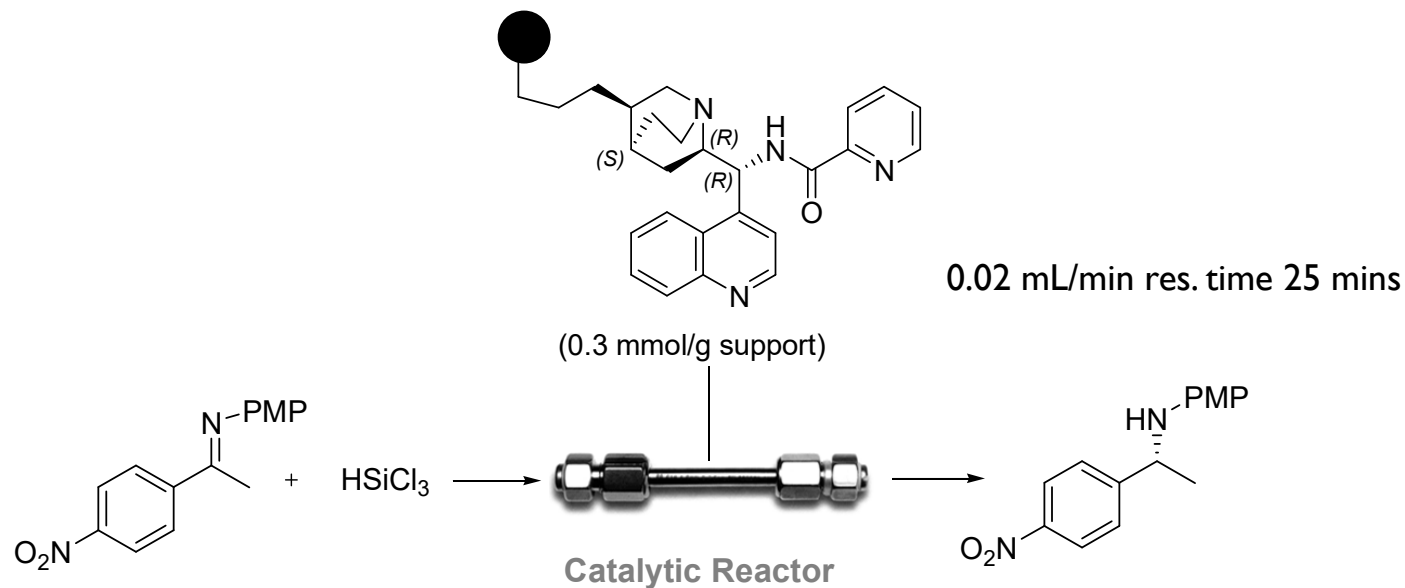
Run	Yield (%)	Ee (%)
1	95	91
2	90	88
3	80	67
4	77	37

# Catalyst Immobilization III



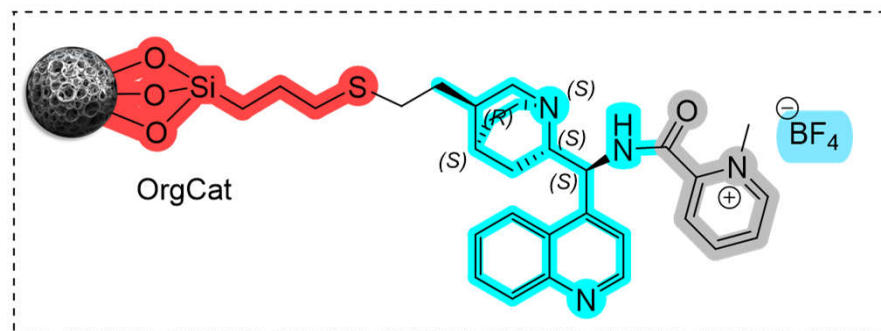


# Continuous Flow



Run	Time (mins)	Yield (%)	Ee (%)
1	0-50	98	47
2	50-75	98	44
3	75-100	93	30
4	100-125	93	25
5	125-150	87	22
6	150-175	86	20

# Immobilizations

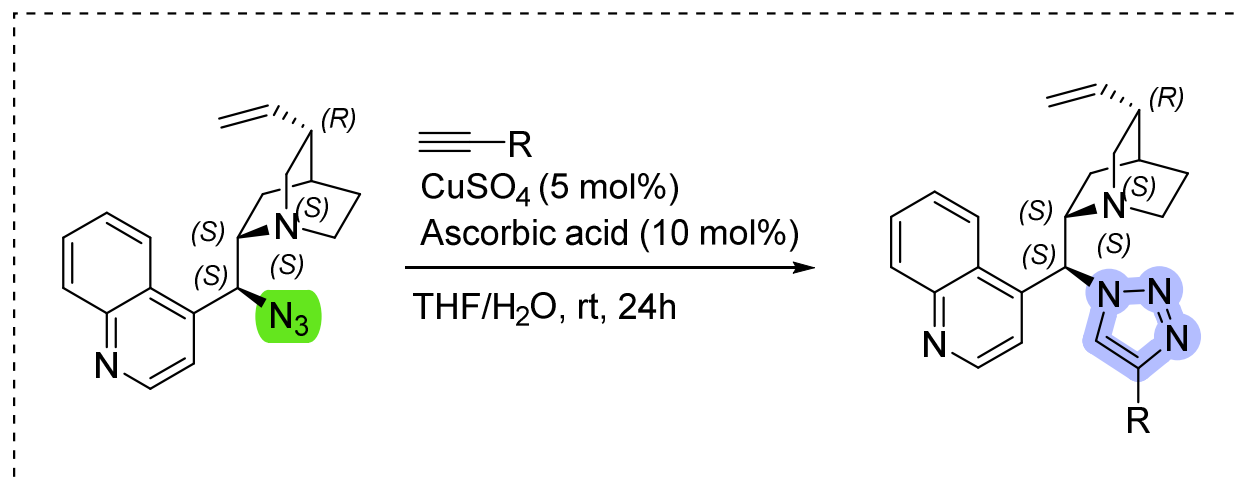


Entry	Support	Cycle	Yield	Ee (%)
1	NanoSilica	1	47	34
2	NanoSilica	2	50	38
3	NanoSilica	3	17	27
4	MCM-41	1	71	37
5	MCM-41	2	36	24
6	MCM-41	3	47	8
7	Magnetic NPs*	1	97	64
8	Magnetic NPs	2	30	50
9	Magnetic NPs	3	30	25

\*Magnetite derived ( $\text{Fe}_3\text{O}_4$ )

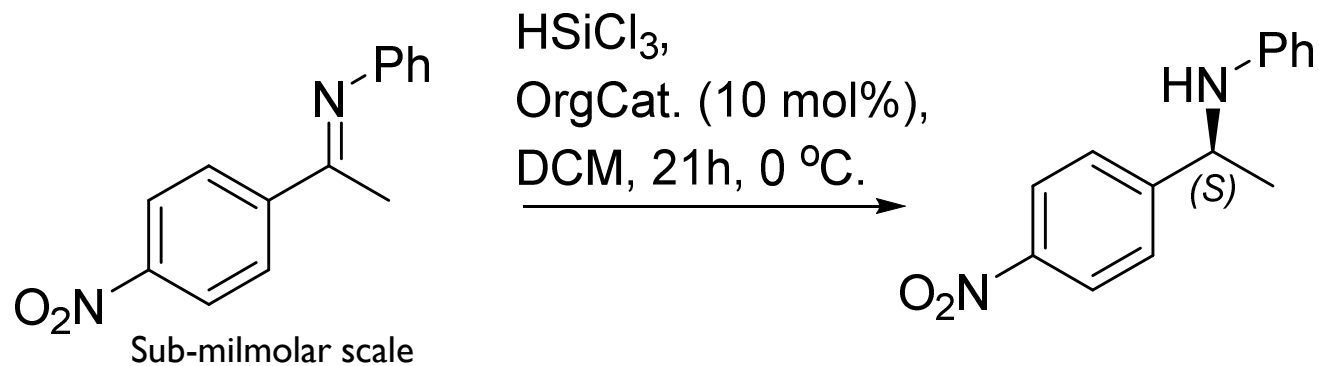


# CINCHONA- TRIAZOLES



Entry	R	Yield <sup>b</sup> (%)	Global yield <sup>c</sup> (%)	Product
1	-C <sub>6</sub> H <sub>5</sub>	85	76	<b>Ia</b>
3	-CH <sub>2</sub> OH	55 <sup>d</sup>	49	<b>Ib</b>
4	-(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	76	68	<b>Ic</b>
5	-C(CH <sub>3</sub> ) <sub>2</sub> NH <sub>2</sub>	98 <sup>e</sup>	87	<b>Id</b>
6	-C(CH <sub>3</sub> ) <sub>2</sub> OH	97 <sup>d</sup>	87	<b>Ie</b>
7	-cyclopropyl	69 <sup>e</sup>	62	<b>If</b>
8	-6-methoxynaphthalene	67 <sup>e</sup>	60	<b>Ig</b>

Barrulas, *Synth. Communications*, **2021**, 51, 2954.



Entry	Organocatalyst	Yield <sup>b</sup> (%)	ee <sup>c</sup> (%)
1	1a	61	3
2	1b	60	Rac
3	1c	58	Rac
4	1d	74	Rac
5	1e	80	Rac
6	1f	27	10
7	1g	23	10

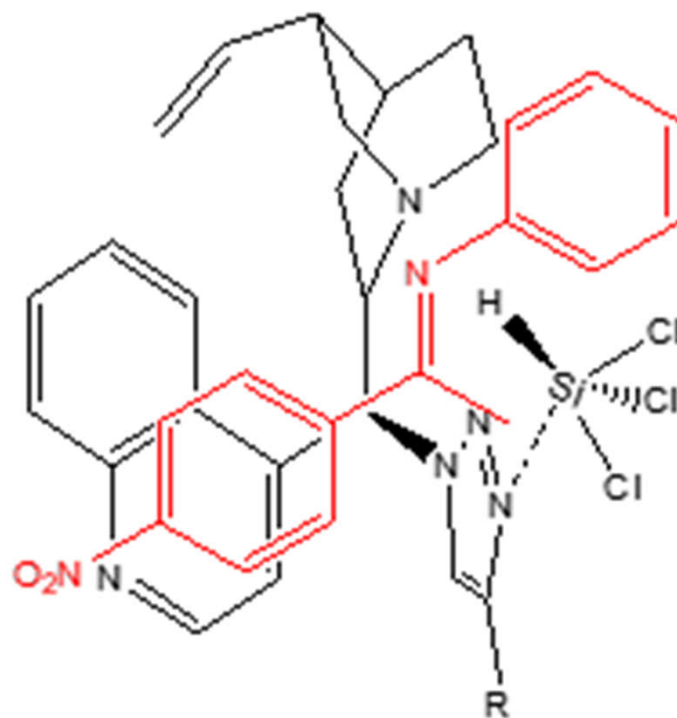
# Enantioselectivity Optimization: Solvent screening

Entry	Solvent	I f		I g	
		Yield (%)	ee (%)	Yield (%)	ee (%)
1	DCM	27	10	23	10
2	THF	54	11	52	10
3	DMF	52	10	41	12
4	Toluene	4	12	23	9
5	Dioxane	74	10	53	9

Barrulas, *Synth. Communications*, **2021**, 51, 2954.

Nordic-Irish Process Chemistry Forum  
2023

# Working Model

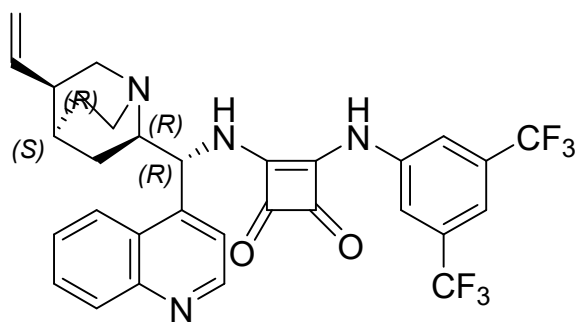


Barrulas, *Synth. Communications*, **2021**, 51, 2954.

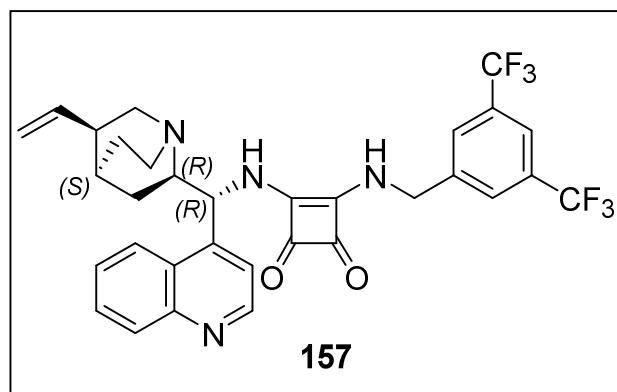
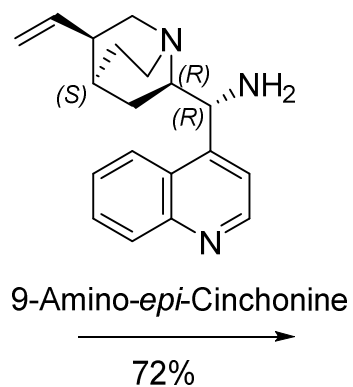
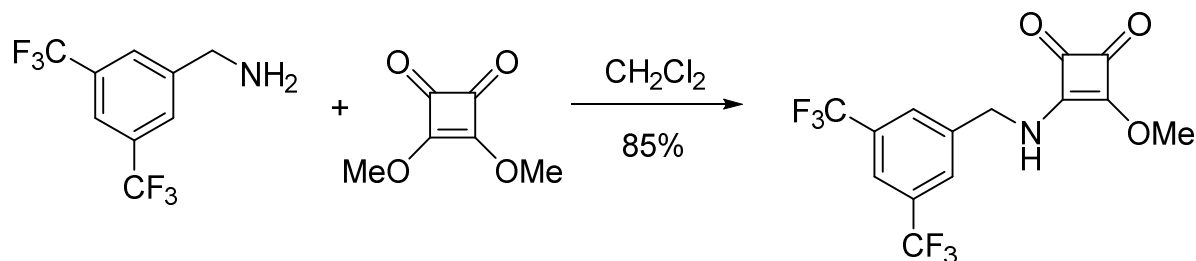
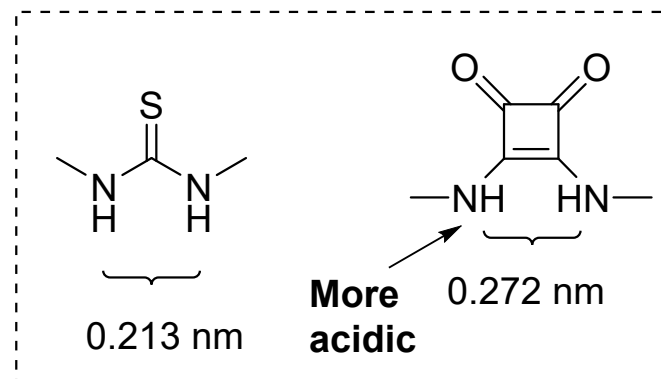


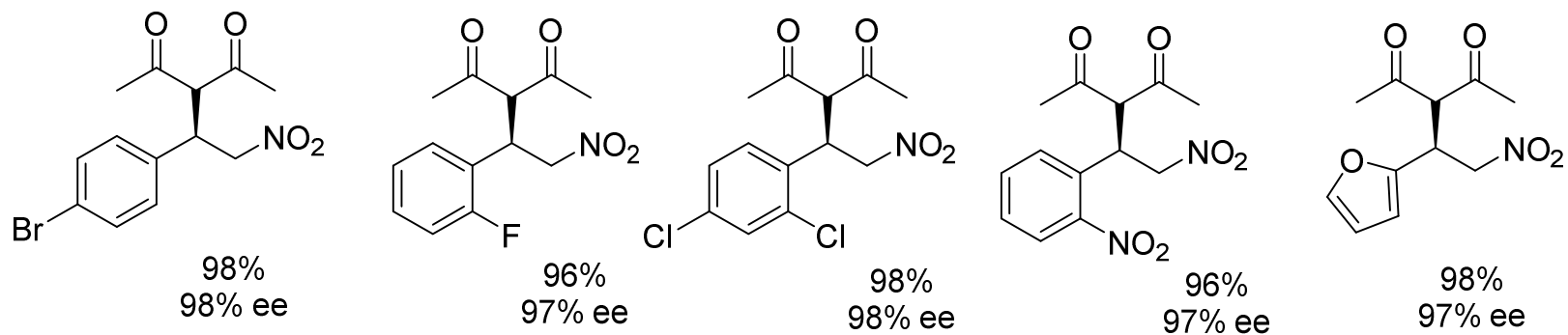
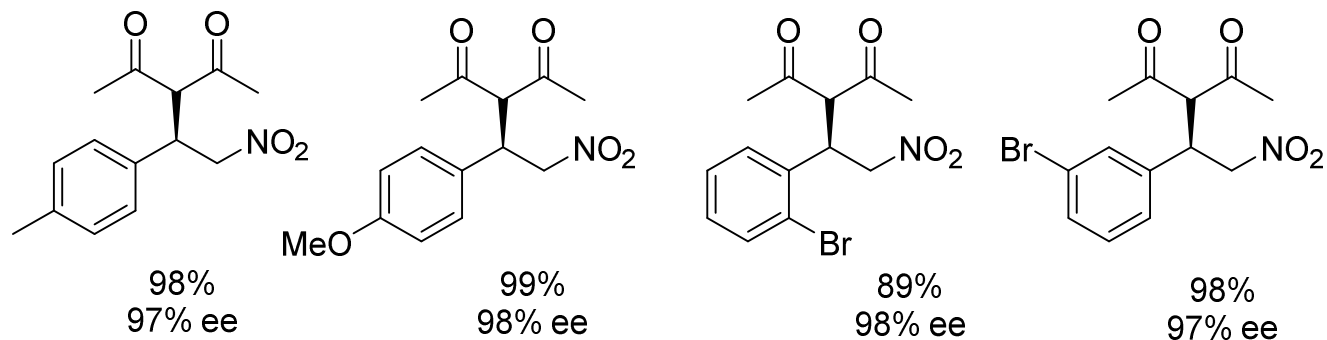
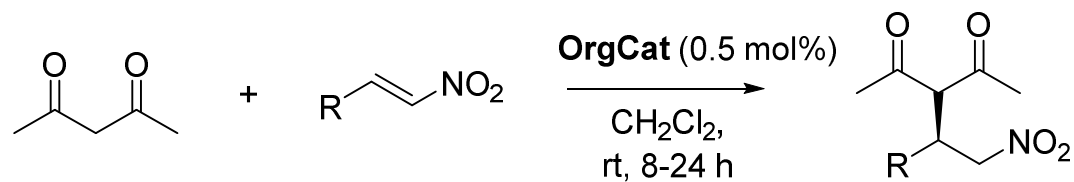
# **CINCHONA- SQUARAMIDES**



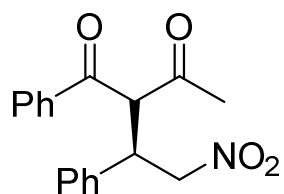
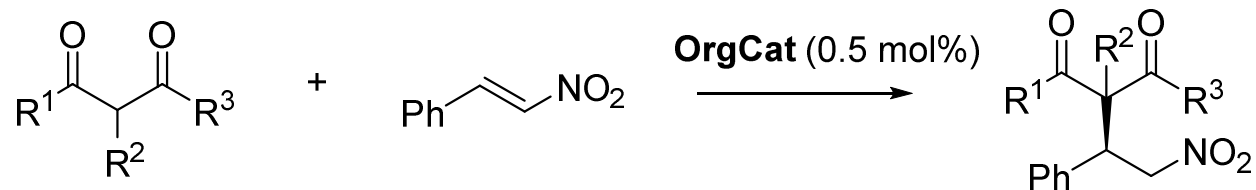


Rawal Cinchona-Squaramide organocatalyst

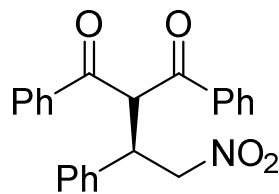




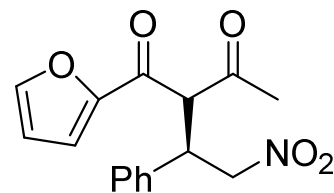
*J. Am. Chem. Soc.* **2008**, 130, 14416.



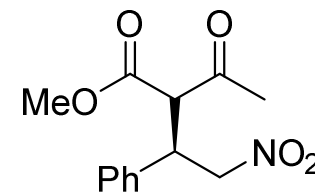
91% (dr = 2:1)  
95% ee (major)  
94% ee (minor)



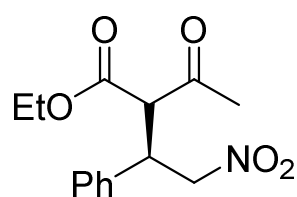
93%  
88% ee



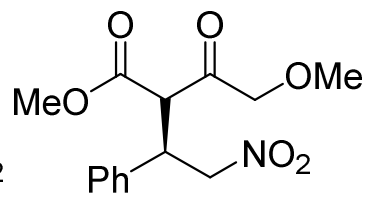
97% (dr = 1.4:1)  
92% (major)  
81% (minor)



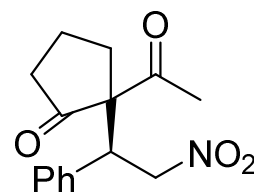
75% (dr = 1:1)  
96% (major)  
98% (minor)



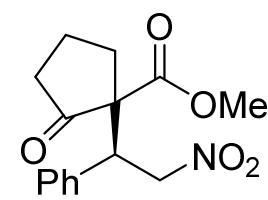
89% (dr = 1.6:1)  
97% ee (major)  
96% ee (minor)



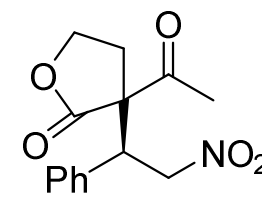
83% (dr = 1.3:1)  
94% ee (major)  
86% ee (minor)



96% (dr = 18:1)  
98% (major)  
96% (minor)

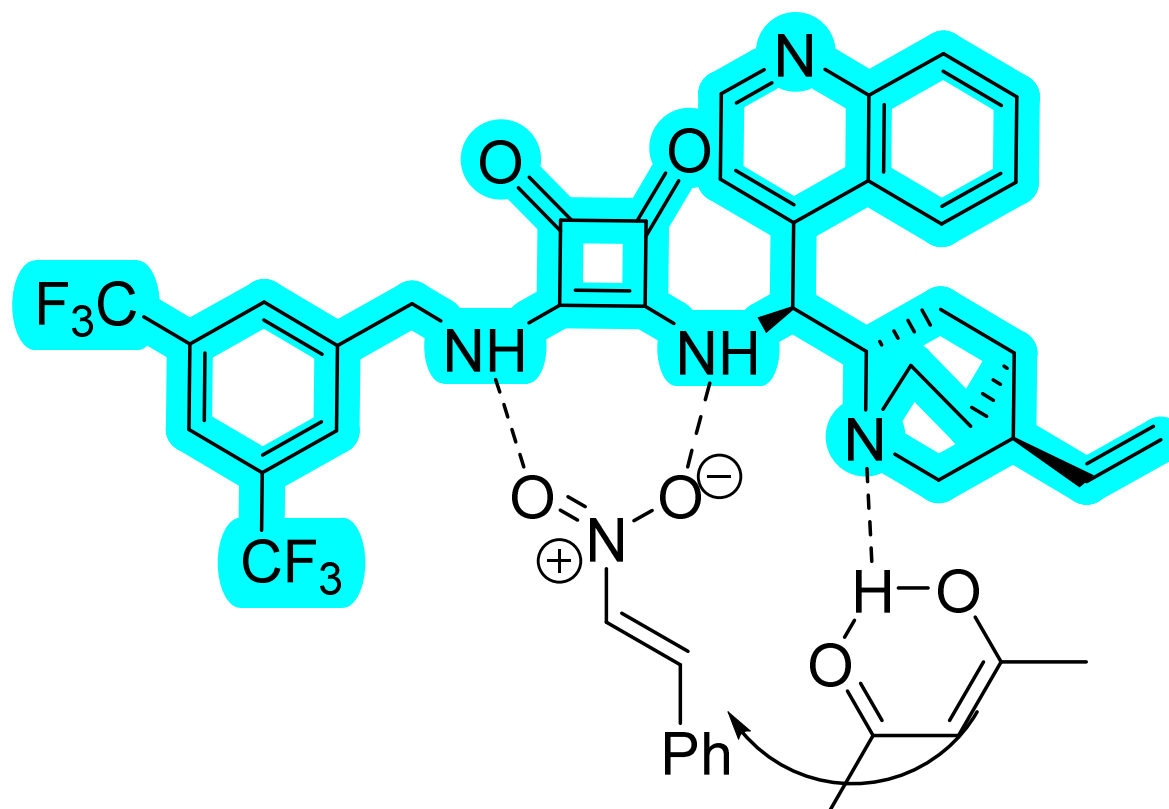


75% (dr = 50:1)  
93% (major)  
77% (minor)

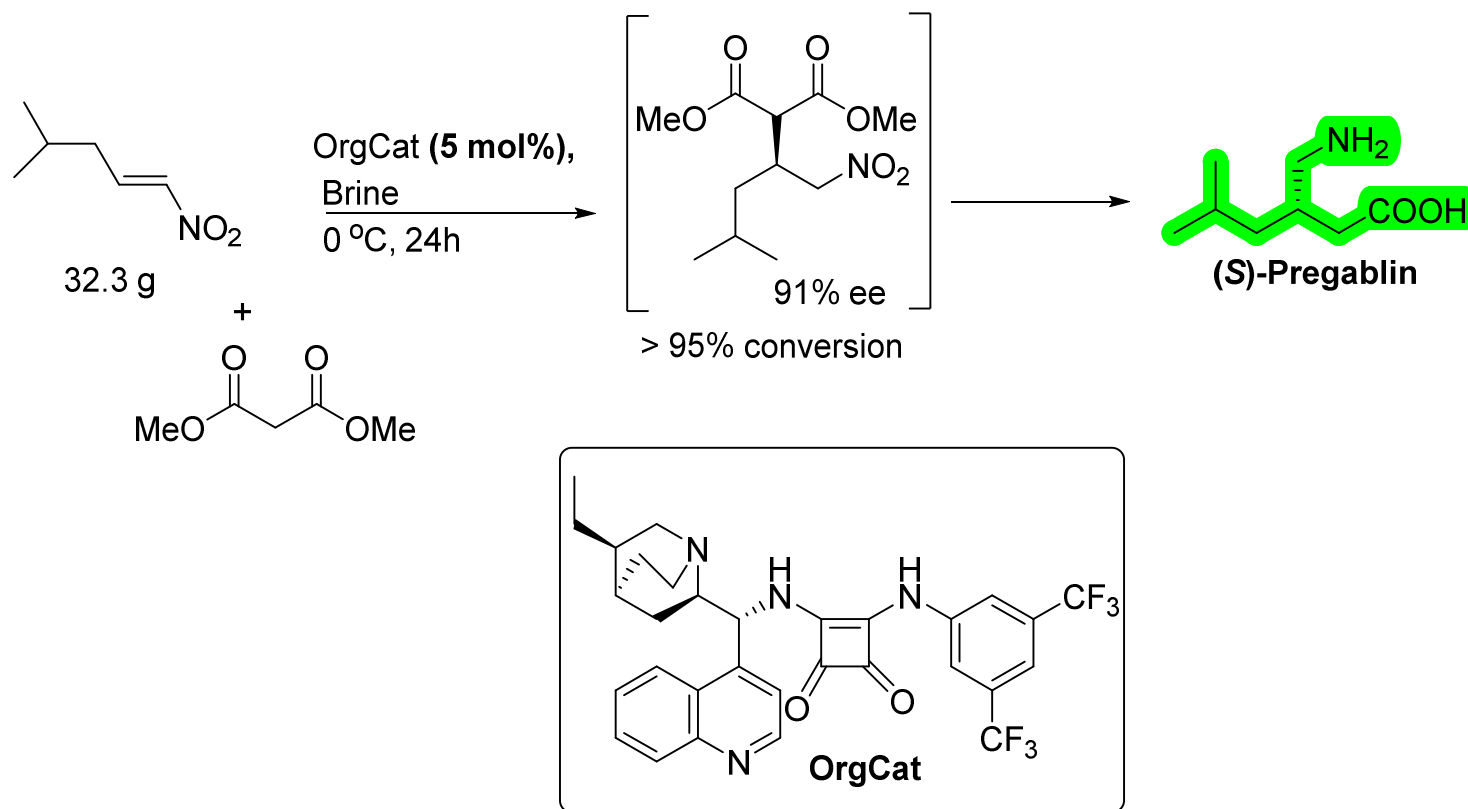


65% (dr = 4:1)  
94% (major)  
91% (minor)

## Rawal Bifunctional Catalyst: Lewis base/Brønsted acid

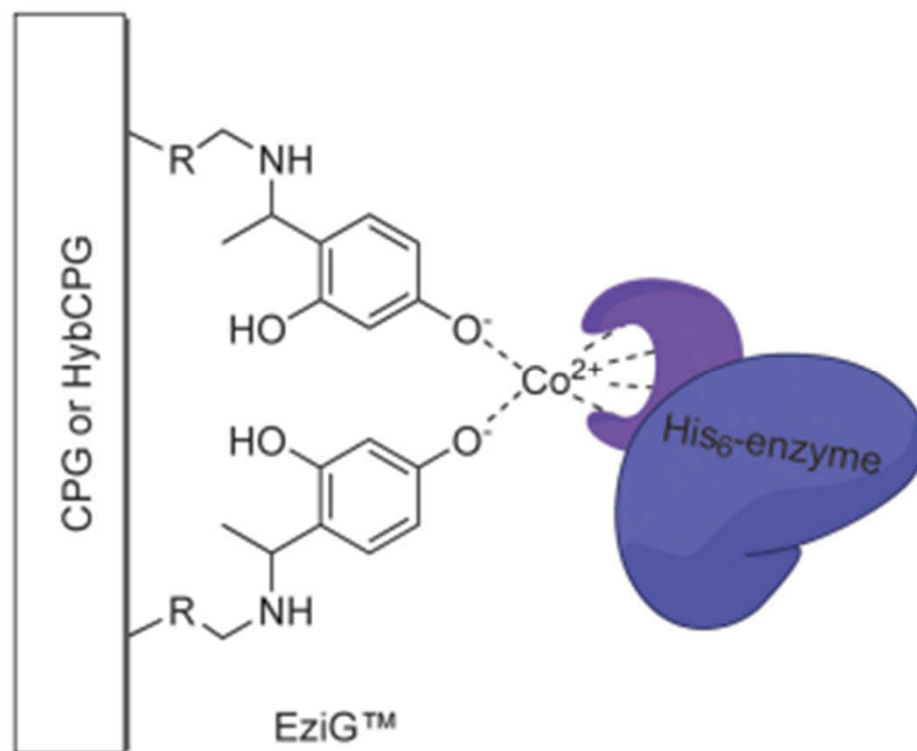


# (S)-Pregabalin (Lyrica®)



Bae and Song, *ACS Cat.* **2015**, 5, 3613.

# Controlled Porosity Glass (CPG) Beads



ChemComm

COMMUNICATION



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## A general protein purification and immobilization method on controlled porosity glass: biocatalytic applications†

Cite this: *Chem. Commun.*, 2014, 50, 9134

Received 8th April 2014,  
Accepted 20th June 2014

DOI: 10.1039/C4CC00000A

K. Engelmark Cassimjee,<sup>a\*</sup> M. Kadow,<sup>b</sup> Y. Wikmark,<sup>b</sup> M. Svedendahl Humble,<sup>b</sup>  
M. L. Rothstein,<sup>b</sup> D. M. Rothstein<sup>b</sup> and J.-E. Backvall<sup>a\*</sup>

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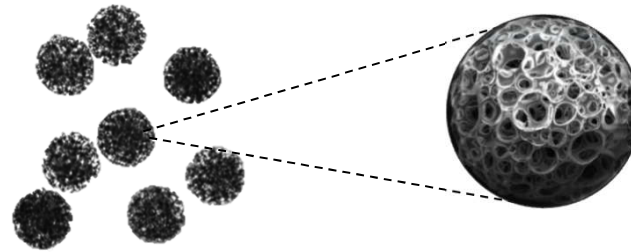
**Opal**  
Hydrophilic  
derivatized silica  
surface



**Amber**  
Semi-  
hydrophobic  
polystyrene  
derivative  
surface



**Coral**  
Hydrophobic  
surface with  
poly(vinylbenzyl  
chloride)  
coating



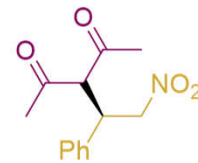
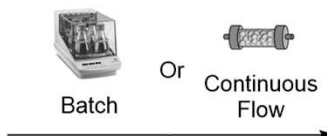
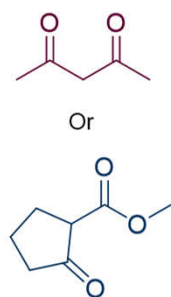
### Continuous Flow

### Batch mode



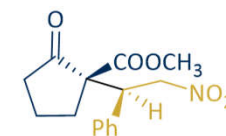


17

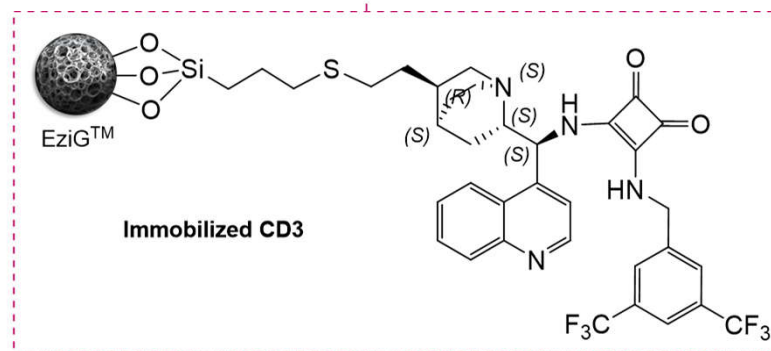


**1.6 mol% of catalyst**  
**Batch:** up to 4 cycles  
 (up to 93% yield and 89% ee)  
**Flow:** up to 4 cycles  
 (up to 99% yield and 92% ee)

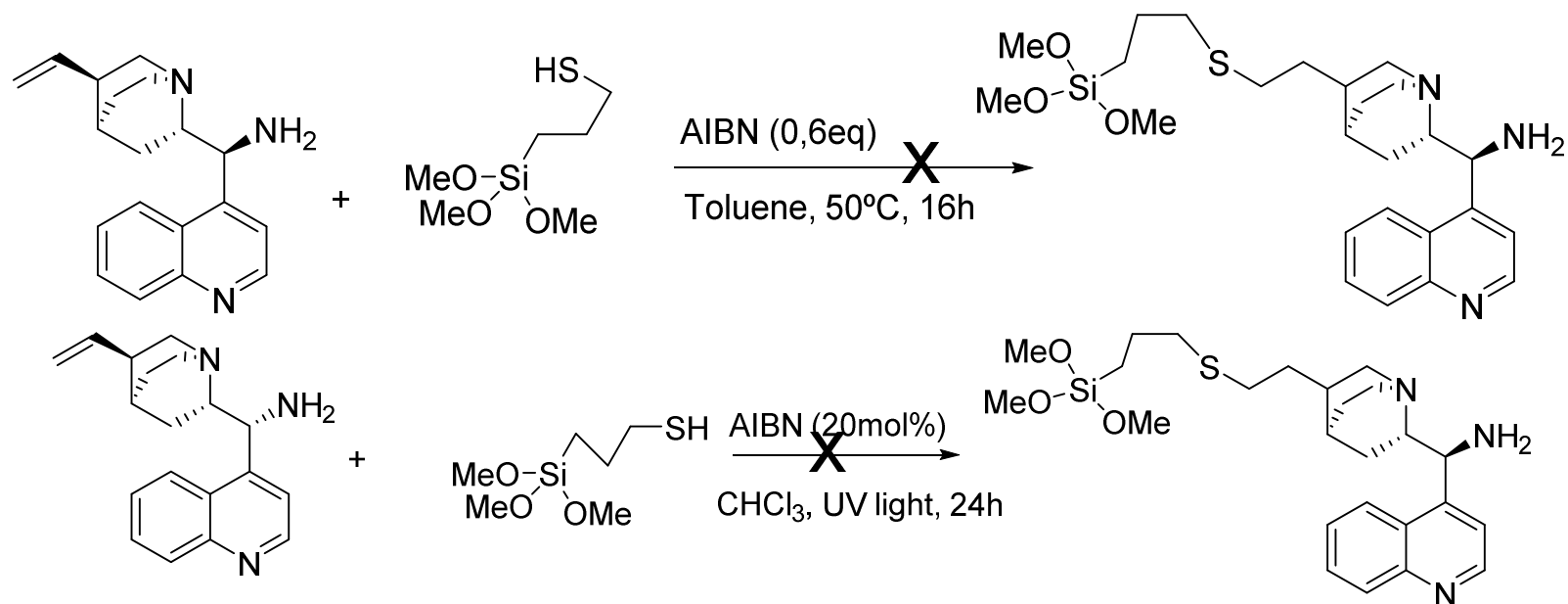
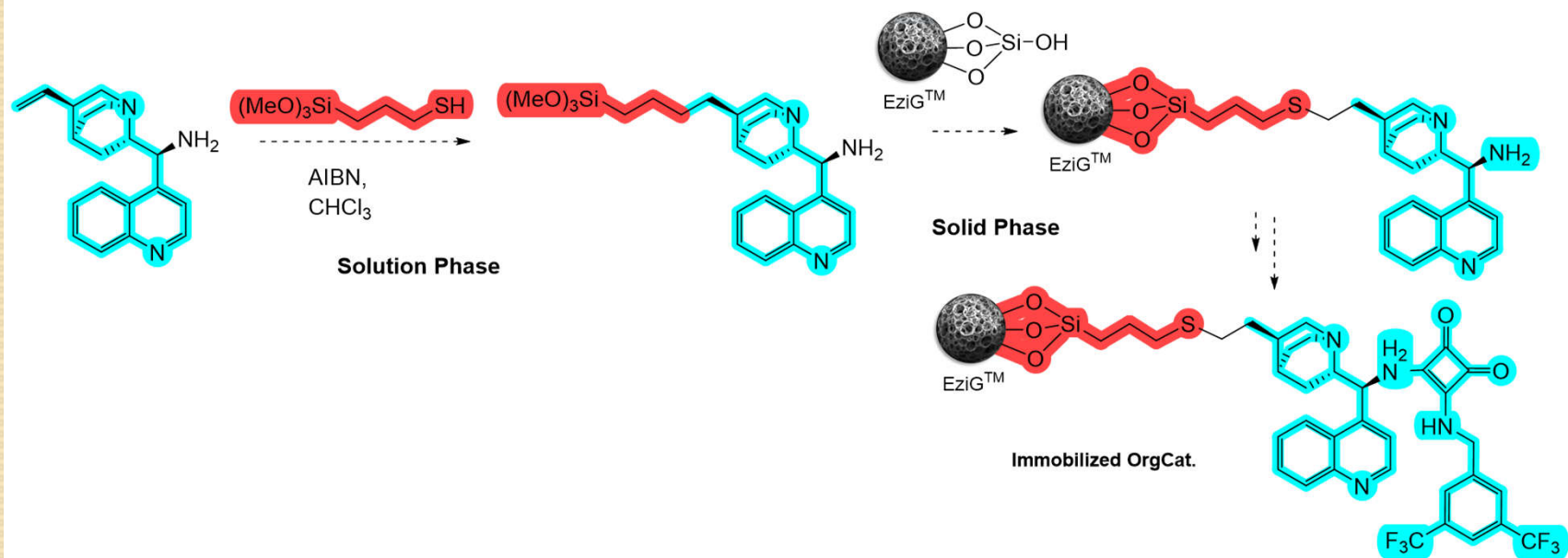
Or

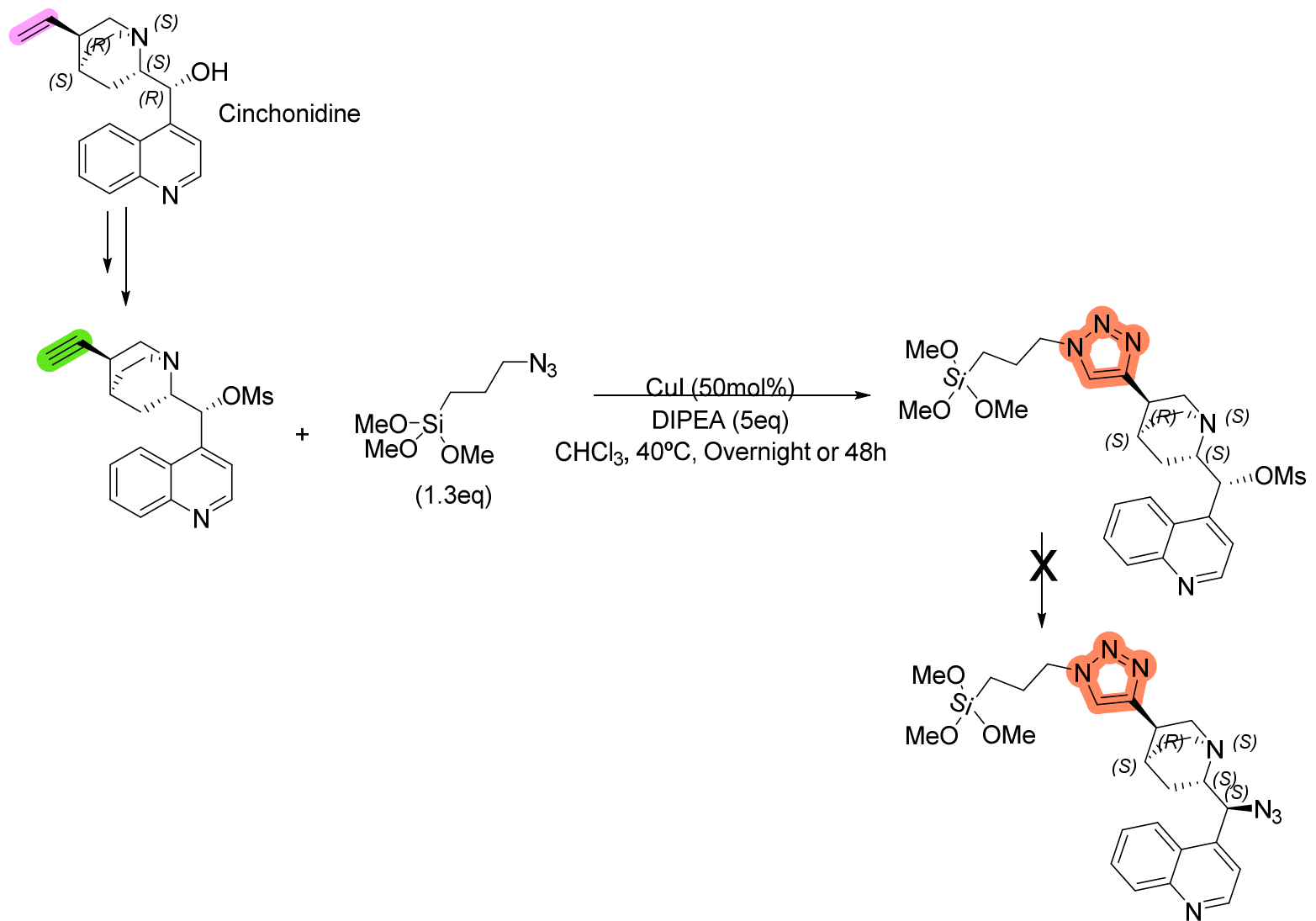


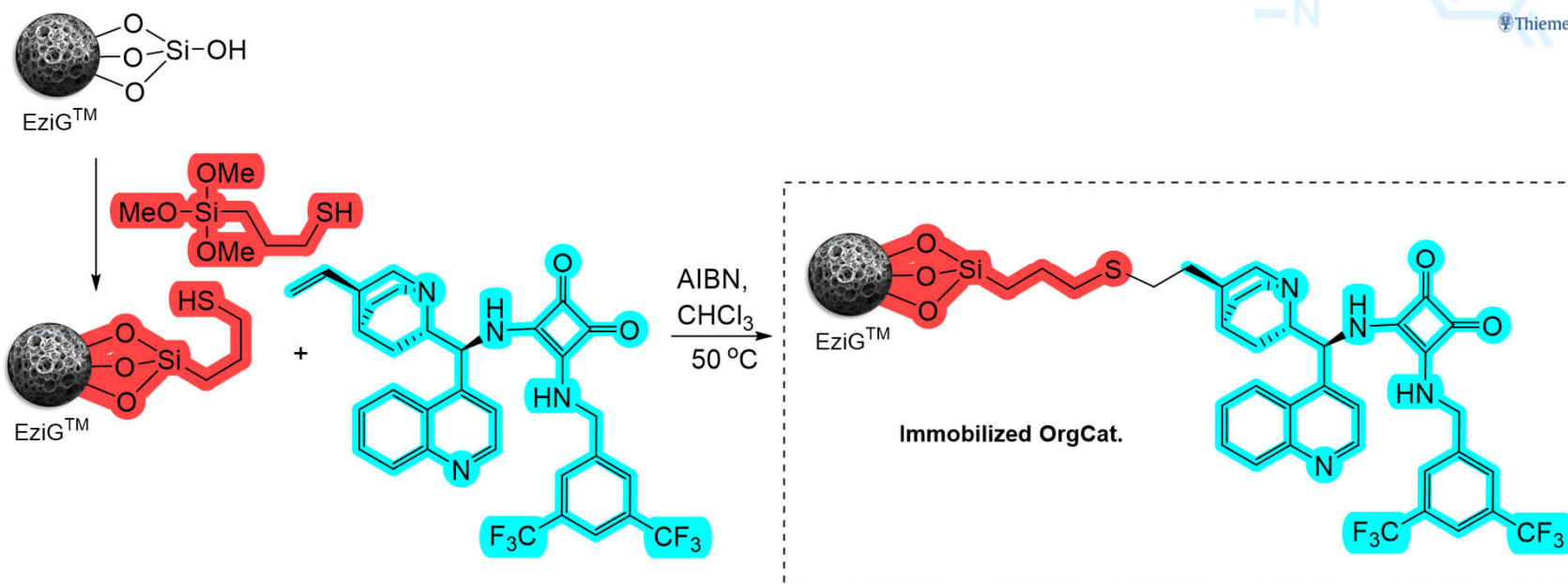
**0.8 mol% of catalyst**  
**Batch:** up to 7 cycles  
 (up to 99% yield, 99% ee and 0,2:99,8 dr)  
**Flow:** up to 5 cycles  
 (up to 99% yield, 97% ee and 0,4:99,6 dr)

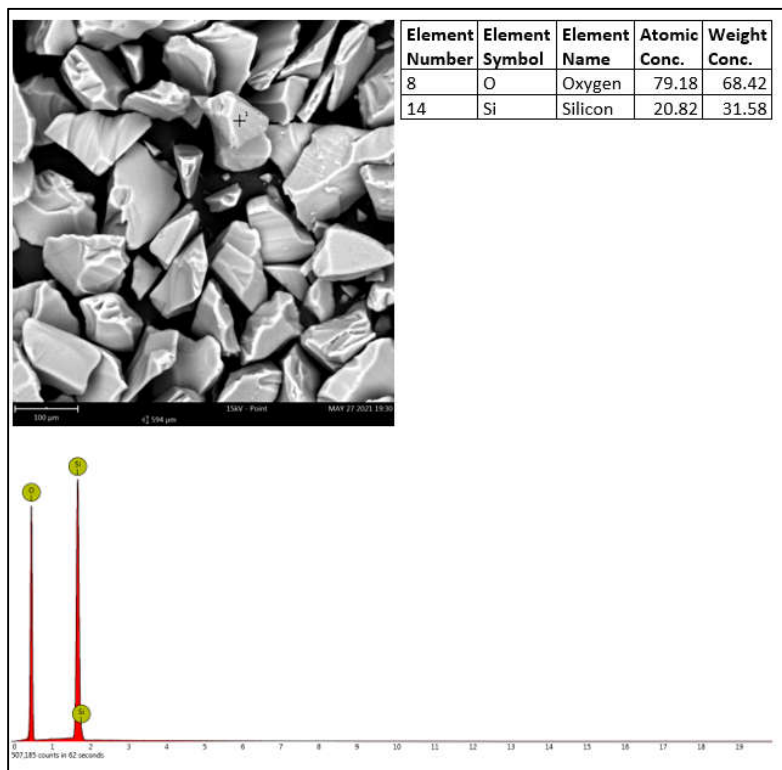




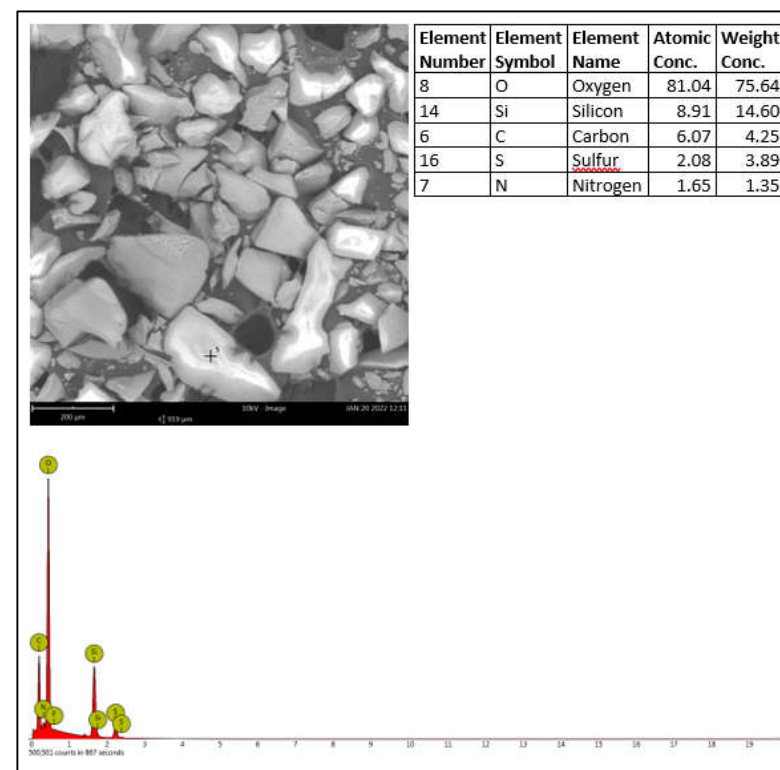








Without catalyst on amber (SEM)



With catalyst on amber (SEM)

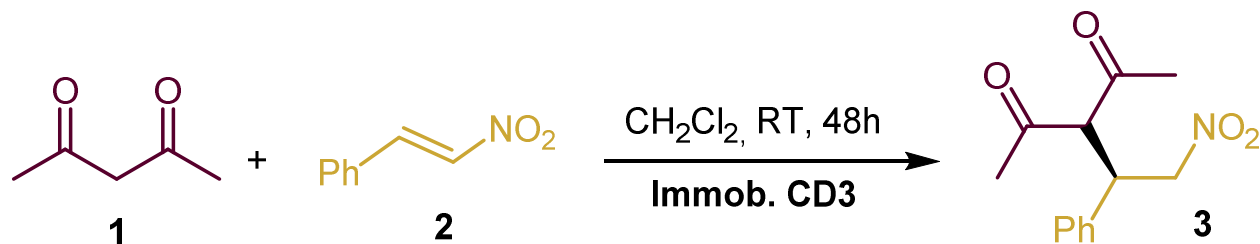


TABLE 1

Entry	Immobilized OrgCat	f (mmol g <sup>-1</sup> )	Amount of beads used (g)	Cycles	Yield (%)	ee (%)
1	OCD3 (OPAL)	0.05176	0.055	1	86	89
2				93	84	
3				56	31	
4				22	77	
5	ACD3 (AMBER)	0.13386	0.030	1	54	87
6				1	65	
7	CCD3 (CORAL)	0.14457	0.028	1	53	78
8				2	6	55

$$f \text{ (mmol g}^{-1}\text{)} = \%N \times 1000 \times (\text{number of N atoms})^{-1} \times \text{MW(N)}^{-1} \times 100^{-1}$$

(Chem. Sci. 2019, 10, 11141)

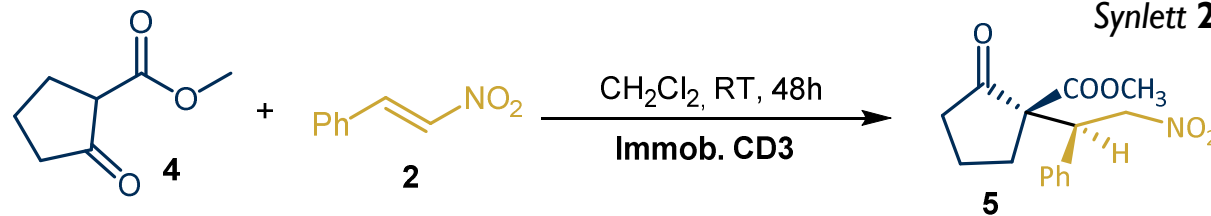
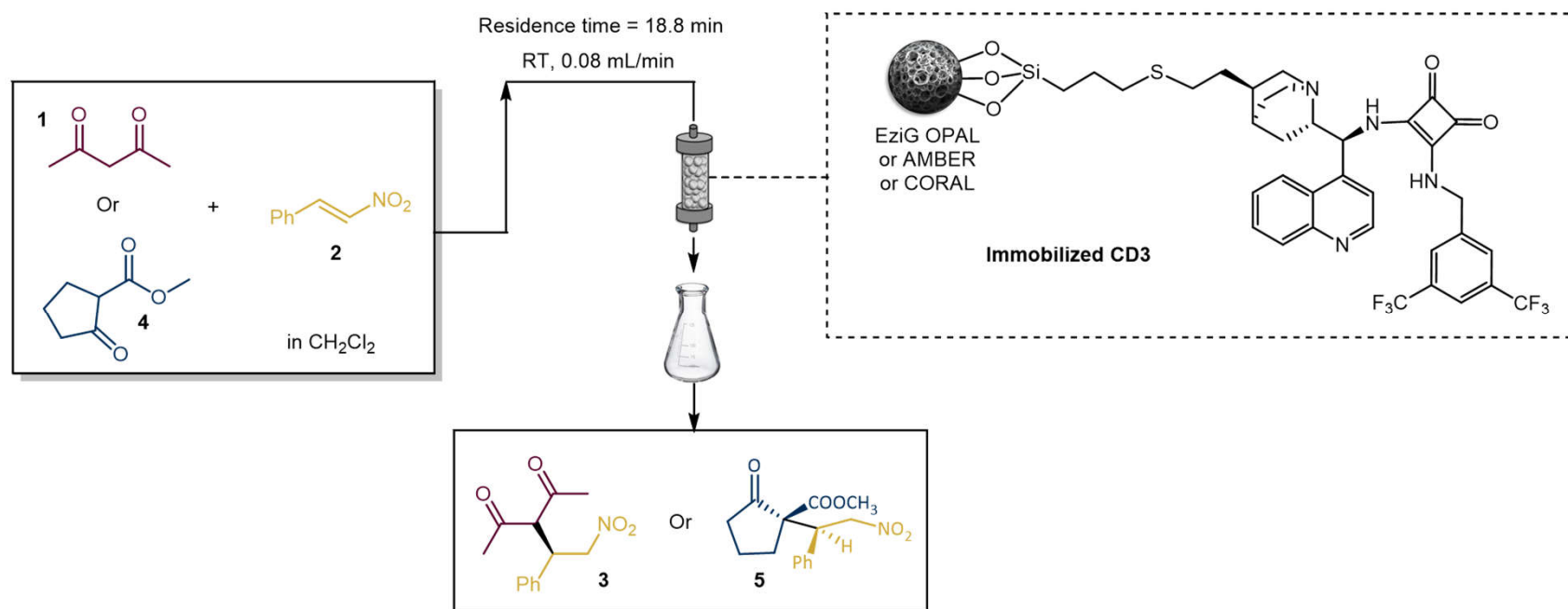


TABLE 2

Entry	Immobilized catalyst	f (mmol g <sup>-1</sup> )	Cycles	Yield (%)	ee (%)	dr
<b>1</b>	<b>OCD3</b>	0.05176	1	88	99	0.3:99.7
<b>2</b>			2	98	95	0.4:99.6
<b>3</b>			3	99	63	0.8:99.2
<b>4</b>			4	99	86	0.2:99.8
<b>5</b>			5	97	96	0.3:99.7
<b>6</b>			6	47	>99	1.2:98.8
<b>7</b>			7	62	97	1.6:98.4
<b>8</b>	<b>ACD3</b>	0.12493	1	84	92	0.1:99.9
<b>9</b>			2	66	88	0.3:99.7
<b>10</b>			3	53	77	1.4:98.6
<b>11</b>	<b>CCD3</b>	0.07853	1	83	93	0.2:99.8
<b>12</b>			2	60	81	0.2:99.8
<b>13</b>			3	21	79	1.5:98.5

Synlett **2022**, 33(17) 1756.





*Synlett* **2022**, 33(17) 1756.

Table 3

Entry	Immobilized Org. Cat.	f (mmol g <sup>-1</sup> )	Amount of Beads used (g)	Cycles	Yield (%)	ee (%)
1	<b>OCD3</b>	0.08567	0.277	1	10	69
2				68	86	
3				-g	-g	
4	<b>ACD3</b>	0.12493	0.173	1	66	95
5				29	95	
6				13	47	
7	<b>CCD3</b>	0.07853	0.179	1	99	92
8				87	91	
9				23	88	
10 <sup>c</sup>				16	61	

<sup>g</sup> Contained impurities



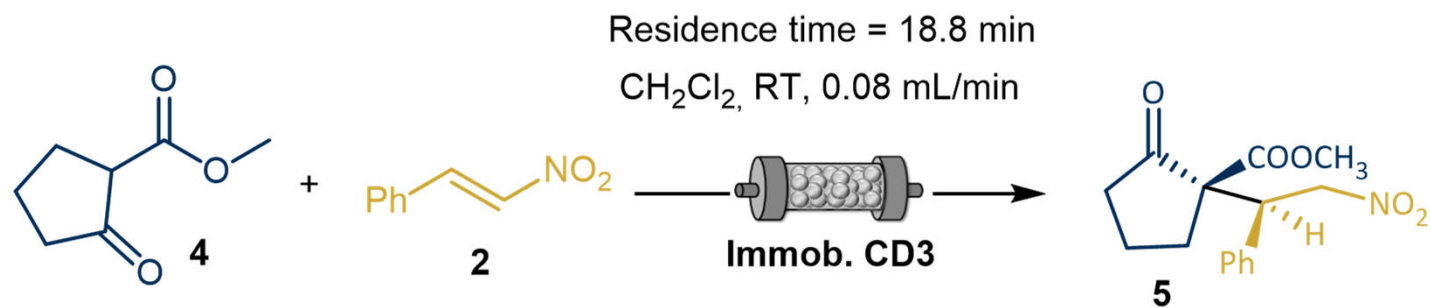


Table 4

Entry	Immobil. OrgCat	f (mmol g <sup>-1</sup> )	Amount of Beads used (g)	Cycles	Yield (%)	ee (%)	dr
1	OCD3	0.08567	0.442	1	24	-g	-g
2	ACD3	0.15706	0.286	1	99	-g	-g
3				2	30	-g	-g
4				3	10	-g	-g
5				1	99	95	0.6:99.4
6	CCD3	0.07853	0.183	2	38	94	0.8:99.2
7				3	47	92	1.3:98.7
8				4	43	86	4.9:95.1
9				5	28	97	0.4:99.6

g Contained impurities



# DEEP EUTECTIC SOLVENTS

## Natural Deep Eutectic Solvents – Solvents for the 21st Cen

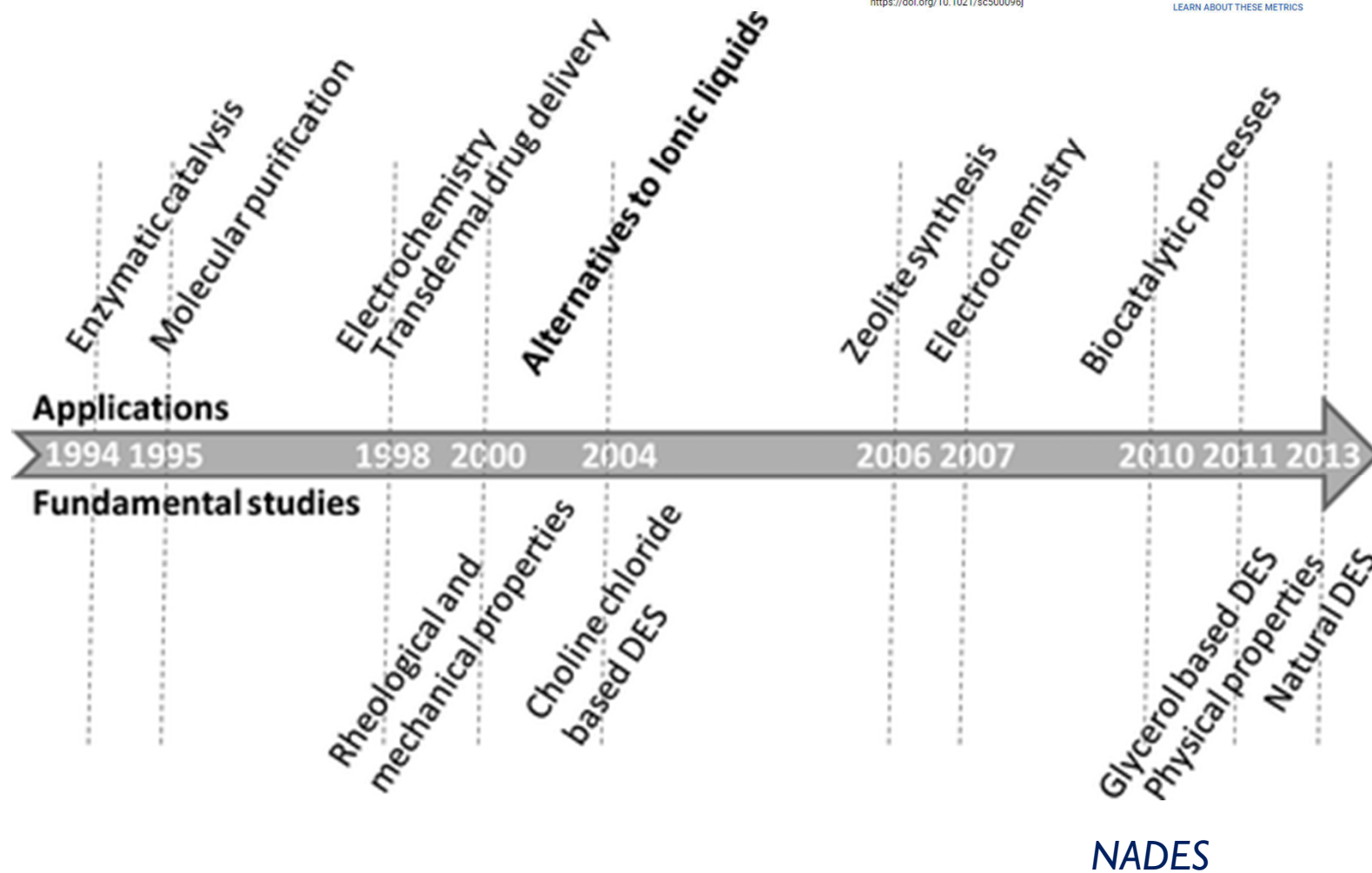
Alexandre Paiva<sup>†</sup>, Rita Craveiro<sup>†</sup>, Ivo Aroso<sup>†§</sup>, Marta Martins<sup>†§</sup>, Rui L. Reis<sup>†§</sup>, and Ana Rita C. Duarte<sup>†§</sup>

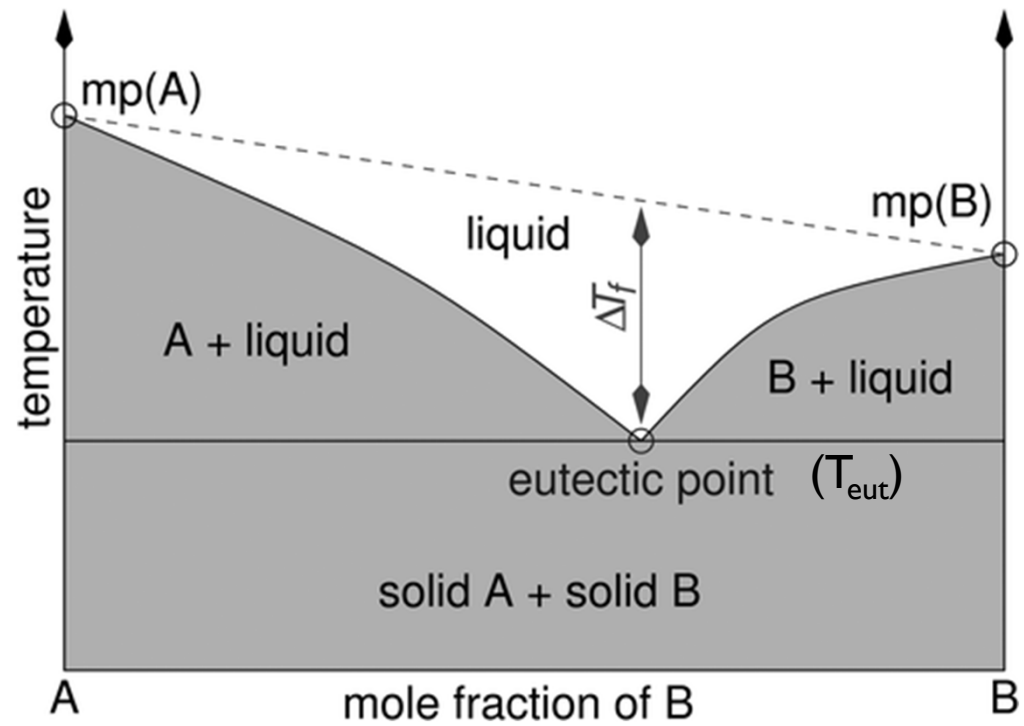
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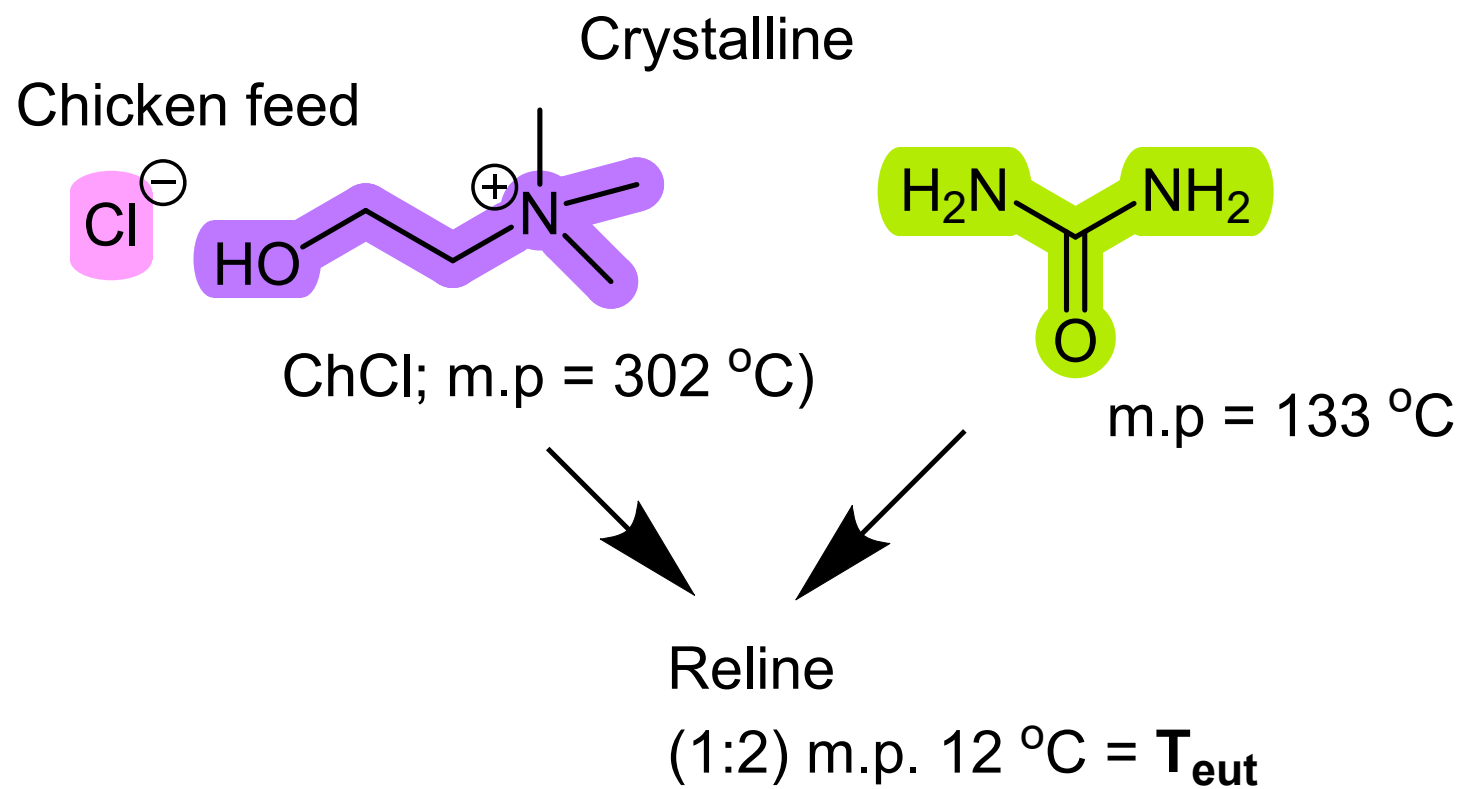
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Cite this: *Chem. Rev.* 2021, 121, 3, 1232–1285

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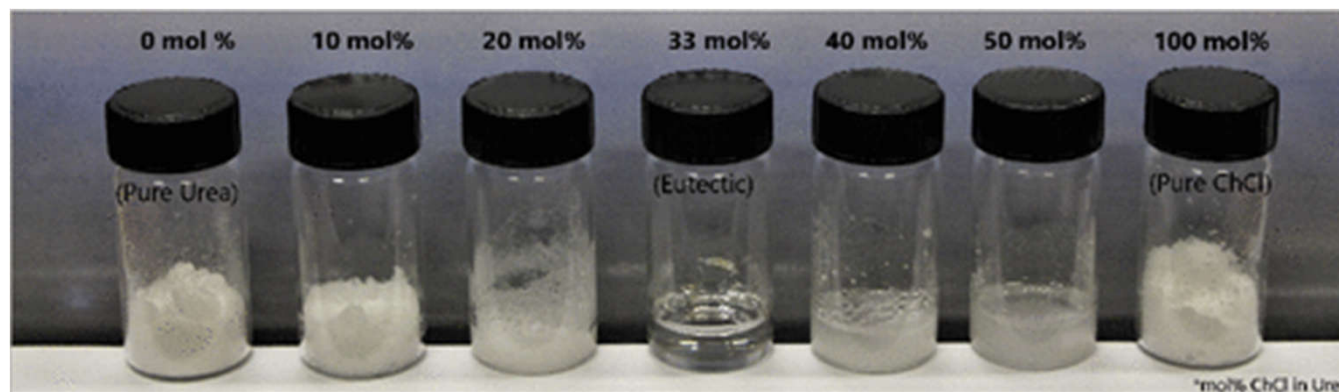
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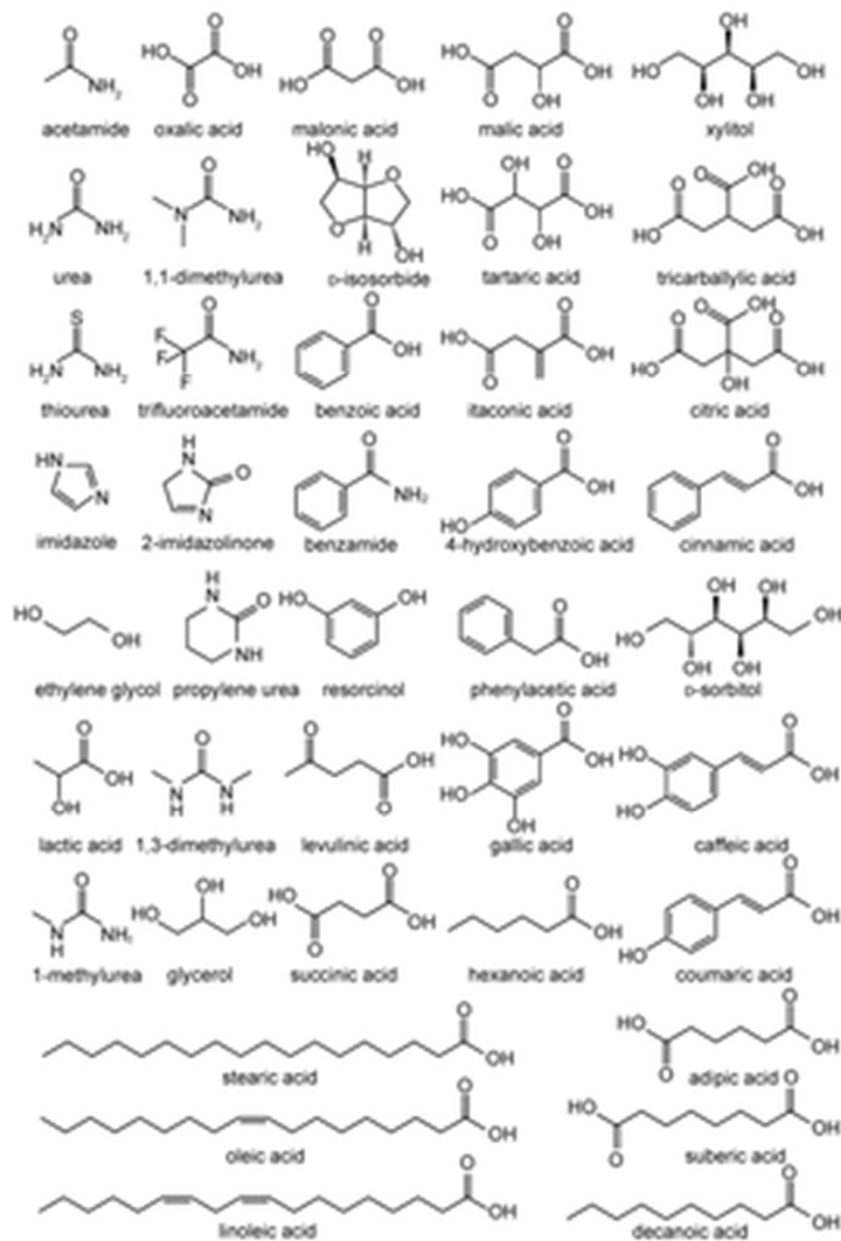
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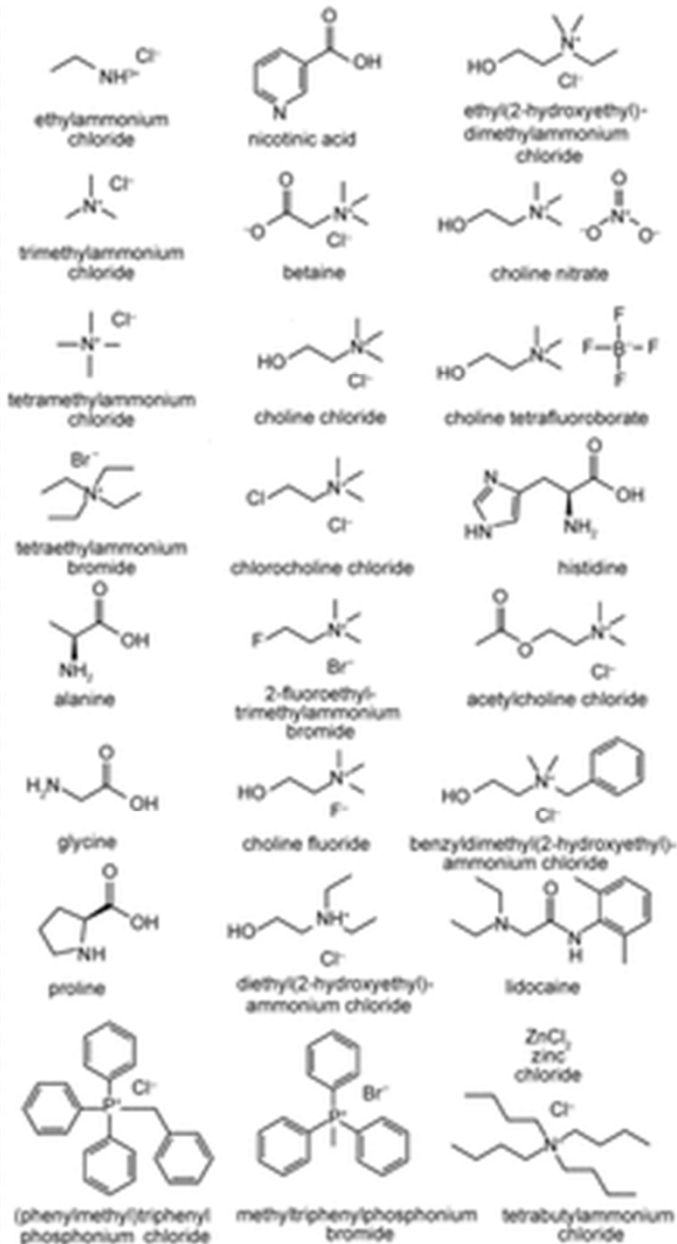


Choline chloride + urea mixtures at 303 K (mol % ChCl, left to right –0, 10, 20, 33 ( $x_{\text{eut}}$ ), 40, 50, 100). The lowest observed melting point for this binary mixture occurs at a 1:2 ratio and has a melting point of 12 °C, *eutectic composition* ( $x_{\text{eut}}$ ).

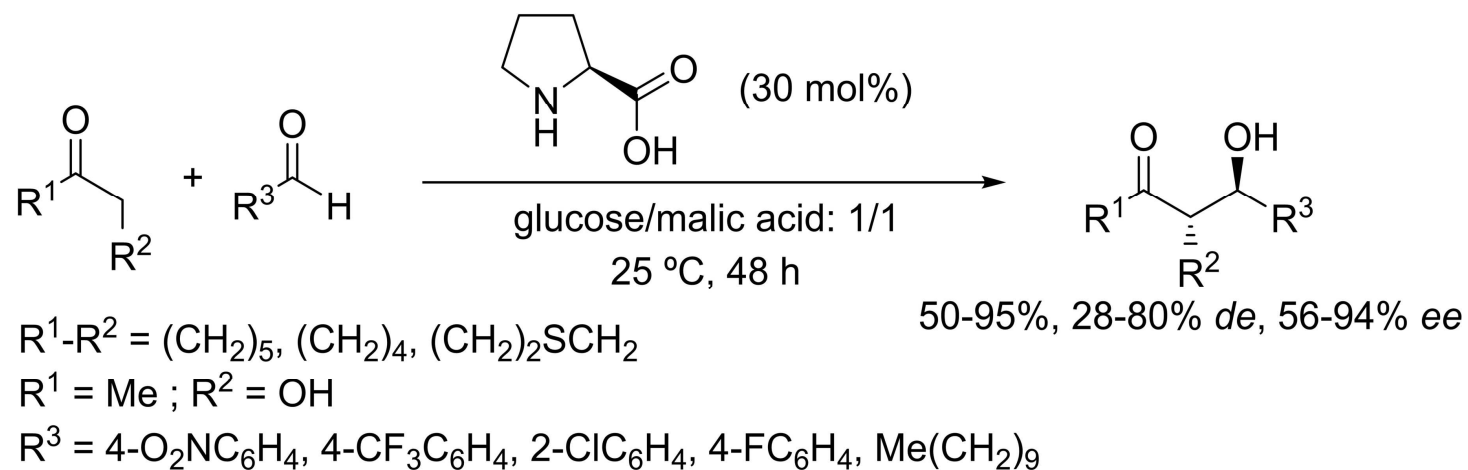
### Hydrogen-Bond Donors



### Hydrogen-Bond Acceptors



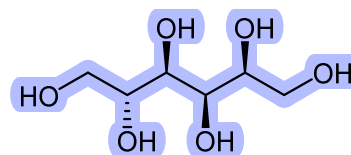
## Benaglia



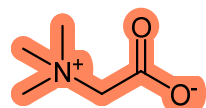
*Beilstein J. Org. Chem.* **2016**, 12, 2620. Benaglia



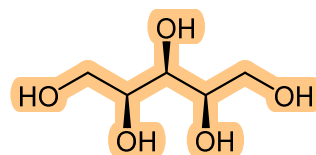
1 : 1 : 3



D-Sorbitol

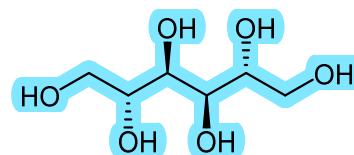


Betaine



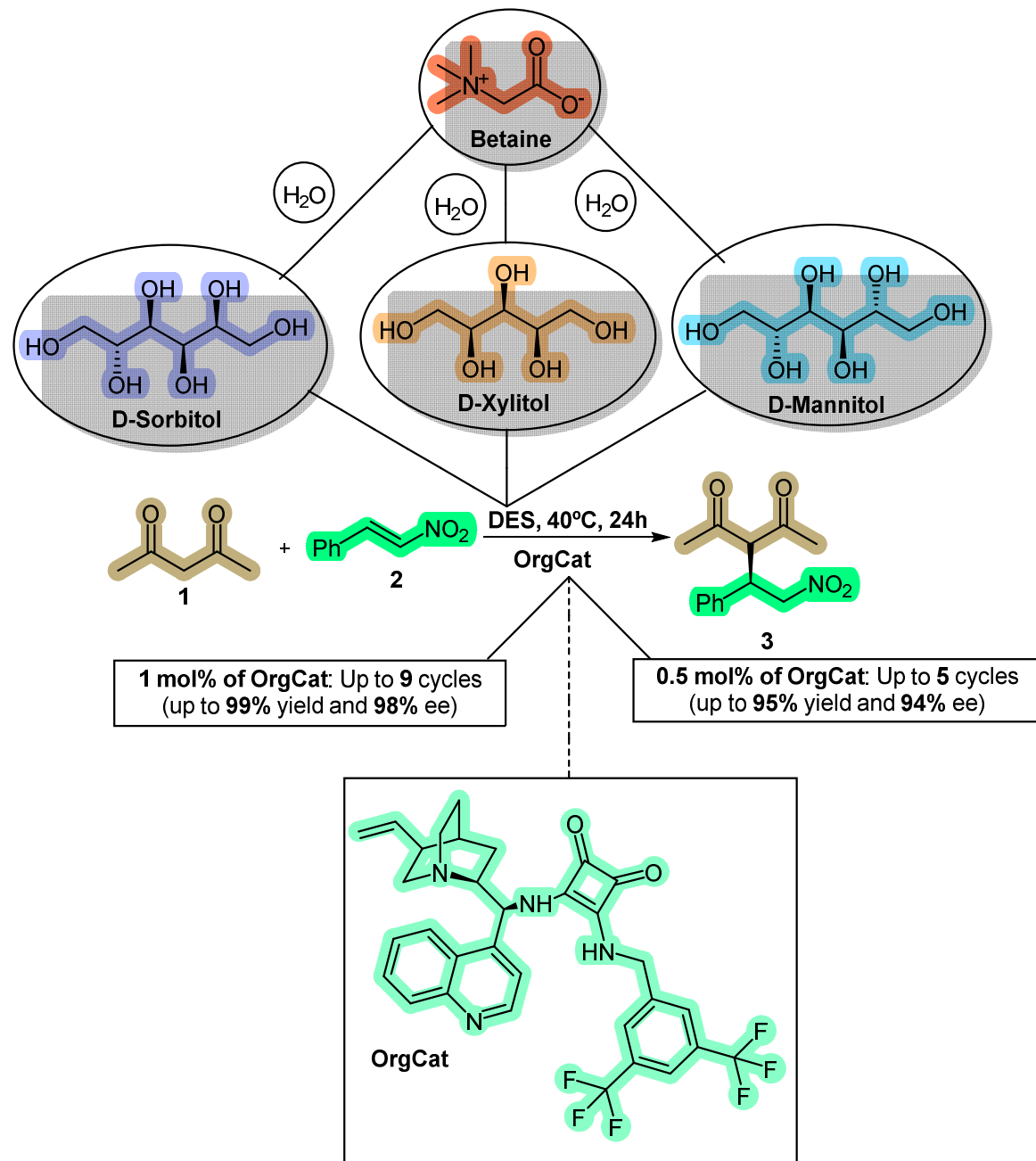
D-Xylitol

H<sub>2</sub>O



D-Mannitol

Chiral DESs



Fonseca, *SynOpen*, in press.

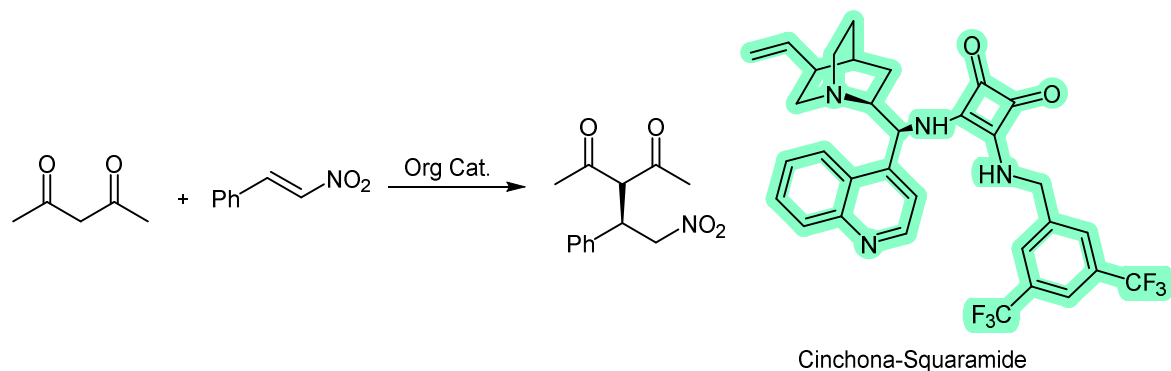


TABLE 1					
Entry	DES	Catalyst (loading)	Cycle	Yield(%)	ee (%)
1	Betaine: D-Sorbitol: Water	(5 mol%)	1	≥99	97
2			2	≥99	62
3			3	≥99	70
4			4	≥99	95
5			5	97	82
6			6	98	90
7			7	≥99	94
8			8	95	90
9			9	88	62

Fonseca, *SynOpen*, in press.

**TABLE 2**

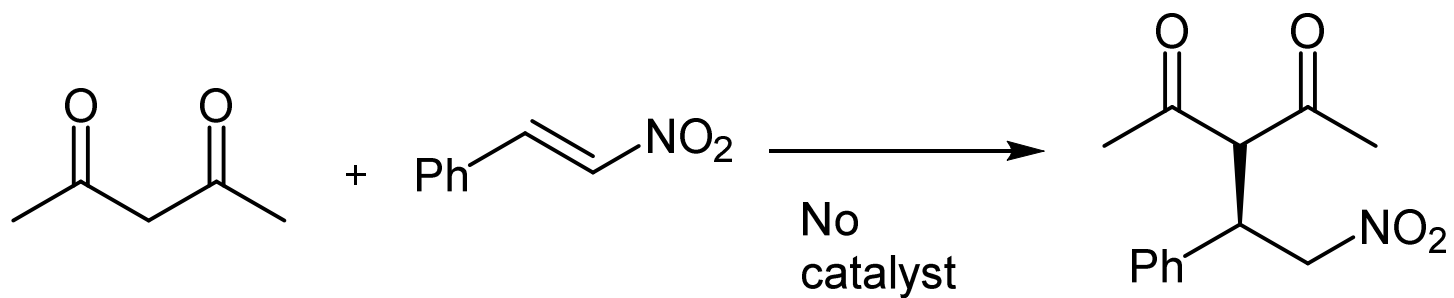
<b>Entry</b>	<b>DES</b>	<b>Catalyst (loading)</b>	<b>Cycle</b>	<b><i>Yield</i> (%)</b>	<b>ee (%)</b>
<b>1</b>	Betaine: D-Sorbitol: Water	(1 mol%)	<b>1</b>	<b>≥99</b>	<b>98</b>
<b>2</b>			<b>2</b>	<b>≥99</b>	<b>69</b>
<b>3</b>			<b>3</b>	<b>≥99</b>	<b>66</b>
<b>4</b>			<b>4</b>	<b>94</b>	<b>79</b>
<b>5</b>			<b>5</b>	<b>95</b>	<b>93</b>
<b>6</b>			<b>6</b>	<b>93</b>	<b>75</b>
<b>7</b>			<b>7</b>	<b>99</b>	<b>60</b>
<b>8</b>			<b>8</b>	<b>94</b>	<b>29</b>
<b>9</b>			<b>9</b>	<b>81</b>	<b>17</b>

**TABLE 3**

Entry	DES	Catalyst (loading)	Cycle	Yield (%)	ee (%)
1	Betaine: D-Xylitol: Water	(0.5 mol%)	1	96	87
2			2	91	93
3			3	89	63
4			4	98	10
5		(1 mol%)	1	98	92
6			2	93	67
7			3	90	93
8			4	86	84

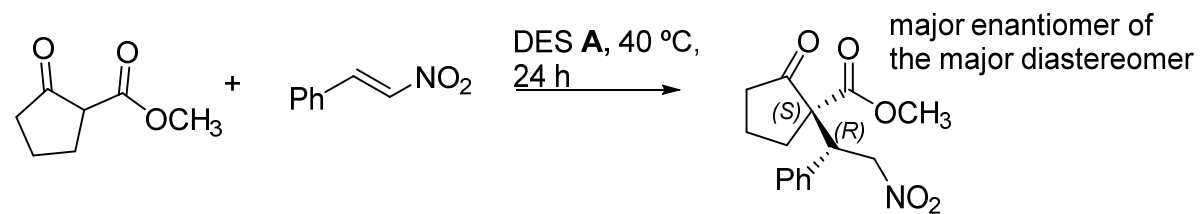
**TABLE 4**

Entry	DES	Catalyst (loading)	Cycle	Yield (%)	ee (%)
1	Betaine: D-Mannitol :Water	(0.5 mol%)	1	57	83
2			2	35	73
3			3	55	99
4			4	67	≥99
5		(1 mol%)	1	19	52
6			2	64	78
7			3	87	87



Back-ground asymmetric catalysis

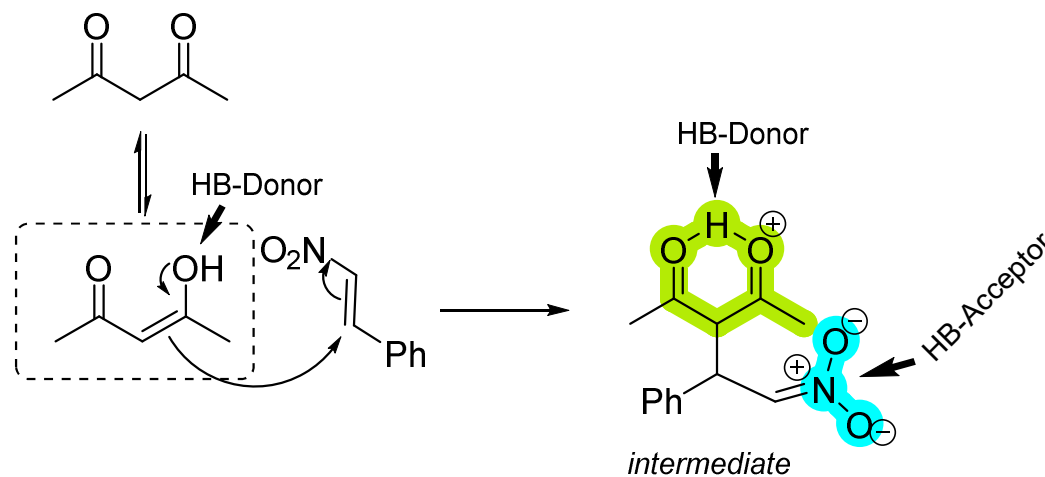
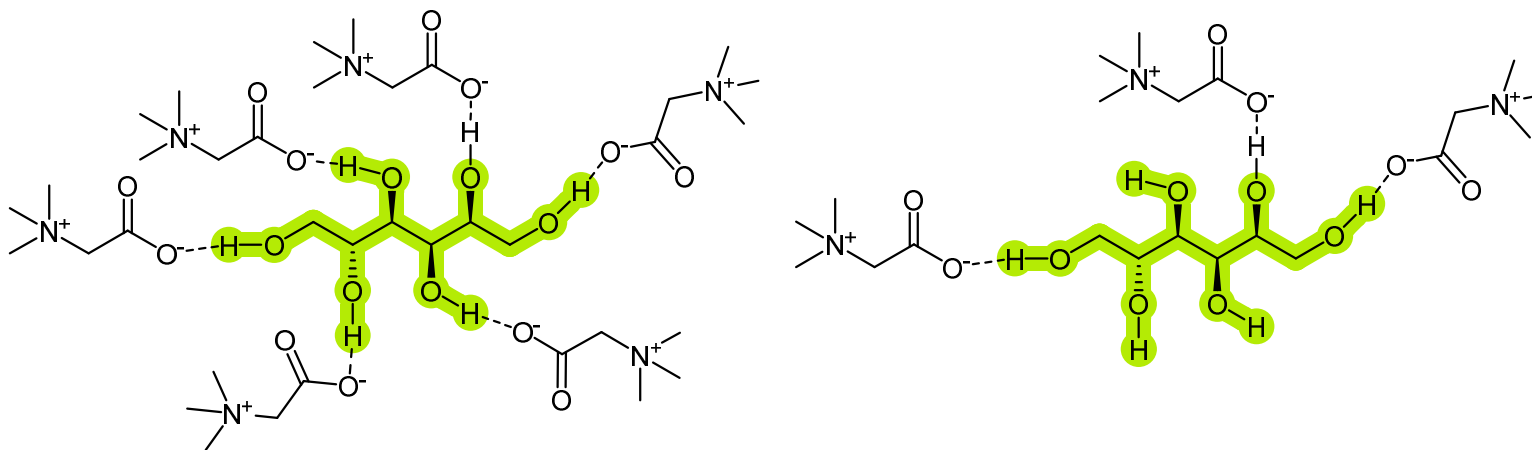
TABLE 5				
Entry	DES	RATIO	Yield(%)	ee (%)
1	Betaine: D-Xylitol: Water	1:1:3	81	75
2			53	3
3			51	3



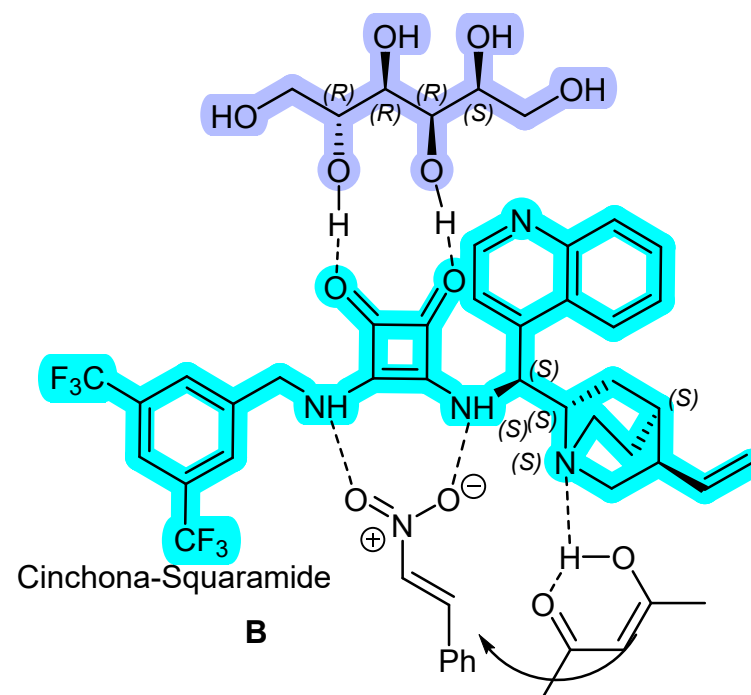
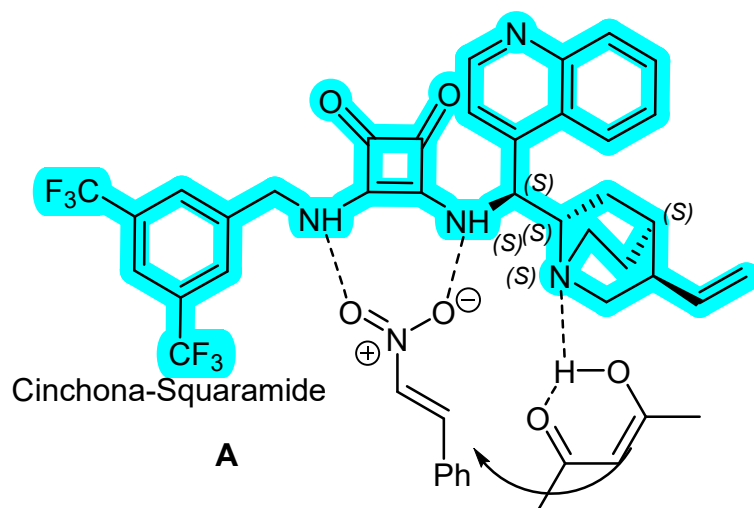
**TABLE 6**

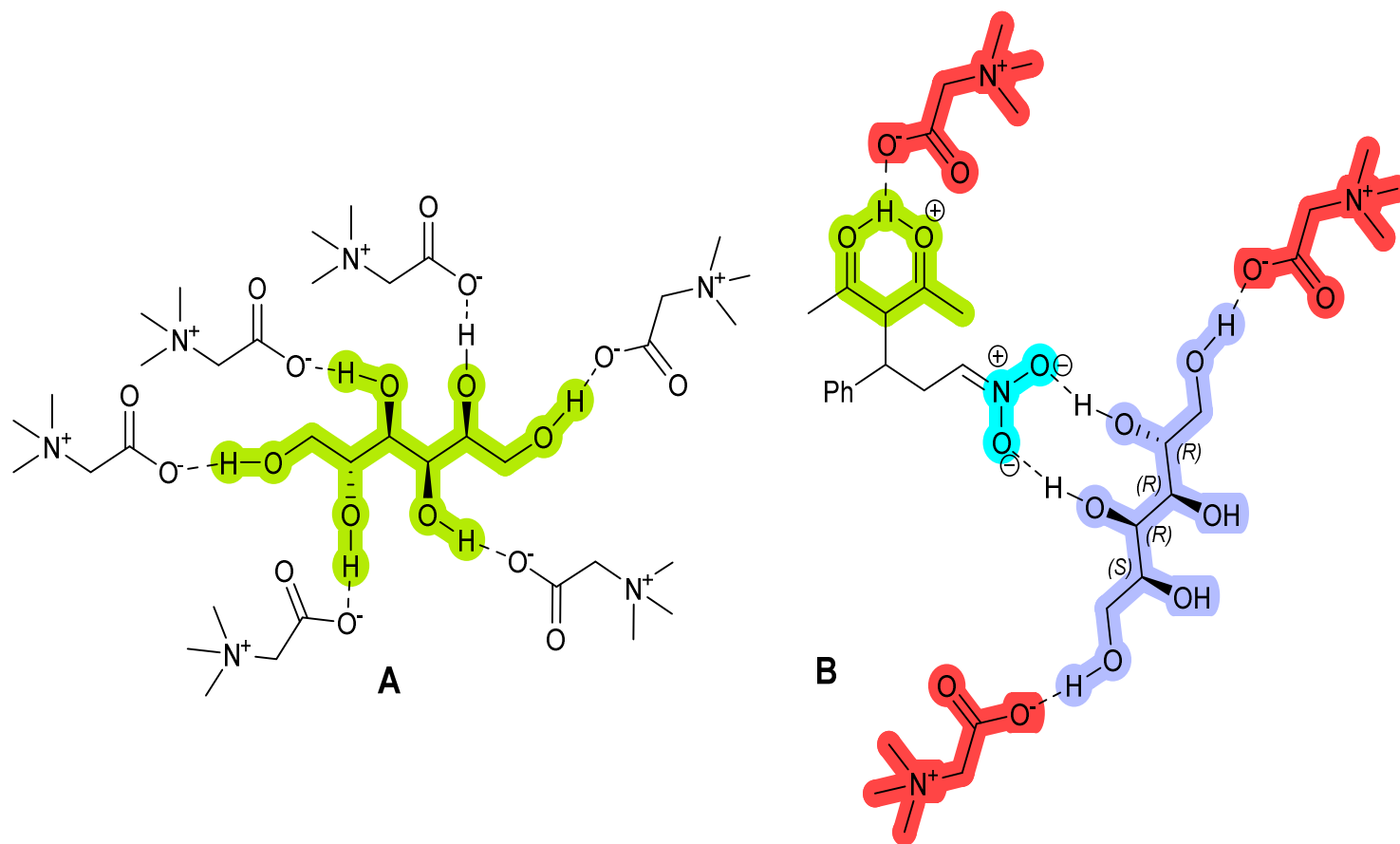
Entry	DES A	Catalyst (loading)	Cycle	Yield (%) <sup>b</sup>	de (%) <sup>c</sup>	ee (%) <sup>c</sup>
1	Betaine: D-Sorbitol: Water	none	-	99	58	rac
2		(1 mol%)	1	99	73	94
3			2	99	92	86
4			3	99	93	84
5			4	99	92	92
6			5	99	94	88
7			6	56	94	96
8			7	98	94	93
9			8	97	97	94
10			9	98	97	95
11			10	97	97	93

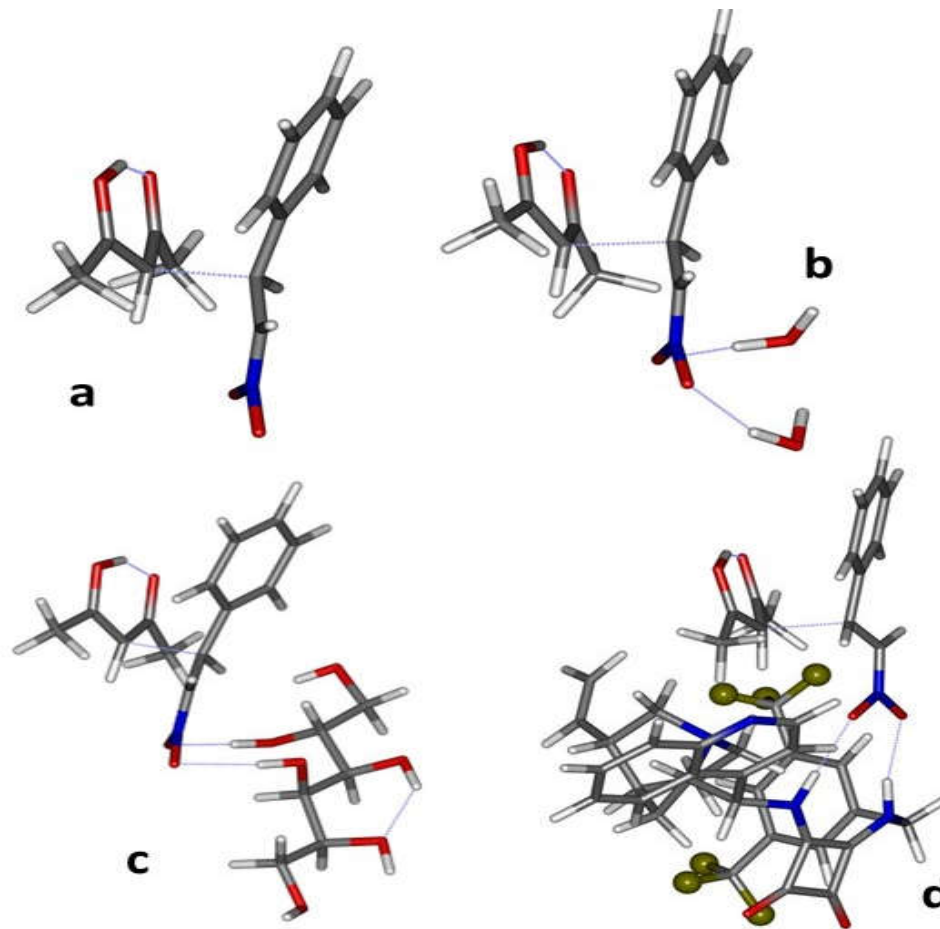




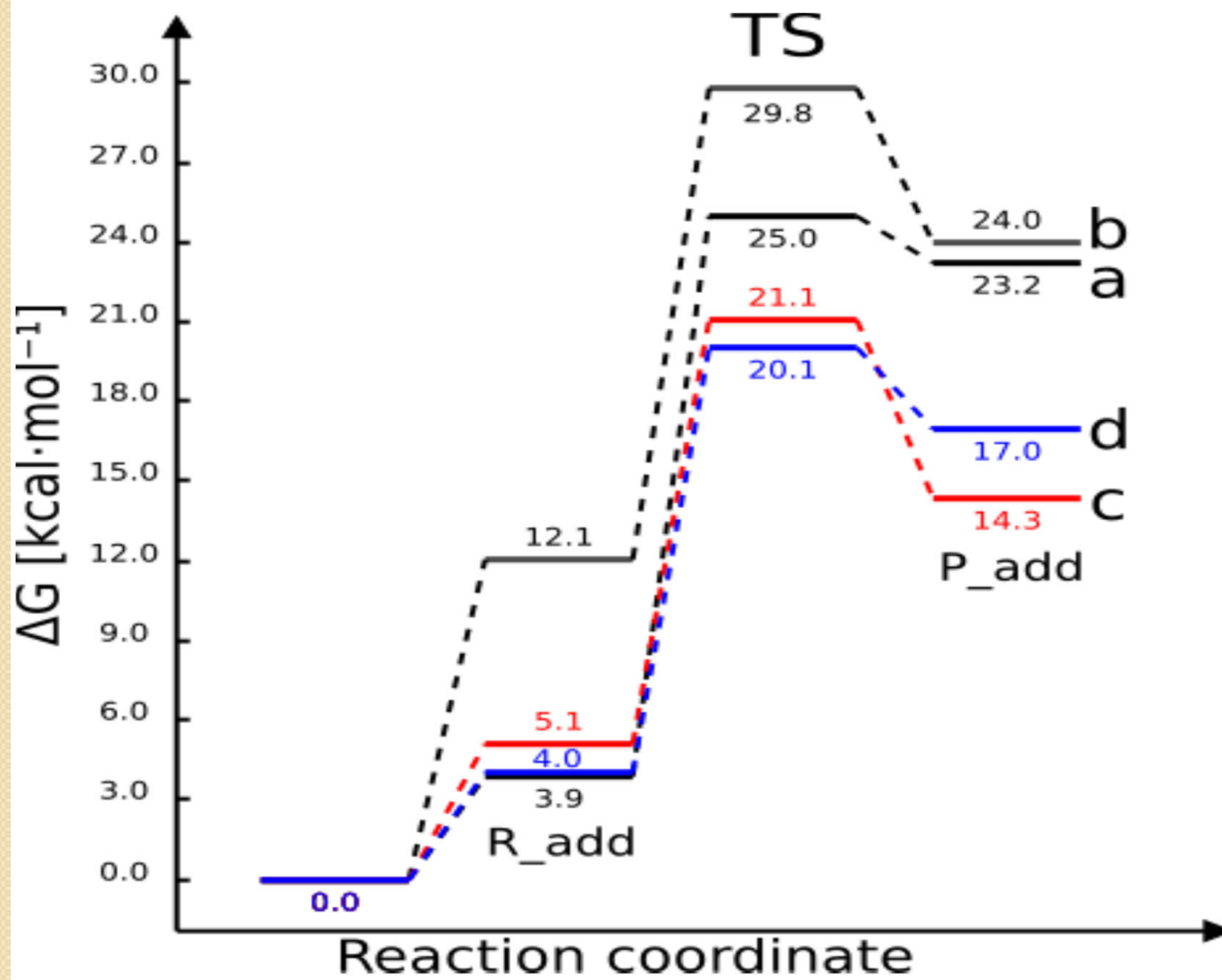
### Transition state stabilization via DES



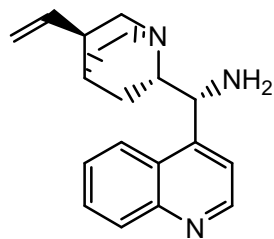




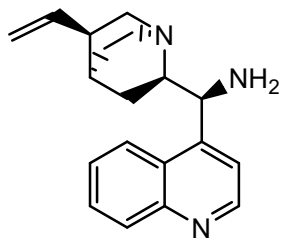
Implicit water (a), implicit water plus explicit water molecules (b), implicit water and sorbitol (c) and implicit water and cinchona-squaramide catalyst (d).



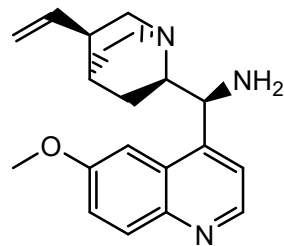
Cinchonine



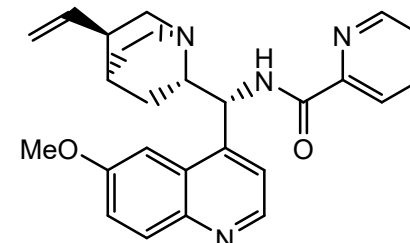
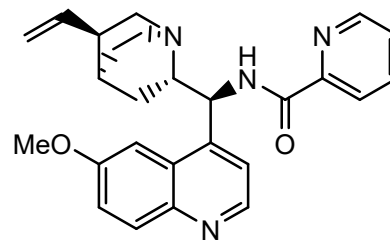
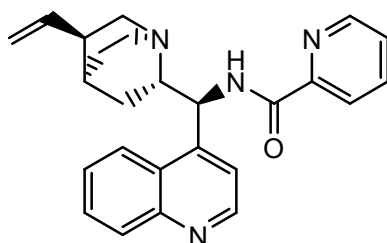
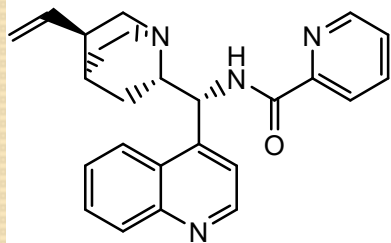
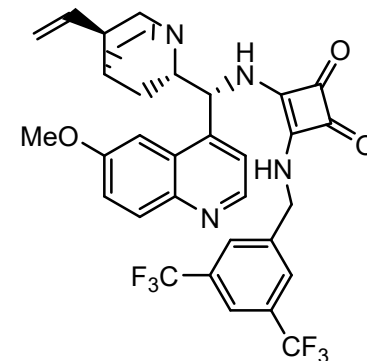
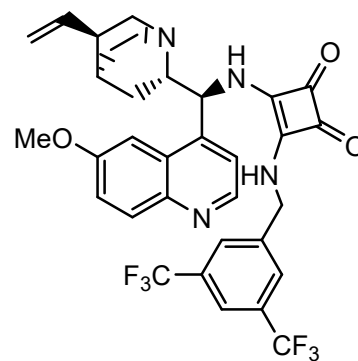
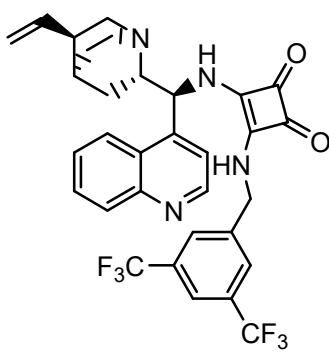
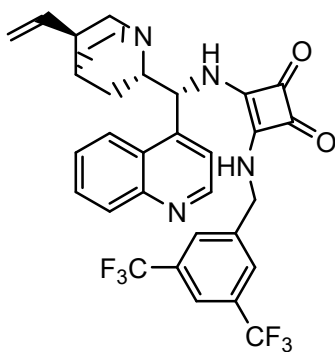
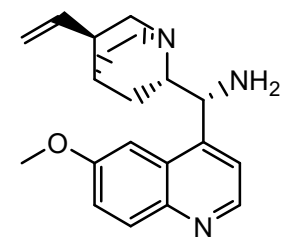
Cinchonidine



Quinine

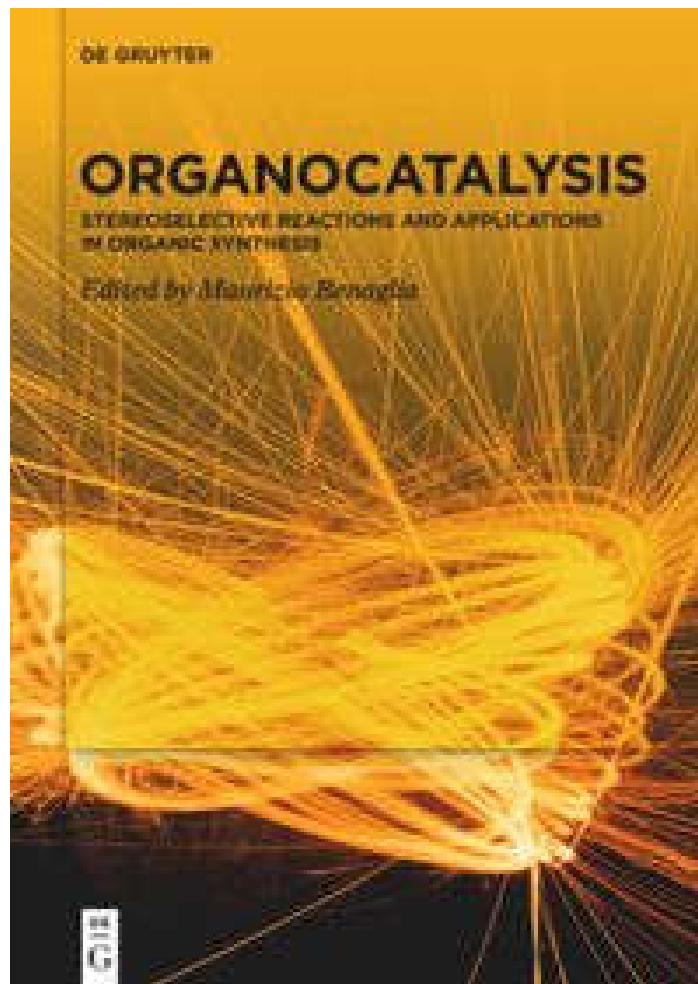


Quinidine



# Summary

- Asymmetric catalytic hydrosilylation reactions.
- With Immobilized cinchona-pyridinamides.
- Cinchona-squaramides.
- Michael Reactions.
- Immobilized Cinchona-squaramides.
- Catalytic Solvents (Deep Eutectic Solvents)





# Acknowledgements

- Dr. Elisabete Carreiro (UEvora)
- Ana Caterina Amorim (ChiraTecnics)
- Dr. Gesine Herrman (ChiraTecnics)
- Edgar Santos (UPorto)
- Prof. Natalia Cordeiro (UPorto)
- Dr. Filipe Teixeira (UBraga)
- Prof. João P. Ramalho (UEvora)
- Daniela Fonseca (UEvora)
- Dr. Pedro Barrulas (UEvora)
- Sílvia Fernandes (UEvora)
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- Dr. Hans-Jürgen Federsel (RISE)



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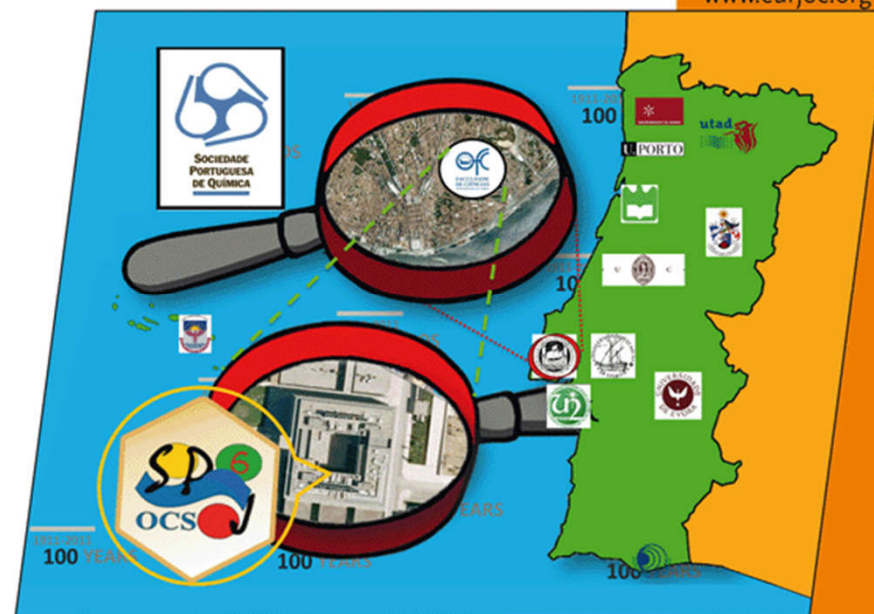


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